

JEE-MAIN EXAMINATION – JANUARY 2025

(HELD ON TUESDAY 28<sup>th</sup> JANUARY 2025)

TIME : 9:00AM TO 12:00 NOON

**MATHEMATICS**

**TEST PAPER WITH SOLUTION**

SECTION-A

1. The number of different 6 digit numbers greater than 50000 that can be formed using the digits 0, 1, 2, 3, 4, 5, 6, 7, such that the sum of their first and last digits should not be more than 8, is

- (1) 5608
- (2) 5720
- (3) 5719
- (4) 5607

Ans. (4)

Sol. Case I 5 \_ \_ \_ \_ .

Case II 5 \_ \_ \_ \_ 1

- 0 2
- 0 3
- 6 0
- 6 1
- 6 2
- 7 0

Case IX 7 \_ \_ \_ \_ 1

$9 \times (8 \times 8 \times 8) = 5768$  but 50000 is not included, so total numbers  $5768 - 1 = 5767$

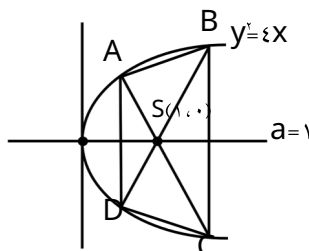
2. Let ABCD be a trapezium whose vertices lie on the parabola  $y = \epsilon x$ . Let the sides AD and BC of the trapezium be parallel to y-axis. If the diagonal

AC is of length  $\frac{70}{\epsilon}$  and it passes through the point

$(1, 0)$ , then the area of ABCD is :

- (1)  $\frac{70}{\epsilon}$
- (2)  $\frac{20}{\epsilon}$
- (3)  $\frac{120}{\lambda}$
- (4)  $\frac{70}{\lambda}$

Ans. (1)



Sol.

$A(at_1^2, \epsilon at_1)$  &  $C(\frac{a}{t_1^2}, \frac{2a}{t_1})$

Length AC =  $a\sqrt{t_1^2 + \frac{1}{t_1^2}} = \frac{25}{4} = t_1 \sqrt{t_1^2 + \frac{1}{t_1^2}}$

$t_1 = 2$  or  $\frac{1}{2}$ .  $A(4, 2\epsilon)$ ,  $D(\frac{1}{4}, 2\epsilon)$ ,  $B(\epsilon, \epsilon)$ ,  $C(\epsilon, -\epsilon)$

So, area of trapezium =  $\frac{1}{2}(8+2)4\epsilon = \frac{1}{4} \times \frac{75}{4}$

3. Two number  $k$  and  $k$  are randomly chosen from the set of natural numbers. Then, the probability that the value of  $\frac{1}{i} + \frac{1}{j}$  ( $i, j \in \mathbb{N}$ ) is non-zero, equals

- (1)  $\frac{1}{2}$
- (2)  $\frac{1}{\epsilon}$
- (3)  $\frac{3}{\epsilon}$
- (4)  $\frac{2}{3}$

Ans. (3)

Sol.  $i, j \in \mathbb{N}$ ,  $i, j \in \mathbb{Z}$  option for  $i, -1, -i, \dots$

Total cases =  $\epsilon \times \epsilon = 16$

Unfavourable cases =  $i, j \in \mathbb{N}$ .

- $1, 1$
- $1, 1$
- $i, i$
- $i, i$
- $i, i$

Cases = Probability =  $\frac{16-4}{16} = \frac{3}{4}$

4. If  $f(x) = \frac{2x}{\sqrt{x}}$ ,  $x \in \mathbb{R}$ , then  $\int_{\frac{1}{\sqrt{2}}}^{\sqrt{2}} f(x) dx$  is equal

to :

- (1)  $\epsilon 1$
- (2)  $\frac{\lambda 1}{\epsilon}$
- (3)  $\lambda 2$
- (4)  $\lambda 1 \sqrt{\epsilon}$

Ans. (2)

Sol.  $f(x) = \frac{2x}{2x\sqrt{2}}$

$f(x) + f(1-x) = \frac{2x}{2x\sqrt{2}} + \frac{2(1-x)}{2(1-x)\sqrt{2}}$

$= \frac{2x}{2x\sqrt{2}} + \frac{2}{2\sqrt{2}2x} = \frac{2}{2\sqrt{2}}$

Now,  $f\left(\frac{1}{82}\right) + f\left(\frac{81}{82}\right) + f\left(\frac{2}{82}\right) + f\left(\frac{80}{82}\right) + \dots + f\left(\frac{81}{82}\right) + f\left(\frac{1}{82}\right)$

$= f\left(\frac{1}{82}\right) + f\left(\frac{81}{82}\right) + f\left(\frac{2}{82}\right) + f\left(\frac{80}{82}\right) + \dots + f\left(\frac{81}{82}\right) + f\left(\frac{1}{82}\right)$

$\dots + f\left(\frac{1}{82}\right) + f\left(\frac{81}{82}\right) + f\left(\frac{2}{82}\right) + f\left(\frac{80}{82}\right) + \dots + f\left(\frac{81}{82}\right) + f\left(\frac{1}{82}\right)$  ... 40 cases  $f\left(\frac{41}{82}\right)$

$40 \times \frac{2}{2\sqrt{2}} = 40 \times \frac{1}{\sqrt{2}}$

$40 \times \frac{1}{\sqrt{2}} = 20\sqrt{2}$

Let  $f: \mathbb{R} \rightarrow \mathbb{R}$  be a function defined by

$f(x) = (r+ra)x + \frac{a}{a-1}x + b, a \neq 1$ . If

$f(x+y) = f(x) + f(y) + 1 - \frac{r}{v}xy$ , then the value of

$\sum_{i=1}^n |f(i)|$  is:

- (1)  $v \cdot 0$
- (2)  $v \cdot 0$
- (3)  $0 \cdot 0$
- (4)  $v \cdot 0$

Ans. (4)

Sol.  $f(x) = (ra+r)x + \frac{a-2}{a-1}x + b$

$f(x) + f(y) = \frac{2}{v}xy \dots (1)$

In (1) Put  $x=y=0 \Rightarrow f(0) = rf(0) + 1 \Rightarrow f(0) = -1$

So,  $f(0) = 0 + 0 + b = -1 \Rightarrow b = -1$   
In (1) Put  $y = -x \Rightarrow f(0) = f(x) + f(-x) + 1 +$

$\frac{2}{v}x^2$

$-1 = r(ra+r)x + \frac{2}{v}x^2$

$-1 = 2(3a-2) - \frac{2}{7}x^2$

$ra+r + \frac{2}{v} = 0$

$a = \frac{5}{7}$

So  $f(x) = \frac{1}{7}x + \frac{3}{4}x + 1$

$|f(x)| = \frac{1}{28} |4x^2 + 21x + 28|$

Now,  $28^5 |f(1)| + 28^4 |f(2)| + \dots + |f(5)|$

$28 \cdot \frac{1}{28} \cdot 675 = 675$

Let  $A(x, y, z)$  be a point in  $xy$ -plane, which is equidistant from three points  $(0, r, r), (r, 0, r)$  and  $(0, 0, 1)$ .

Let  $B = (1, 1, -1)$  and  $C = (r, 0, -r)$ . Then among the statements

(S1) :  $\triangle ABC$  is an isosceles right angled

triangle

and

(S2) : the area of  $\triangle ABC$  is  $\frac{9\sqrt{r}}{r}$ .

- (1) both are true
- (2) only (S1) is true
- (3) only (S2) is true
- (4) both are false

Ans. (2)

Sol.  $A(x, y, z)$  Let  $P(0, r, r), Q(r, 0, r), R(0, 0, 1)$

$AP = AQ = AR$

$x^2 + (y-r)^2 + (z-r)^2 = (x-r)^2 + y^2 + (z-r)^2 = x^2 + y^2 + (z-1)^2$

In  $xy$  plane  $z = 1$

So,  $x^2 - 2x + 1 + y^2 + 1 = x^2 + y^2 + 1$

$\Rightarrow y = 1$

$x = r$

$1 + y - 1y + 1 + 1 = x + y + 1$

So,  $A(r, 1, 1)$  also  $B(1, 1, -1)$  &  $C(r, 0, -r)$

Now  $AB = \sqrt{4+4} = 2\sqrt{2}$

$$AC = \sqrt{16+9} = 5$$

$$BC = \sqrt{16+9} = 5$$

$$AB = AC$$

isosceles  $\triangle$  &  $AB + AC = BC$

right angle  $\triangle$

$$\text{Area of } \triangle ABC = \frac{1}{2} \times \text{base} \cdot \text{height}$$

$$\frac{1}{2} \times 3 \times 3 = \frac{9}{2}$$

So only S is true

v. The relation  $R = \{(x, y) : x, y \in \mathbb{Z} \text{ and } x+y \text{ is even}\}$  is :

- (i) reflexive and transitive but not symmetric
- (ii) reflexive and symmetric but not transitive
- (iii) an equivalence relation
- (iv) symmetric and transitive but not reflexive

Ans. (iii)

Sol.  $R = \{(x, y) : x, y \in \mathbb{Z} \text{ and } x+y \text{ is even}\}$

reflexive  $x+x = 2x$  even

symmetric if  $x+y$  is even, then  $(y+x)$  is also even

transitive if  $x+y$  is even &  $y+z$  is even then  $x+z$  is also even

So, relation is an equivalence relation.

Λ. Let the equation of the circle, which touches x-axis at the point  $(a, 0)$ ,  $a < 0$  and cuts off an intercept of length  $b$  on y-axis be  $x^2 + y^2 - 2ax + 2by + c = 0$ . If

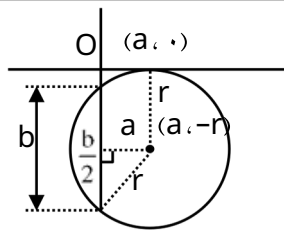
the circle lies below x-axis, then the ordered pair

$(\frac{c}{a}, \frac{b}{a})$  is equal to :

- (i)  $(\frac{c}{a}, \frac{b}{a}) = (-1, -1)$
- (ii)  $(\frac{c}{a}, \frac{b}{a}) = (-1, 1)$
- (iii)  $(\frac{c}{a}, \frac{b}{a}) = (1, -1)$
- (iv)  $(\frac{c}{a}, \frac{b}{a}) = (1, 1)$

Ans. (iv)

Sol.



By pythagorus  $a^2 + \frac{b^2}{4} = r^2$

$$r = \sqrt{\frac{4a^2 + b^2}{4}}$$

Equation of circle is  $(x-a)^2 + (y+r)^2 = r^2$

$$x^2 + y^2 - 2ax - 2ry + a^2 + r^2 = 0$$

comparision  $x^2 + y^2 - 2ax + 2by + c = 0$

$$-2a = -2a, 2b = -2r, c = a^2 + r^2$$

$$2a = 2a, 2b = -2r, c = a^2 + r^2$$

$$2b = -2r$$

$$b = -r$$

$$\text{So, } (2a, b) = (2a, -r)$$

v. Let  $\{a_n\}$  be a sequence such that  $a_1 = 1, a_n = \frac{1}{n}$  and

$$ra_{n+1} = a_n - ra_n, n = 1, 2, 3, \dots \text{ Then } \sum_{k=1}^{\infty} a_k$$

is equal to :

$$(i) \frac{1}{2}$$

$$(ii) \frac{1}{3}$$

$$(iii) \frac{1}{4}$$

$$(iv) \frac{1}{5}$$

Ans. (ii)

$$\text{Sol. } a_1 = 1, a_n = \frac{1}{n}$$

$$ra_{n+1} = a_n - ra_n$$

$$rx^2 - 2x + r = 0, x = 1, r/r$$

$$a_n = A(1) + B^n \begin{pmatrix} 3 \\ 2 \end{pmatrix}$$

$$n \rightarrow 0, 0 = A + B \Rightarrow A = -B$$

$$n \rightarrow 1, 1 = A + 2B \Rightarrow B = 1, A = -1$$

$$a_n = -1 + \frac{3^n - 1}{2}$$

$$\sum_{k=1}^{100} a_k = \sum_{k=1}^{100} \left( \frac{3^k - 1}{2} \right) = \frac{1}{2} \left( \sum_{k=1}^{100} 3^k - 100 \right)$$

$$100 \times \frac{3 \times 3 \times 100}{2 \times 3}$$

$$= 100 \times \frac{3 \times 3 \times 100}{2 \times 3}$$

$$= r \cdot (a_{\dots}) - 100$$

10.  $\cos \sin \frac{1}{5} \sin \frac{5}{13} \sin \frac{13}{70}$  is equal to :

(1) 1

(2) 2

(3)  $\frac{33}{70}$

(4)  $\frac{32}{70}$

Ans. (2)

Sol.  $\cos \sin \frac{1}{5} \sin \frac{5}{13} \sin \frac{13}{70}$

$$\cos \tan \frac{3}{4} \tan \frac{1}{12} \tan \frac{13}{56}$$

$$\cos \tan \frac{3}{4} \frac{5}{12} \tan \frac{13}{56}$$

$$\cos \tan \frac{156}{33} \cot \frac{156}{33}$$

$$\cos \frac{0}{2} = 1$$

11. Let  $T_r$  be the  $r^{\text{th}}$  term of an A.P. If for some  $m$ ,

$$T_m = \frac{1}{r_0}, T_{r_0} = \frac{1}{r} \text{ and } r \cdot \sum_{r=1}^{r_0} T_r = 13, \text{ then}$$

$\sum_{r=1}^{r_0} T_r$  is equal to:

(1) 112

(2) 126

(3) 98

(4) 142

Ans. (2)

Sol.  $T_m = \frac{1}{r_0}, T_{r_0} = \frac{1}{r}, r \cdot \sum_{r=1}^{r_0} T_r = 13$

$$T_m = a + (m-1)d = \frac{1}{25} \dots \dots (1)$$

$$T_{r_0} = a + r_0 d = \frac{1}{20}$$

$$20 \cdot \frac{25}{2} a \cdot \frac{1}{20} \cdot 13 = a \cdot \frac{1}{500}$$

also,  $r \cdot S_r = 20 \cdot \frac{25}{2} \cdot 2a \cdot 24d \cdot 13 = d \cdot \frac{1}{500}$

from (1)  $\frac{1}{500} = \frac{m \cdot 1}{500} = \frac{1}{25} \cdot m \cdot 20$

Now,

$$\sum_{r=1}^{r_0} T_r = \dots \sum_{r=1}^{r_0} T_r = 126$$

12. If the image of the point  $(x, y, z)$  in the line

$$\frac{x-1}{2} = \frac{y-2}{1} = \frac{z-1}{3} \text{ is } (x', y', z') \text{ then } x' \text{ is}$$

equal

to (1) 9

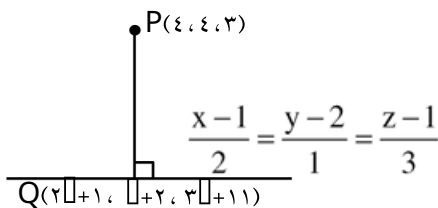
(2) 12

(3) 8

(4) 7

Ans.

Sol.



$$PQ = \dots \hat{i} \hat{j} \hat{k}$$

$$\dots + \dots + \dots = \dots$$

$$\dots - \dots = \dots$$

So,  $Q(x', y', z')$

Let image in  $R(x, y, z)$

$$\frac{x-1}{2} = \frac{y-2}{1} = \frac{z-1}{3}$$

$$\dots = (x, y, z)$$

$$\dots + \dots = 9$$

13. If  $\int \frac{1}{\sqrt{1-x^2}} dx = \dots$ , then

$\dots$  equals :

(1) 144

(2) 196

(3) 100

(4) 64

Ans. (3)

Sol.  $\int \frac{1}{\sqrt{1-x^2}} dx$  (Apply King Property)

$$\int_0^{\pi} x \cos 2x \, dx + \int_0^{\pi} x \cos x \, dx$$

$$\int_0^{\pi} x \cos 2x \, dx + \int_0^{\pi} x \cos x \, dx$$

On solving  $\int_0^{\pi} x \cos 2x \, dx$

$$\int_0^{\pi} x \cos 2x \, dx = -\frac{1}{2} \int_0^{\pi} x \sin 2x \, dx$$

$$\int_0^{\pi} x \sin 2x \, dx = \frac{1}{2} \int_0^{\pi} x \cos 2x \, dx$$

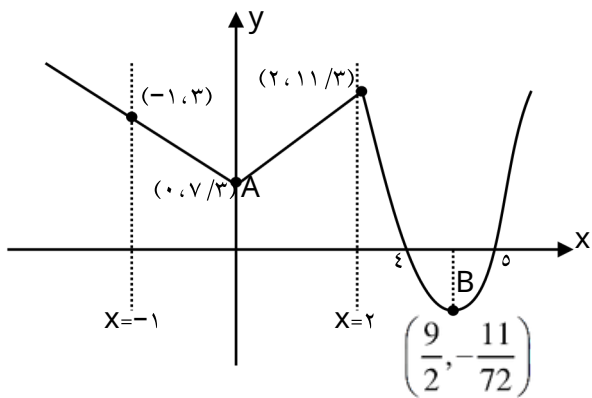
14. The sum of all local minimum values of the

$$f(x) = \frac{1}{\sqrt{x}} + \sqrt{x} + \frac{1}{x} + x$$

- (1)  $\frac{171}{\sqrt{2}}$
- (2)  $\frac{131}{\sqrt{2}}$
- (3)  $\frac{107}{\sqrt{2}}$
- (4)  $\frac{167}{\sqrt{2}}$

Ans. (3)

$$f(x) = \frac{1}{\sqrt{x}} + \sqrt{x} + \frac{1}{x} + x$$



Local minimum values at A & B

- (1)  $\frac{11}{\sqrt{2}}$
- (2)  $\frac{167}{\sqrt{2}}$
- (3)  $\frac{107}{\sqrt{2}}$
- (4)  $\frac{171}{\sqrt{2}}$

15. The sum of the squares of all the roots of the equation  $x^2 + |2x - 3| - \epsilon = 0$  is:

- (1)  $2(3 + \sqrt{2})$
- (2)  $2(3 - \sqrt{2})$
- (3)  $2(2 + \sqrt{2})$
- (4)  $2(2 - \sqrt{2})$

Ans. (3)

$$\text{Sol. } x^2 + |2x - 3| - \epsilon = 0$$

Case I :  $x \geq \frac{3}{2}$

$$x^2 + 2x - 3 - \epsilon = 0$$

$$x^2 + 2x - 7 = 0$$

$$x = 2\sqrt{2} - 1$$

Case II :  $x < \frac{3}{2}$

$$x^2 + 3 - 2x - \epsilon = 0$$

$$x^2 - 2x - 1 = 0$$

$$x = 1 - \sqrt{2}$$

Sum of squares =  $(2\sqrt{2} - 1)^2 + (1 - \sqrt{2})^2 = 2(2 + 1) = 6$

$$= 2 + 1 - 2\sqrt{2} + 1 + 2 = 6$$

Let for some function  $y = f(x)$ .

16.  $\int_0^1 t f(t) dt = x + (x)^2$ ,  $x < 1$

$x < 1$  and  $f(1) = 2$ . Then  $f(1)$  is equal to:

- (1) 1
- (2) 2
- (3) 3
- (4) 4

Ans. (1)

Sol.  $\int_0^1 t f(t) dt = x + (x)^2$ ,  $x < 1$

Diff. both side w.r. to x

$$x f(x) = x f'(x) + 2x f(x)$$

$$-x f(x) = x f'(x)$$

$$\int \frac{f'(x)}{f(x)} dx = \int \frac{-1}{x} dx$$

$$\log f(x) = -\log x + \log c$$

$$f(x) = \frac{c}{x}$$

$$f(1) = 2 \implies 2 = \frac{c}{1} \implies c = 2$$

$$f(x) = \frac{2}{x}$$

$$f(1) = 2 \implies \text{Ans. (1)}$$

17. Let  $C_{r-1} = 28$ ,  $C_r = 56$  and  $C_{n-r+1} = 35$ . Let  $A(\xi \cos t, \xi \sin t)$ ,  $B(\eta \sin t, -\eta \cos t)$  and  $C(r^2 - n, r - n)$  be the vertices of a triangle ABC, where  $t$  is a parameter. If  $(rx - y) + (ry) = 1$  is the locus of the centroid of triangle ABC, then  $\frac{1}{r}$  equals :

- (1)  $\frac{1}{2}$
- (2)  $\frac{1}{3}$
- (3)  $\frac{1}{4}$
- (4)  $\frac{1}{5}$

Ans. (1)

Sol.  ${}^nC_{r-1} = 28$ ,  ${}^nC_r = 56$

$$\frac{{}^nC_{r-1}}{{}^nC_r} = \frac{28}{56}$$

$$\frac{n!}{(r-1)!(n-r+1)!} \cdot \frac{r!}{n!} = \frac{28}{56}$$

$$\frac{r}{(n-r+1)} = \frac{1}{2}$$

$2r = n + 1$  —(i)

$$\frac{{}^nC_r}{{}^nC_{n-r}} = \frac{56}{35}$$

$$\frac{r!}{(n-r)!} \cdot \frac{(n-r)!}{r!} = \frac{56}{35}$$

$\frac{1}{r} = \frac{8}{5}$  —(ii)

By (i) & (ii)

$(r=3), (n=5)$   
 $A(\xi \cos t, \xi \sin t)$

$A(\xi \cos t, \xi \sin t)$   $B(\eta \sin t, -\eta \cos t)$   $C(r^2 - n, r - n)$   
 $B(\eta \sin t, -\eta \cos t)$   $C(1, -1)$

$(rx - 1) + (ry) = (\xi \cos t + \eta \sin t) + (\xi \sin t - \eta \cos t)$

$(rx - 1) + (ry) = 1$  — Option (1)

18. Let O be the origin, the point A be  $z_1 = \sqrt{3} + i\sqrt{3}$ , the point B(z) be such that

$\sqrt{3}|z_1| = |z|$  and  $\arg(z) = \arg(z_1) + \frac{\pi}{6}$ . Then

(1) area of triangle ABO is  $\frac{11}{\sqrt{3}}$

(2) ABO is a scalene triangle

(3) area of triangle ABO is  $\frac{11}{4}$

(4) ABO is an obtuse angled isosceles triangle

Ans. (4)

Sol.  $z_1 = \sqrt{3} + i\sqrt{3}$  &  $\frac{|z_1|}{|z|} = \frac{1}{\sqrt{3}}$

given  $\arg \frac{z_1}{z} = \frac{\pi}{6}$

$z_1 = \frac{|z_1|}{|z|} \cdot z \cdot e^{i\frac{\pi}{6}}$

$z_1 = \frac{1}{\sqrt{3}} \cdot \frac{\sqrt{3} + i\sqrt{3}}{\sqrt{3}} \cdot z \cdot e^{i\frac{\pi}{6}}$

$z_1 = \frac{\sqrt{3} + i\sqrt{3}}{3} \cdot z \cdot e^{i\frac{\pi}{6}}$

Now,

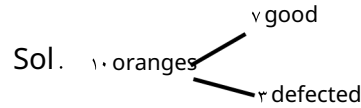
$z_1 - z = \frac{\sqrt{3} + i\sqrt{3}}{3} z \cdot e^{i\frac{\pi}{6}} - z$

$|z_1 - z| = |z|$  ABO is isosceles with angles  $\frac{\pi}{3}, \frac{\pi}{3}$  &  $\frac{\pi}{2}$

19. Three defective oranges are accidentally mixed with seven good ones and on looking at them, it is not possible to differentiate between them. Two oranges are drawn at random from the lot. If x denote the number of defective oranges, then the variance of x is :

- (1)  $\frac{28}{70}$
- (2)  $\frac{14}{70}$
- (3)  $\frac{26}{70}$
- (4)  $\frac{18}{70}$

Ans.



Probability distribution

$X_i$	$P_i$
$X=0$	$\frac{{}^7C_2}{{}^{10}C_2}$
$X=1$	$\frac{{}^7C_1 \cdot {}^3C_1}{{}^{10}C_2}$
$X=2$	$\frac{{}^3C_2}{{}^{10}C_2}$

Now,

$E(X) = 0 \cdot \frac{{}^7C_2}{{}^{10}C_2} + 1 \cdot \frac{{}^7C_1 \cdot {}^3C_1}{{}^{10}C_2} + 2 \cdot \frac{{}^3C_2}{{}^{10}C_2}$

$E(X^2) = 0^2 \cdot \frac{{}^7C_2}{{}^{10}C_2} + 1^2 \cdot \frac{{}^7C_1 \cdot {}^3C_1}{{}^{10}C_2} + 2^2 \cdot \frac{{}^3C_2}{{}^{10}C_2}$

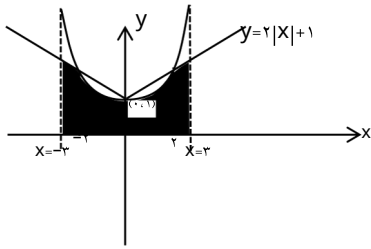
$E(X) = \frac{14}{10}$

$E(X^2) = \frac{28}{10}$  (1)

20. The area (in sq. units) of the region  $\{(x, y) : y \geq |x+1|, y \leq x^2+1, |x| \leq 2\}$  is

- (1)  $\frac{10}{3}$  (2)  $\frac{74}{3}$   
 (3)  $\frac{17}{3}$  (4)  $\frac{32}{3}$

Ans. (2)



Sol.

$$\text{Area} = \int_{-2}^{-1} (x^2 + 1 - (x + 1)) dx + \int_{-1}^1 (x^2 + 1 - |x + 1|) dx + \int_{1}^2 (x^2 + 1 - (x + 1)) dx$$

$$= \frac{74}{3} \quad \text{Ans. (2)}$$

SECTION-B

21. Let  $M$  denote the set of all real matrices of order  $3 \times 3$  and let  $S = \{A \in M : A = A^T \text{ and } a_{ii} = 1, \forall i\}$ .

$$S_1 = \{A \in M : A = -A^T \text{ and } a_{ii} = 0, \forall i\}$$

$$S_2 = \{A \in M : a_{11} + a_{22} + a_{33} = 0 \text{ and } a_{ii} = 1, \forall i\}$$

$$\text{If } n(S_1 \cap S_2 \cap S_3) = 120, \text{ then } n \text{ equals.}$$

Ans. (1613)

$$\text{Sol. } \begin{bmatrix} a_{11} & a_{12} & a_{13} \\ a_{12} & a_{22} & a_{23} \\ a_{13} & a_{23} & a_{33} \end{bmatrix}$$

No. of elements in  $S : A = A^T \Rightarrow 6 \times 6 = 36$

No. of elements in  $S : A = -A^T \Rightarrow 6 \times 6 = 36$

since no zero in  $S$

No. of elements in  $S_3$

$$a_{11} + a_{22} + a_{33} = 0 \Rightarrow (1, 2, -3) \Rightarrow 1$$

or  $(1, 1, -2) \Rightarrow 2$   
 or  $(-1, -1, 2) \Rightarrow 2$

$$\boxed{12 \times 6^6}$$

$$n(S_1 \cap S_2) = 12 \times 6^5$$

$$n(S_1 \cup S_2 \cup S_3) = 36 + 36 + 3 - 12 \times 6^5$$

$$= 72 + 3 - 12 \times 6^5 = 120$$

$$n(S_1 \cap S_2 \cap S_3) = 120$$

22. If  $\vec{a} = \hat{i} + \hat{j} + \hat{k}$ ,  $\vec{b} = \hat{i} + \hat{j} + \hat{k}$  and  $\vec{c} = \hat{i} + \hat{j} + \hat{k}$  then the distance of the point  $(1, 2, \sqrt{3})$  from the line  $\vec{r} = \vec{a} + \lambda \vec{b} + \mu \vec{c}$  is

Ans. (6)

$$\text{Sol. } \vec{a} = \hat{i} + \hat{j} + \hat{k}, \vec{b} = \hat{i} + \hat{j} + \hat{k}, \vec{c} = \hat{i} + \hat{j} + \hat{k}$$

$$\vec{r} = \vec{a} + \lambda \vec{b} + \mu \vec{c} = (\hat{i} + \hat{j} + \hat{k}) + \lambda(\hat{i} + \hat{j} + \hat{k}) + \mu(\hat{i} + \hat{j} + \hat{k})$$

$$= (1 + \lambda + \mu)\hat{i} + (1 + \lambda + \mu)\hat{j} + (1 + \lambda + \mu)\hat{k}$$

$$= (1 + \lambda + \mu)(\hat{i} + \hat{j} + \hat{k})$$

$$= (1 + \lambda + \mu)\sqrt{3}(\frac{\hat{i} + \hat{j} + \hat{k}}{\sqrt{3}})$$

so distance of  $(1, 2, \sqrt{3})$  from  $\vec{r} = \vec{a} + \lambda \vec{b} + \mu \vec{c}$

$$\frac{|(1, 2, \sqrt{3}) \cdot \vec{a}|}{|\vec{a}|} = 6$$

23. Let  $\vec{a} = \hat{i} + \hat{j} + \hat{k}$ ,  $\vec{b} = \hat{i} + \hat{j} + \hat{k}$  and  $\vec{c} = \hat{i} + \hat{j} + \hat{k}$ . If  $\vec{d}$  is a vector such that  $\vec{a} \cdot \vec{c} = \vec{b} \cdot \vec{c} = |\vec{a}| |\vec{c}| \cos \theta$  and the angle between  $\vec{d}$  and  $\vec{c}$  is  $\frac{\pi}{4}$ ,

$|\vec{a} \cdot \vec{b} \cdot \vec{c}|$  is equal to ....

Ans. (6)

$$\text{Sol. } \vec{a} = \hat{i} + \hat{j} + \hat{k}, \vec{b} = \hat{i} + \hat{j} + \hat{k}, \vec{c} = \hat{i} + \hat{j} + \hat{k}$$

$$|\vec{a} \cdot \vec{b}| = 3$$

$$|\vec{a}| = \sqrt{3}, |\vec{b}| = \sqrt{3}$$

$$|\vec{c}| = \sqrt{3}$$

$$|\vec{c}| + \lambda |\vec{a}| = \lambda (\vec{a} \cdot \vec{c}) = \lambda$$

$$3 + \lambda \sqrt{3} - \lambda \sqrt{3} = \lambda$$

$$3 - \lambda \sqrt{3} + \lambda = 0$$

$$|\vec{c}| = 2$$

$$|\vec{d} \cdot \vec{a}| = 6$$

$$|\vec{d} \cdot \vec{b}| = 6$$

$$|\vec{d} \cdot \vec{c}| = 6 \cos \theta$$

$$|\vec{d}| |\vec{c}| \cos \theta = 6 \cos \theta \Rightarrow |\vec{d}| = 6$$

$$\epsilon = \epsilon \vec{b} \cdot (\vec{b} \cdot \vec{c}) \vec{a} - \epsilon (\vec{b} \cdot \vec{c}) (\vec{a} \cdot \vec{b})$$

Let  $b \cdot c = x$

$$\xi = 3\tau + 3X - 2 \cdot X$$

$$3X^2 - 2 \cdot X + 3\tau = 0$$

$$3X^2 - 12X - 18X + 3\tau = 0$$

$$X \in \left[ \frac{1}{3}, \xi \right]$$

$$b \cdot c \in \left[ \frac{1}{3}, \xi \right]$$

$$b \cdot c \in \left[ \frac{1}{3}, \xi \right]$$

Now  $|\xi - 3b \cdot c| + |d \cdot c|$

$$|\xi - 1| + (2)$$

Ans.

Let

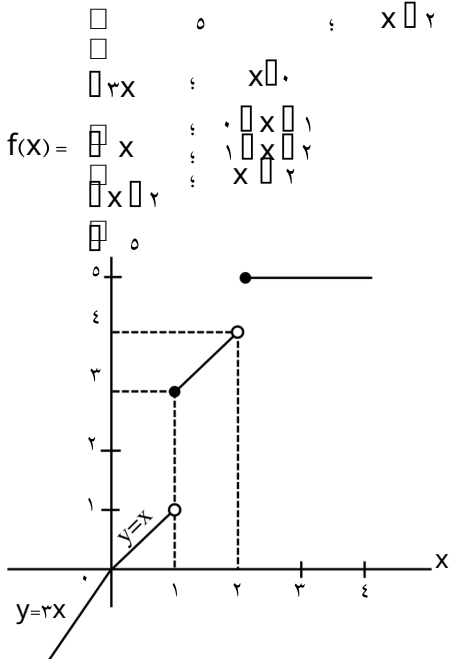
$$f(x) = \min_{x \in \mathbb{R}} \{ \lfloor x \rfloor \cdot \lfloor x \rfloor \}$$

where  $\lfloor x \rfloor$  denotes greatest integer function. If  $\alpha$  and  $\beta$  are the number of points where  $f$  is not continuous and is not differentiable, respectively.

Ans. then  $\alpha + \beta$  equals

$$(a) \lfloor \min_{x \in \mathbb{R}} \{ \lfloor x \rfloor \cdot \lfloor x \rfloor \} \rfloor$$

$$\text{Sol. } f(x) = \lfloor \min_{x \in \mathbb{R}} \{ \lfloor x \rfloor \cdot \lfloor x \rfloor \} \rfloor$$



Not continuous at  $x \in \{1, 2, 3, 4, 5\}$

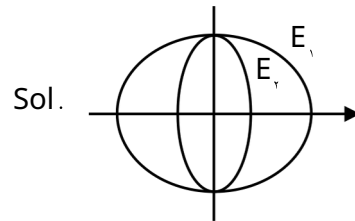
Not diff. at  $x \in \{1, 2, 3, 4, 5\}$

$$\alpha + \beta = 0$$

Let  $E_1 : \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$  be an ellipse. Ellipses  $E_i$  are constructed such that their centres and eccentricities are same as that of  $E_1$  and the length of minor axis of  $E_i$  is the length of major axis of  $E_{i-1}$ . If  $A_i$  is the area of the ellipse  $E_i$ , then

$\sum_{i=1}^{\infty} A_i$  is equal to ...

Ans.  $(\frac{6}{5})$



Sol.

$$E_1 : \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1 \quad e = \sqrt{1 - \frac{b^2}{a^2}} = \frac{\sqrt{a^2 - b^2}}{a}$$

$$E_2 : \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$$

$$e = \frac{\sqrt{a^2 - b^2}}{a} = \sqrt{1 - \frac{b^2}{a^2}} = \frac{a - b}{a} = 1 - \frac{b}{a}$$

$$a = \frac{b}{1 - e}$$

$$E_2 : \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$$

$$E_3 : \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$$

$$e = \frac{\sqrt{a^2 - b^2}}{a} = \sqrt{1 - \frac{b^2}{a^2}} = b \cdot \frac{1 - e}{a}$$

$$E_2 : \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$$

$$A_1 = \pi a b$$

$$A_2 = \pi \left( \frac{a}{3} \right) \left( \frac{b}{3} \right)$$

$$A_3 = \pi \left( \frac{a}{9} \right) \left( \frac{b}{9} \right)$$

$$\sum_{i=1}^{\infty} A_i = \pi a b \left( 1 + \frac{1}{9} + \frac{1}{81} + \dots \right) = \pi a b \left( \frac{1}{1 - \frac{1}{9}} \right) = \frac{9}{8} \pi a b$$

$$\sum_{i=1}^{\infty} A_i = \frac{6}{5} \pi a b$$

**JEE-MAIN EXAMINATION – JANUARY 2025**

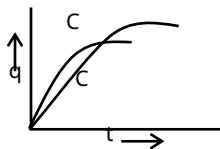
(HELD ON TUESDAY 28<sup>th</sup> JANUARY 2025)

TIME : 9 : 00AM TO 12 : 00 NOON

**PHYSICS**

SECTION-A

26. Two capacitors  $C_1$  and  $C_2$  are connected in parallel to a battery. Charge-time graph is shown below for the two capacitors. The energy stored with them are  $U_1$  and  $U_2$ , respectively. Which of the given statements is true?



- (1)  $C_1 < C_2, U_1 < U_2$       (2)  $C_1 < C_2, U_1 > U_2$   
 (3)  $C_1 < C_2, U_1 > U_2$       (4)  $C_1 < C_2, U_1 < U_2$

Ans. (4)  
 Sol.  $V$  same

$$U = \frac{1}{2} CV^2$$

as  $q_1 > q_2$   
 $C_1 > C_2$   
 $\& U_1 > U_2$

27. In the experiment for measurement of viscosity ' $\eta$ ' of given liquid with a ball having radius  $R$ , consider following statements.

- A. Graph between terminal velocity  $V$  and  $R$  will be a parabola  
 B. The terminal velocities of different diameter balls are constant for a given liquid.  
 C. Measurement of terminal velocity is dependent on the temperature.  
 D. This experiment can be utilized to assess the density of a given liquid.  
 E. If balls are dropped with some initial speed, the value of ' $\eta$ ' will change.  
 Choose the correct answer from the options given below:

**TEST PAPER WITH SOLUTION**

- (1) B, D and E only  
 (2) A, C and D only  
 (3) C, D and E only  
 (4) A, B and E only

Ans. (2)

Sol.  $V \propto R^2$

28. Consider following statements:

- A. Surface tension arises due to extra energy of the molecules at the interior as compared to the molecules at the surface of a liquid.  
 B. As the temperature of liquid rises, the coefficient of viscosity increases.  
 C. As the temperature of gas increases, the coefficient of viscosity increases.  
 D. The onset of turbulence is determined by Reynold's number.  
 E. In a steady flow two stream lines never intersect.

Choose the correct answer from the options given below:

- (1) A, B, E only  
 (2) B, C, E only  
 (3) A, B, C only

Ans. (2)

29. Three infinitely long wires with linear charge density  $\lambda$  are placed along the x-axis, y-axis and z-axis respectively. Which of the following denotes

an equipotential surface?

- (1)  $xy + yz + zx = \text{constant}$   
 (2)  $(x + y)(y + z)(z + x) = \text{constant}$   
 (3)  $(x + y)(y + z)(z + x) = \text{constant}$   
 (4)  $xyz = \text{constant}$

Ans. (r)

Sol. 
$$V = \int \vec{E} \cdot d\vec{r} = \int \frac{k\lambda}{r} dr = k\lambda \ln r + c$$

Net potential due to all wire

$$V = k\lambda \ln \sqrt{x^2 + y^2} + k\lambda \ln \sqrt{y^2 + z^2} + k\lambda \ln \sqrt{z^2 + x^2} + c$$

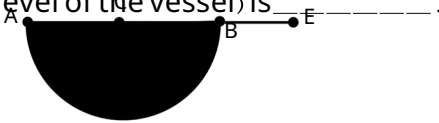
for  $V = c$

$$\sqrt{x^2 + y^2} \sqrt{y^2 + z^2} \sqrt{z^2 + x^2} = C$$

$$x^2 y^2 y^2 z^2 z^2 x^2 = C^2$$

where  $C = \text{constant}$

30. A hemispherical vessel is completely filled with a liquid of refractive index  $\mu$ . A small coin is kept at the lowest point (O) of the vessel as shown in figure. The minimum value of the refractive index of the liquid so that a person can see the coin from point E (at the level of the vessel) is



(1)  $\sqrt{3}$

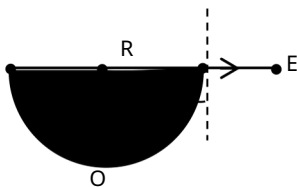
(2)  $\frac{3}{2}$

(3)  $\sqrt{2}$

(4)  $\frac{\sqrt{3}}{2}$

Ans. (r)

Sol.



$\sin c = \frac{1}{\mu}$

for  $\mu$  least,  $c = \text{maximum}$

$\mu = c = \epsilon_0$

$\epsilon_0 = \frac{1}{\sin \epsilon_0} = \sqrt{2}$

31.

Consider a long thin conducting wire carrying a uniform current  $I$ . A particle having mass "M" and charge "q" is released at a distance "a" from the wire with a speed  $v$  along the direction of current. The particle gets attracted to the wire due to magnetic force. The particle turns round when it is at distance  $x$  from the wire. The value of  $x$  is  $\frac{a}{2}$  is vacuum permeability  $\mu_0$

(1)  $a \frac{mv_0}{2q\mu_0 I}$

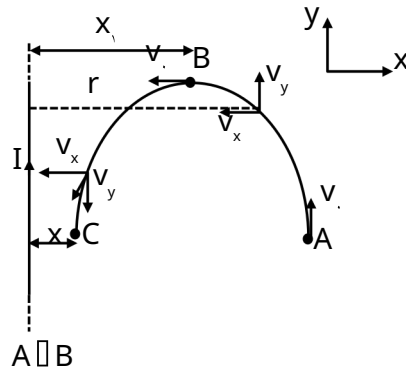
(2)  $\frac{a}{2}$

(3)  $a \frac{1}{\mu_0} \frac{mv_0}{q\mu_0 I}$

(4)  $ae^{-\frac{4\mu_0 m v_0}{q\mu_0 I}}$

Ans. (4)

Sol.



$\vec{V} = -v_x \hat{i} + v_y \hat{j}$

$\vec{B} = \frac{\mu_0 I}{2r} (-\hat{k})$

$\vec{F} = q(\vec{v} \times \vec{B}) = \frac{\mu_0 I q}{2r} (-v_x \hat{j} - v_y \hat{i})$

$a_x = -\frac{\mu_0 I q}{2m} \cdot \frac{v_y}{r}$

$a_y = -\frac{\mu_0 I q}{2m} \cdot \frac{v_x}{r}$

$\frac{v_x dv_x}{dr} = -\frac{\mu_0 I q}{2m r} v_y$

$\frac{v_x dv_x}{v_y} = -\frac{\mu_0 I q}{2m} \frac{dr}{r}$

$\int \frac{v_x dv_x}{\sqrt{v_0^2 - v_x^2}} = \frac{\mu_0 I q}{2m} \int \frac{dr}{r}$

Let:  $z = v_0^2 - v_x^2$

$r dz = -2v_x dv_x$

$$z dz = -v dv_x$$

$$\frac{v_x dv_x}{\sqrt{v_0^2 - v_x^2}} = \frac{z dz}{z} = -dz$$

then integral becomes

$$\int_{v_0}^0 \frac{v_x dv_x}{\sqrt{v_0^2 - v_x^2}} = \int_{\frac{0}{2m}}^{\frac{Qq}{2m}} \frac{z dz}{z} = -dz$$

$$v_x = -\frac{Qq}{2m} \ln \frac{x_1}{a}$$

$$x_1 = a e^{-\frac{2m v_x}{Qq}} \dots \dots (1)$$

For B  $\hat{i}$   $\hat{j}$   $\hat{k}$

$$\vec{v} = v_x \hat{i} - v_y \hat{j}$$

$$\vec{B} = \frac{0I}{2r} (-\hat{k})$$

$$\vec{F} = q(\vec{v} \times \vec{B}) = \frac{0Iq}{2r} (-v_x \hat{j} \times v_y \hat{i})$$

$$a_x = + \frac{0Iqv_y}{2mr} \quad a_y = - \frac{0Iqv_x}{2mr}$$

$$\frac{v_x dv_x}{dr} = \frac{0Iqv_y}{2mr}$$

$$\int_{v_0}^0 \frac{v_x dv_x}{\sqrt{v_0^2 - v_x^2}} = \int_{\frac{0Iq}{2m}}^{\frac{0Iq}{2m}} \frac{dr}{r}$$

$$\frac{0Iq}{2m} \ln \frac{x}{x_1} = \int_{v_0}^0 \frac{v_x dv_x}{\sqrt{v_0^2 - v_x^2}} - v_0$$

$$x = x_1 e^{-\frac{2m v_0}{Qq}} \dots \dots (2)$$

From equation (1) and (2)

$$x = a e^{-\frac{4m v_0}{Qq}}$$

32. A Carnot engine (E) is working between two temperatures  $\epsilon_1$  K and  $\epsilon_2$  K. In a new system two engines - engine E works between  $\epsilon_1$  K to  $\epsilon_2$  K and engine E' works between  $\epsilon_1$  K to  $\epsilon_2$  K. If  $\eta_1$  and  $\eta_2$  are the efficiencies of the engines E, E' and E'', respectively, then

- (1)  $\eta_1 > \eta_2 + \eta_3$
- (2)  $\eta_1 = \eta_2 + \eta_3$
- (3)  $\eta_1 = \eta_2 \eta_3$
- (4)  $\eta_1 \eta_2 \eta_3 = \eta_1 + \eta_2 + \eta_3$

Ans. (1)

Sol.  $\eta_1 = 1 - \frac{\epsilon_2}{\epsilon_1} = \frac{\epsilon_1 - \epsilon_2}{\epsilon_1}$

(1)  $\eta_1 = \frac{\epsilon_1 - \epsilon_2}{\epsilon_1} = \frac{100 - 40}{100} = 0.6$

(2)  $\eta_2 = \frac{\epsilon_1 - \epsilon_2}{\epsilon_2} = \frac{100 - 40}{40} = 1.5$

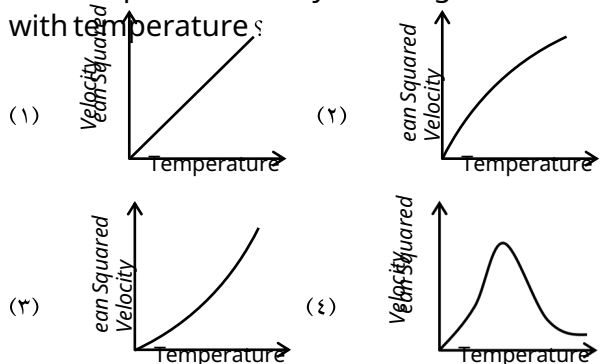
33. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R. Assertion A: A sound wave has higher speed in solids than gases. Reason R: Gases have higher value of Bulk modulus than solids. In the light of the above statements, choose the

- (1) Both A and R are true and R is the correct explanation of A
- (2) A is false but R is true
- (3) Both A and R are true but R is NOT the correct explanation of A
- (4) A is true but R is false.

Ans. (4)

Sol. (4)

34. Solids have higher value of bulk modulus than gases. For a particular ideal gas which of the following graphs represents the variation of mean squared velocity of the gas molecules with temperature?



Ans. (1)

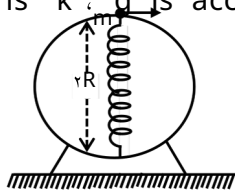
Sol.  $V_{rms} = \sqrt{\frac{3RT}{M}}$

$V_{rms}^2 = \frac{3RT}{M}$

Hence we can conclude that  $V_{rms}^2$  is directly proportional to temperature  $y = mx$

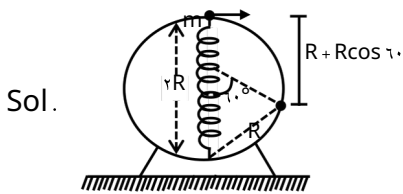
Graph will be straight line

30. A bead of mass 'm' slides without friction on the wall of a vertical circular hoop of radius 'R' as shown in figure. The bead moves under the combined action of gravity and a massless spring (k) attached to the bottom of the hoop. The equilibrium length of the spring is 'R'. If the bead is released from top of the hoop with (negligible) zero initial speed, velocity of bead, when the length of spring becomes 'R', would be (spring constant is 'k', g is acceleration due to gravity)



- (1)  $2\sqrt{gR\left[\frac{kR^2}{m}\right]}$       (2)  $\sqrt{2Rg\left[\frac{4kR^2}{m}\right]}$   
 (3)  $\sqrt{2Rg\left[\frac{kR^2}{m}\right]}$       (4)  $\sqrt{3Rg\left[\frac{m}{k}\right]}$

Ans. (4)



Sol.

Work energy theorem

$$Mg(R + R\cos\theta) + \frac{1}{2}k(R - R)^2 = \frac{1}{2}mv^2$$

$$Mg\frac{3R}{2} = \frac{1}{2}mv^2$$

$$v = \sqrt{3gR\left[\frac{m}{k}\right]}$$

36. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R

Assertion A: In a central force field, the work done is independent of the path chosen

Reason R: Every force encountered in mechanics does not have an associated potential energy.

In the light of the above statements, choose the most appropriate answer from the options given below (1) A is true but R is false (2) Both A and R are true but R is NOT the correct explanation of A (3) Both A and R are true and R is the correct explanation of A (4) A is false but R is true

Ans. (2)

Sol. Both statement are correct but Reason is not the correct explanation of Assertion.

37. Choose the correct nuclear process from the below options

p: proton, n: neutron, e: electron, e+: positron,  $\bar{\nu}$ : antineutrino

(1)  $n \rightarrow p + e + \bar{\nu}$       (2)  $n \rightarrow p + e + \nu$

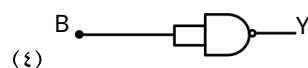
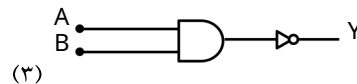
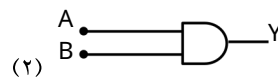
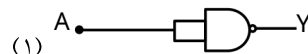
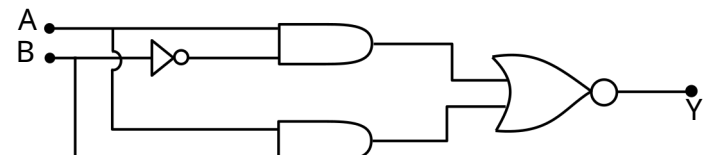
(3)  $n \rightarrow p + e + \nu$       (4)  $n \rightarrow p + e + \bar{\nu}$

Ans. (1)

Sol. Theoretical equation for  $\beta^-$  decay

$$n_0^1 \rightarrow p_1^1 + e_{-1}^0 + \bar{\nu}$$

38. Which of the following circuits has the same output as that of the given circuit:

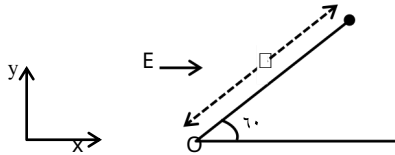


Ans. (1)



εγ. A particle of mass 'm' and charge 'q' is fastened to one end 'A' of a massless string having equilibrium length ℓ, whose other end is fixed at point 'O'. The whole system is placed on a frictionless horizontal plane and is initially at rest. If uniform electric field is switched on along the direction as shown in figure, then the speed of the particle when it crosses the x-axis is

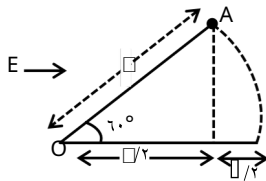
A



- (1)  $\sqrt{\frac{2qE\ell}{m}}$       (2)  $\sqrt{\frac{qE\ell}{4m}}$   
 (3)  $\sqrt{\frac{qE\ell}{m}}$       (4)  $\sqrt{\frac{qE\ell}{2m}}$

Ans. (3)

Sol.



$$W_{\text{all}} = k$$

$$W_e = k_f = k_i$$

$$qE\ell = \frac{1}{2}mv^2 - 0$$

$$v = \sqrt{\frac{qE\ell}{m}}$$

εδ. A proton of mass 'm<sub>p</sub>' has same energy as that of a photon of wavelength 'λ'. If the proton is moving at non-relativistic speed, then ratio of its de Broglie wavelength to the wavelength of photon is.

- (1)  $\frac{12E}{cm_p}$       (2)  $\frac{1}{c}\sqrt{\frac{E}{m_p}}$   
 (3)  $\frac{1}{c^2m_p}\sqrt{E}$       (4)  $\frac{1}{2c}\sqrt{\frac{E}{m_p}}$

Ans. (3)

sol. E is missing in the question but considering E as energy, the solution will be

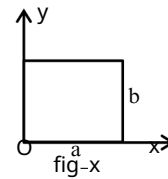
$$E_{\text{photon}} = \frac{hc}{\lambda} = E; E_{\text{proton}} = \frac{1}{2}m_p v^2 = E$$

$$\frac{\lambda_{\text{proton}}}{\lambda_{\text{photon}}} = \frac{h/p}{hc/E} = \frac{h/\sqrt{2m_p E}}{hc/E}$$

$$= \frac{E}{c\sqrt{2m_p E}}$$

$$\frac{\lambda_{\text{proton}}}{\lambda_{\text{photon}}} = \frac{c}{\sqrt{2m_p E}}$$

εε. The centre of mass of a thin rectangular plate (fig - x) with sides of length a and b, whose mass per unit area (σ) varies as σ = σ<sub>0</sub>x / ab (where σ<sub>0</sub> is a constant), would be \_\_\_\_\_

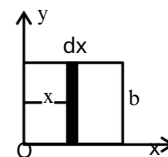


- (1)  $\frac{2}{3}a, \frac{2}{3}b$       (2)  $\frac{2}{3}a, \frac{2}{3}b$   
 (3)  $\frac{a}{2}, \frac{b}{2}$       (4)  $\frac{1}{3}a, \frac{b}{2}$

Ans. (1)

Sol. σ is constant in y-direction

$$\text{So, } y_{\text{cm}} = b/2$$



$$x_{\text{cm}} = \frac{\int_0^a x dm}{\int_0^a dm}$$



$$\frac{I_1}{I_2} = 2.5 \times \frac{\frac{MR_1^2}{4}}{\frac{MR_2^2}{2}} = \frac{5}{2} \times \frac{R_1^2}{R_2^2} \dots (1)$$

Now we are provided with information that

$$\frac{I_3}{I_2} = n$$

$$\frac{2MR_1^2}{5MR_2^2} = n = \frac{4R_1^2}{5R_2^2} \dots (2)$$

From Eq', (1) and (2)  $n = \frac{2}{5}$ . In a measurement, it is asked to find modulus of elasticity per unit torque applied on the system.

The measured quantity has dimension of  $\frac{ML^2}{T^2}$ . If  $b = \gamma$ , the value of  $c$  is \_\_\_\_\_

NTA Ans. (2)

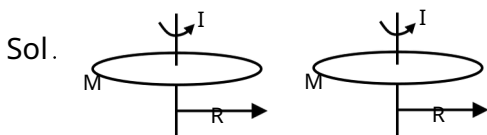
Sol.  $\frac{\text{modulus of elasticity}}{\text{Torque}} = \frac{\text{Stress}}{\text{strain} \times \text{torque}}$

$$= \frac{[\text{Force}]}{[\text{Area}] \times [\text{Force} \times \text{length}]}$$

$$= \frac{1}{[\text{Area} \times \text{length}]} \dots [L^{-3}]$$

Q9. Two iron solid discs of negligible thickness have radii  $R$  and  $\gamma R$  and moment of inertia  $I$  and  $I$  respectively. For  $R = \gamma R$ , the ratio of  $I$  and  $I$  would be  $1/x$ , where  $x =$  \_\_\_\_\_

Ans. (16)



Given  $R_2 = \gamma R_1$

$$M_1 = \frac{4}{3} \pi R_1^3 \rho = M_0$$

$$M_2 = \frac{4}{3} \pi R_2^3 \rho = M_0$$

$$M_1 = \frac{4}{3} \pi R_1^3 \rho = M_0 \quad [2R_1]^3 = 8R_1^3 \quad 4R_1^2 \times 4M_0$$

$$\frac{I_1}{I_2} = \frac{M_1 R_1^2}{M_2 R_2^2} = \frac{1}{4} \times \frac{1}{4} = \frac{1}{16}$$

Q10. A double slit interference experiment performed with a light of wavelength  $600 \text{ nm}$  forms an interference fringe pattern on a screen with  $10^{\text{th}}$  bright fringe having its centre at a distance of  $10 \text{ mm}$  from the central maximum. Distance of the centre of the same  $10^{\text{th}}$  bright fringe from the central maximum when the source of light is replaced by another source of wavelength  $660 \text{ nm}$  would be \_\_\_\_\_ mm.

Ans. (11)

Sol. In case of YDSE the distance of  $n^{\text{th}}$  maxima from central maxima is given by

$$y = \frac{n \lambda D}{d}$$

Here  $n, D$  &  $d$  are same

So,  $y \propto \lambda$

$$\frac{y_1}{y_2} = \frac{\lambda_1}{\lambda_2} \Rightarrow \frac{10 \text{ mm}}{y_2} = \frac{600 \text{ nm}}{660 \text{ nm}}$$

$$y_2 = 11 \text{ mm}$$

**JEE-MAIN EXAMINATION – JANUARY 2025**

(HELD ON TUESDAY 28<sup>th</sup> JANUARY 2025)

TIME : 9 : 00AM TO 12 : 00 NOON

**CHEMISTRY**

SECTION-A

Q1. The incorrect decreasing order of atomic radii is :

- (1)  $Mg < Al < C < O$                       (2)  $Al < B < N < F$   
 (3)  $Be < Mg < Al < Si$                     (4)  $Si < P < Cl < F$

Ans. (3)

Sol. Correct order of atomic radii :  $Be > Mg < Al < Si$

Q2. Given below are two statements :

Statement I : In the oxalic acid vs  $KMnO_4$  (in the presence of dil  $H_2SO_4$ ) titration the solution needs to be heated initially to  $60^\circ C$ , but no heating is required in Ferrous ammonium sulphate (FAS) vs

$KMnO_4$  titration (in the presence of dil  $H_2SO_4$ )

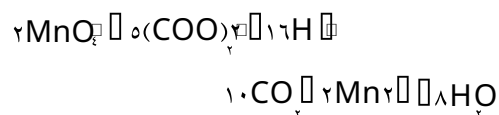
Statement II : In oxalic acid vs  $KMnO_4$  titration, the initial formation of  $MnSO_4$  takes place at high temperature, which then acts as catalyst for further reaction.

In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is false but Statement II is true  
 (2) Both Statement I and Statement II are true  
 (3) Statement I is true but Statement II is false  
 (4) Both Statement I and Statement II are false

Ans. (2)

Sol. For the titration : Oxalic acid v/s  $KMnO_4$



This reaction is slow at room temperature, but becomes fast at  $60^\circ C$ . Manganese(II) ions catalyse the reaction; thus, the reaction is autocatalytic; once manganese(II) ions are formed, it becomes faster and faster.

**TEST PAPER WITH SOLUTIONS**

The titration of FAS v/s  $KMnO_4$  do not require heating because at higher temperature the oxidation of  $Fe^{2+}$  to  $Fe^{3+}$  by atmospheric  $O_2$  will be prominent.

Q3. Match the List-I with List-II

	List-I (Redox Reaction)		List-II (Type of Redox Reaction)
A	$CH_4(g) + 2O_2(g) \rightarrow CO_2(g) + 2H_2O(l)$	(I)	Disproportionation reaction
B	$2NaH(s) + 2H_2O(l) \rightarrow 2NaOH(aq) + H_2(g)$	(II)	Combination reaction
C	$2V_2O_5(s) + 5Ca(s) \rightarrow 2V_2O(s) + 5CaO(s)$	(III)	Decomposition reaction
D	$2H_2O_2(aq) \rightarrow 2H_2O(l) + O_2(g)$	(IV)	Displacement reaction

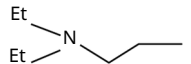
Choose the correct answer from the options given below :

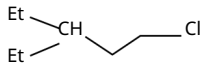
- (1) A-II, B-III, C-IV, D-I  
 (2) A-II, B-III, C-I, D-IV  
 (3) A-III, B-IV, C-I, D-II  
 (4) A-IV, B-I, C-II, D-III

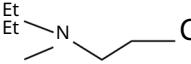
Ans. (1)

Sol. (A) Combustion of hydrocarbon  
 (B) Decomposition into gaseous product.  
 (C) Displacement of „V" by „Ca" atom.  
 (D) Disproportionation of  $H_2O_2$  into  $O_2$  and  $O$  oxidation states.

04. Given below are two statements :

Statement I :  will undergo alkaline hydrolysis at a faster rate than

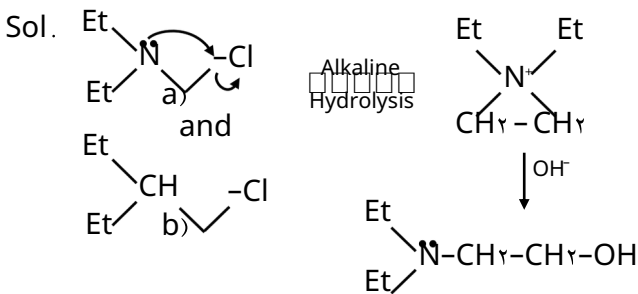


Statement II :  intramolecular substitution takes place first by involving lone pair of electrons on nitrogen.

In the light of the above statements, choose the most appropriate answer from the options given below :

- (1) Both Statement I and Statement II are incorrect
- (2) Statement I is incorrect but statement II is correct
- (3) Both Statement I and Statement II are correct
- (4) Statement I is correct but Statement II is incorrect

Ans. (3)

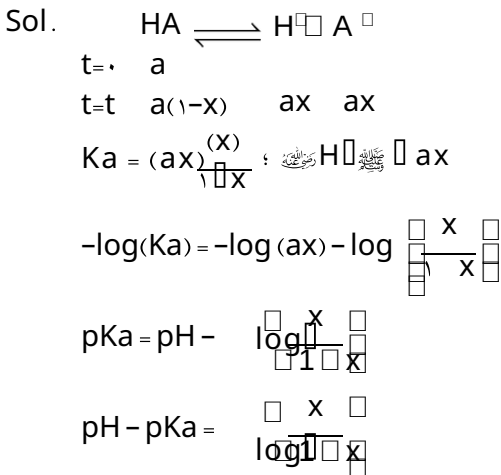


Rate of (a) is faster than rate of (b) because it is an intramolecular substitution. A weak acid HA has degree of dissociation  $\alpha$ . Which option gives the correct expression of

$\text{pH} = \text{pK}_a + \dots$

- (1)  $\log(1 + \alpha x)$
- (2)  $\log \frac{1 + \alpha x}{x}$
- (3)  $\dots$
- (4)  $\log \frac{x}{1 + \alpha x}$

Ans. (4)



06. Consider „n” is the number of lone pair of electrons present in the equatorial position of the most stable

structure of  $\text{ClF}_3$ . The ions from the following

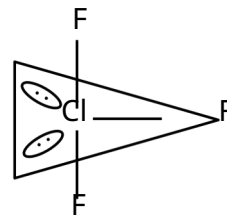
- A.  $\text{M}^+$
- B.  $\text{Ni}^+$
- C.  $\text{Cu}^+$
- D.  $\text{Ni}^+$
- E.  $\text{Ti}^{2+}$

Choose the correct answer from the options given below :

- (1) A and C only
- (2) A, D and E only
- (3) B and C only
- (4) B and D only

Ans. (2)

Sol.  $\text{ClF}_3$



$n = 2$  (No of lone pair present in equatorial plane) (Unpaired e)

- (A)  $\text{V}^{+3} : [\text{Ar}]3d^2 \quad 2$
- (B)  $\text{Ti} : [\text{Ar}]3d^2 \quad 1$
- (C)  $\text{Cu} : [\text{Ar}]3d^9 \quad 1$
- (D)  $\text{Ni} : [\text{Ar}]3d^8 \quad 2$
- (E)  $\text{Ti} : [\text{Ar}]3d^2 \quad 2$

07.

$A$ / mol $l^{-1}$	$t_{1/2}$ / min
0.100	200
0.200	100

For a given reaction  $R \rightarrow P$ ,  $t_{1/2}$  is related to  $A$ , as given in table :

Given :  $\log 2 = 0.3$

Which of the following is true :

- A. The order of the reaction is  $\frac{1}{2}$
  - B. If  $A$  is 1 M, then  $t_{1/2}$  is  $\sqrt{2} \times 100$  min
  - C. The order of the reaction changes to 1 if the concentration of reactant changes from 0.100 M to 0.500 M.
  - D.  $t_{1/2}$  is 800 min for  $A = 1.1$  M
- Choose the correct answer from the options given below :

- (1) A and C only
- (2) A and B only
- (3) A, B and D only
- (4) C and D only

Ans. (3)

Sol.  $t_{1/2} \propto \frac{1}{A^{n-1}}$

$$\frac{(t_{1/2})_1}{(t_{1/2})_2} = \left(\frac{A_2}{A_1}\right)^{n-1}$$

$$\frac{200}{100} = \left(\frac{1}{4}\right)^{n-1}$$

$$2 = \frac{1}{4^{n-1}}$$

$n - 1 = \frac{1}{2}$

$n = \frac{3}{2}$  (order)

$t_{1/2} \propto \frac{1}{\sqrt{A}}$

$$\frac{200}{100} = \left(\frac{1}{4}\right)^{1/2}$$

when  $A_0 = 1 \text{ M}$

$t_{1/2} = 200 \cdot \sqrt{4} \text{ min}$

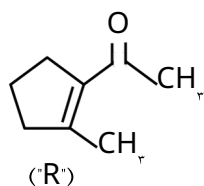
❖ 1st order kinetics have  $t_{1/2}$  independent of their concentration. So upon changing the concentration  $t_{1/2}$  should not change for first order reaction.

$\frac{200}{100} = \left(\frac{1}{4}\right)^{1/2}$

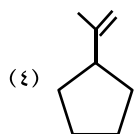
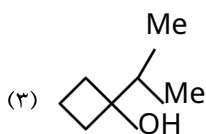
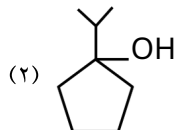
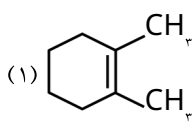
when  $A_0 = 1.6 \text{ M}$

$t_{1/2} = 800 \text{ min}$

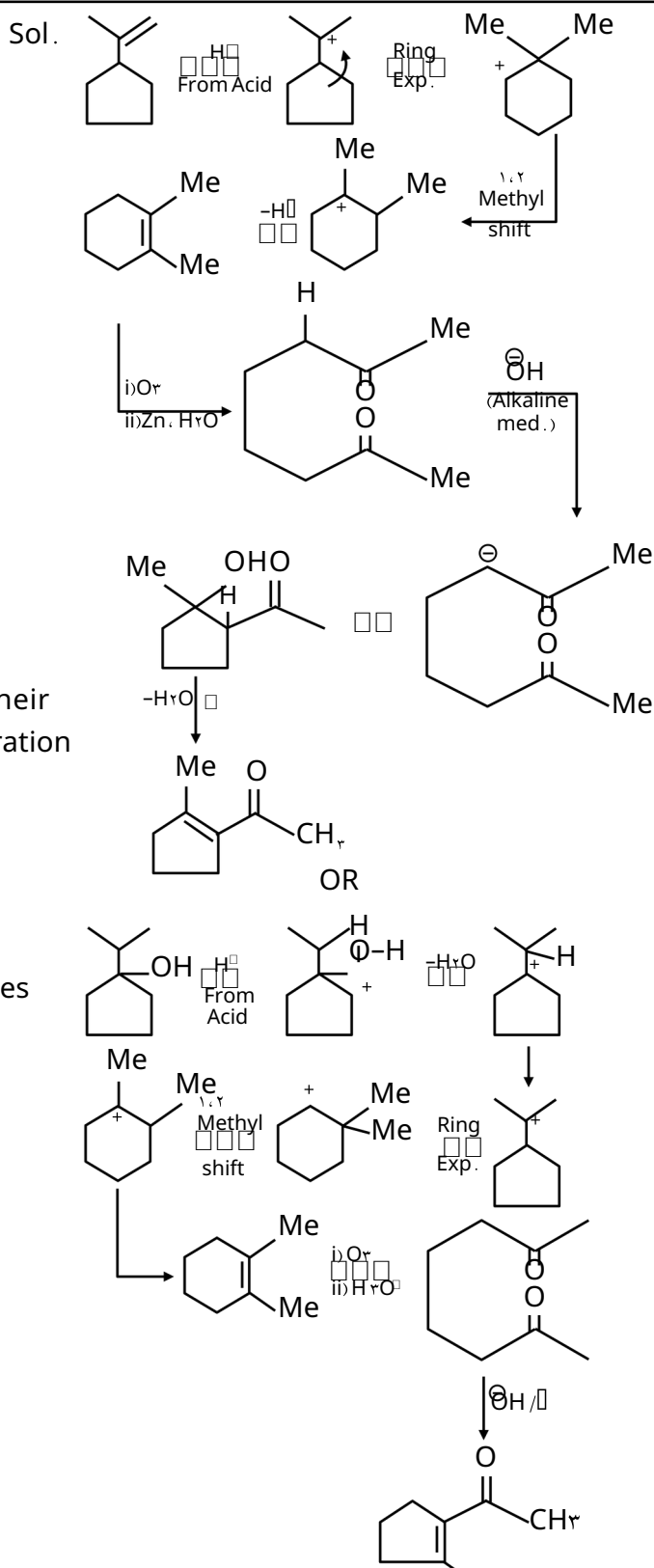
Q. A molecule (P) on treatment with acid undergoes rearrangement and gives ozonolysis followed by reflux under alkaline condition gives (R). The structure of (R) is given below :



The structure of (P) is



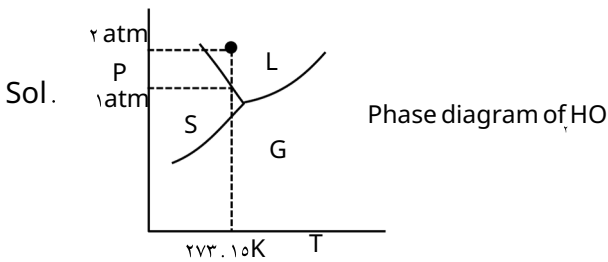
NTA Ans. (2)



Note : In question about molecule "P" is not clarified, weather it is alcohol or alkene and as in question language rearrangement product is asking hence according to question language ans. is either (2) or (4). As alkene also undergoes rearrangement in presence of acid but option (2) also fulfil all conditions.

59. Ice and water are placed in a closed container at a pressure of 1 atm and temperature 273.15 K. If pressure of the system is increased  $\gamma$  times, keeping temperature constant, then identify correct observation from following : (1) Volume of system increases. (2) Liquid phase disappears completely. (3) The amount of ice decreases. (4) The solid phase (ice) disappears completely. (5)

Ans. (2)



If pressure is made two times then mixture of ice and water will completely convert into water (liquid) form. The molecules having

60. square pyramidal geometry

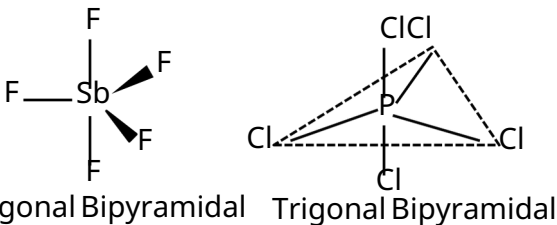
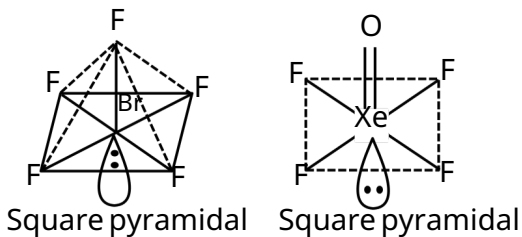
are (1)  $BrF_5$  &  $XeOF_4$

(2)  $SbF_5$  &  $PCl_5$

(3)  $SbF_5$  &  $XeOF_4$

(4)  $BrF_5$  &  $PCl_5$

Ans. (1)



Sol.

$BrF_5$  : Square pyramidal

$XeOF_4$  : Square pyramidal

$SbF_5$  : Trigonal bipyramidal

$PCl_5$  : Trigonal bipyramidal

61. The metal ion whose electronic configuration is not affected by the nature of the ligand and which gives a violet colour in non-luminous flame under hot condition in borax bead test

(1)  $Ti^{3+}$

(2)  $Ni^{2+}$

(3)  $Mn^{2+}$

(4)  $Cr^{3+}$

Ans. (2)

Sol. Ni gives violet coloured bead in non-luminous

flame under hot conditions.  $Ni^{2+}$  has  $d^8$  configuration which does not depend on nature of ligand present in octahedral complex.

$Ni^{2+}$  configuration

62.

Both acetaldehyde and acetone (individually) undergo which of the following reactions:

A. Iodoform Reaction

B. Cannizaro Reaction

C. Aldol condensation

D. Tollen's Test

E. Clemmensen Reduction

Choose the correct answer from the options given below :

(1) A, B and D only

(2) A, C and E only

(3) C and E only

(4) B, C and D only

Ans. (2)

Sol.

S.No.	Name of Reaction	Acetaldehyde $CH_3-C(=O)-H$	Acetone $CH_3-C(=O)-CH_3$
1	Iodoform reaction	[-ve]	[-ve]
2	Cannizaro	[+ve]	[+ve]
3	Aldol reaction	[-ve]	[-ve]
4	Tollen's test	[-ve]	[+ve]
5	Clemmensen reduction	[-ve]	[-ve]

Ans. (2) A, C and E only

73. In a multielectron atom, which of the following orbitals described by three quantum numbers will have same energy in absence of electric and magnetic fields:

- A.  $n = 1, l = 0, m = 0$
- B.  $n = 2, l = 0, m = 0$
- C.  $n = 2, l = 1, m = 1$
- D.  $n = 3, l = 2, m = 1$
- E.  $n = 3, l = 2, m = 0$

Choose the correct answer from the options given below :

- (1) A and B only
- (2) B and C only
- (3) C and D only
- (4) D and E only

Ans. (4)

Sol. orbital

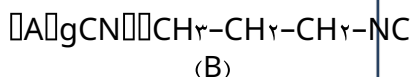
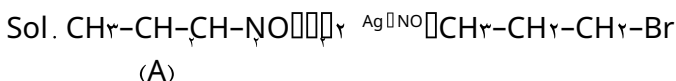
- A :  $n = 1, l = 0, m = 0$        $1s$
- B :  $n = 2, l = 0, m = 0$        $2s$
- C :  $n = 2, l = 1, m = 1$        $2p$
- D :  $n = 3, l = 2, m = 1$        $3d$
- E :  $n = 3, l = 2, m = 0$        $3d$

In absence of electric and magnetic fields, all orbitals of  $3d$  are degenerate

74. The products A and B in the following reactions respectively are

- A  $\xrightarrow{AgNO_3}$   $CH_3-CH_2-CH_2-Br$   $\xrightarrow{AgCN}$  B
- (1)  $CH_3-CH_2-CH_2-ONO_2$ ,  $CH_3-CH_2-CH_2-NC$
  - (2)  $CH_3-CH_2-CH_2-ONO_2$ ,  $CH_3-CH_2-CH_2-CN$
  - (3)  $CH_3-CH_2-CH_2-NO_2$ ,  $CH_3-CH_2-CH_2-CN$
  - (4)  $CH_3-CH_2-CH_2-NO_2$ ,  $CH_3-CH_2-CH_2-NC$

Ans. (4)



75. What is the freezing point depression constant of a solvent, 50 g of which contain 1 g solute volatile mass (206 g mol) and the decrease in freezing point is 0.4 K?

- (1)  $0.12 \text{ K kg mol}^{-1}$
- (2)  $0.12 \text{ K kg mol}^{-1}$
- (3)  $0.12 \text{ K kg mol}^{-1}$
- (4)  $0.12 \text{ K kg mol}^{-1}$

Ans. (1)

Sol.  $\Delta T_f = K_b \cdot m$

$$0.4 = K_b \frac{1}{50 \times \frac{1}{206}}$$

$$K_b = 0.12 \text{ K kg / mol}$$

76. Consider the following elements In, Tl, Al, Pb, Sn and Ge.

The most stable oxidation states of elements with highest and lowest first ionisation enthalpies, respectively, are

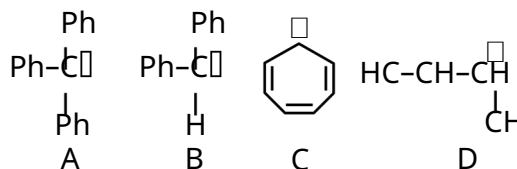
- (1) +2 and +3
- (2) +3 and +4
- (3) +3 and +1
- (4) +1 and +3

NTA Ans. (3)

Sol. Among Al, In, Tl, Ge, Sn, Pb, the metal having highest IE is Ge and lowest IE is In.

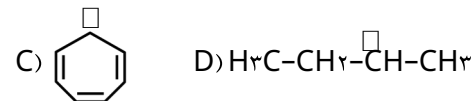
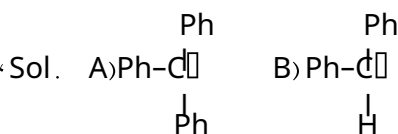
Most stable oxidation state of Ge is +4 and In is +3.

77. The correct order of stability of following carbocation is :



- (1)  $A < B < C < D$
- (2)  $B < C < A < D$
- (3)  $C < B < A < D$
- (4)  $C < A < B < D$

Ans.



Solution :-

C is aromatic due to +ve charge hence it is most stable

A have more resonance structure

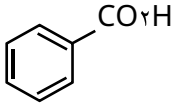
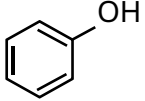
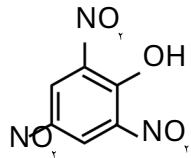
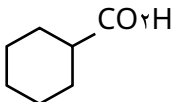
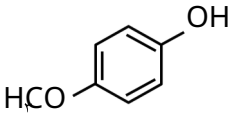
B have less resonance structure

D have only hyper conjugation

Consider First Aromaticity < Resonance < Hyper conjugation

Ans.  $D > B > A > C$

78. The compounds that produce  $\text{CO}_2$  with aqueous  $\text{NaHCO}_3$  solution are :

- A. 
- B. 
- C. 
- D. 
- E. 

Choose the correct answer from the options given below :

- (1) A and C only                      (2) A, B and E only  
(3) A, C and D only                  (4) A and B only

Ans. (3)

Sol. A, C, D produce  $\text{CO}_2$  with aqueous  $\text{NaHCO}_3$  solution.

A, C, D acids are stronger acid than  $\text{H}_2\text{CO}_3$  (Carbonic acid)

79. Which of the following oxidation reactions are carried out by both  $\text{K}_2\text{Cr}_2\text{O}_7$  and  $\text{KMnO}_4$  in acidic medium :

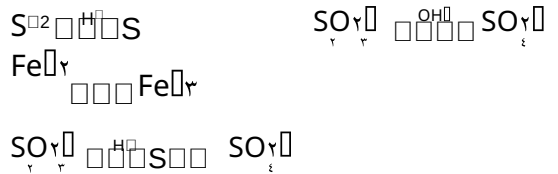
- A.  $\text{I}^- \rightarrow \text{I}_2$   
B.  $\text{S}^{2-} \rightarrow \text{S}$   
C.  $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+}$   
D.  $\text{I}^- \rightarrow \text{IO}_3^-$   
E.  $\text{S}_2\text{O}_3^{2-} \rightarrow \text{SO}_4^{2-}$

Choose the correct answer from the options given below :

- (1) B, C and D only                      (2) A, D and E only  
(3) A, B and C only                      (4) C, D and E only

Ans. (3)

Sol.  $\text{I}^- \rightarrow \text{I}_2$                                        $\text{I}^- \rightarrow \text{IO}_3^-$



80. Given below are two statements :

Statement I : D-glucose pentaacetate reacts with  $\alpha, \beta$ -dinitrophenylhydrazine.

Statement II : Starch, on heating concentrated sulfuric acid at  $100^\circ\text{C}$  and  $2-3$  atmosphere pressure produces glucose.

In the light of the above statements, choose the correct answer from the options given below

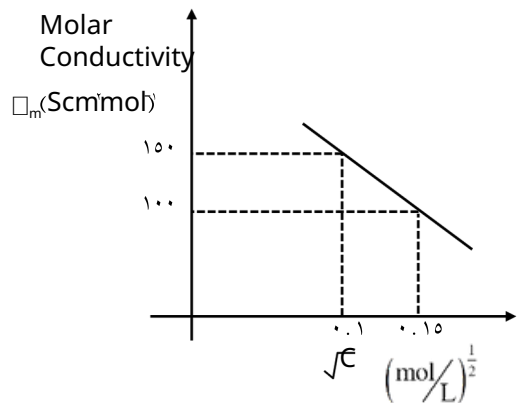
- (1) Both Statement I and Statement II are false  
(2) Statement I is false but Statement II is true  
(3) Statement I is true but Statement II is false  
(4) Both Statement I and Statement II are true

Ans. (2)

Sol.

SECTION-B

81. Given below is the plot of the molar conductivity vs  $\sqrt{\text{concentration}}$  for KCl in aqueous solution.



If, for the higher concentration of KCl solution, the resistance of the conductivity cell is  $100 \Omega$ , then the resistance of the same cell with the dilute solution is „x“  $\Omega$ . The value of x is \_\_\_\_\_ (Nearest integer)

Ans. 150

Sol.  $R = \frac{\rho \ell}{A}$

$\rho = G \cdot G \diamond G = \frac{\rho \ell}{A}$

$G \diamond = \frac{\ell}{A}$

R = Resistance

$\rho$  = Resistivity

$\frac{\ell}{A}$  cell constant ( $G \diamond$ )

$\frac{\rho \ell}{A} = \frac{R}{G \diamond} \Rightarrow \rho = \frac{R \cdot A}{\ell}$

c concentrated sol.  
d diluted solution

$\frac{100 \cdot (0.10)^2}{100 \cdot (0.1)^2} = \frac{R_d}{100}$

$R_d = 100$

22. Quantitative analysis of an organic compound (X) shows following % composition.

C : 14.0% Cl : 74.46%

H : 1.8%

(Empirical formula mass of the compound (X) is \_\_\_\_\_)

(Given molar mass in g mol<sup>-1</sup> of C : 12, H : 1,

O : 16, Cl : 35.5)

Ans. 160

Sol. C : Cl : H : O

%mass 14.0 74.46 1.8 19.24

Molar ratio	$\frac{14.0}{12}$	$\frac{74.46}{35.5}$	$\frac{1.8}{1}$	$\frac{19.24}{16}$
	1.2	1.8	1.8	1.2

Minimum 2 3 3 2

integral ratio

Empirical formula = C<sub>2</sub>H<sub>3</sub>Cl<sub>3</sub>O<sub>2</sub>

Mass = 160

Mass = 160 × 10<sup>-1</sup>

23. The molarity of a 10% (mass/mass) aqueous solution of a monobasic acid (X) is \_\_\_\_\_

M (Nearest integer)

Given : Density of aqueous solution of (X) is

1.20 g mL<sup>-1</sup>

Molar mass of the acid is 100 g mol<sup>-1</sup>

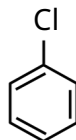
Ans. 120

Sol. Assuming 100 gm solution contain 10 gm solute.

Volume of 100 gm solution will be  $\frac{100}{1.20}$  ml.

Molarity =  $\frac{10/100}{100/1.20} \times 1000 = 12.0$  or  $120 \times 10^{-1}$

24. Consider the following sequence of reactions :



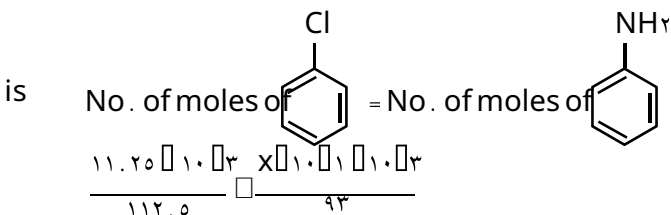
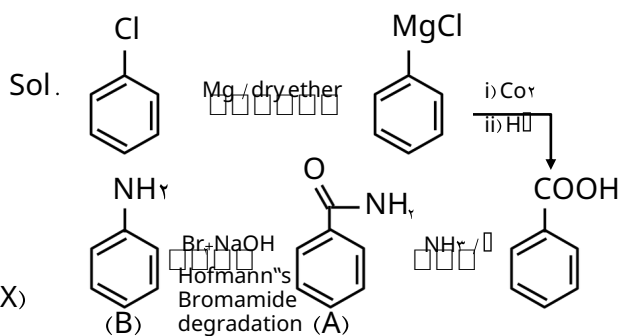
i) Mg, dry ether  
ii) CO<sub>2</sub>, HO  
iii) NH<sub>3</sub>

Chlorobenzene

11.20 mg of chlorobenzene will produce \_\_\_\_\_ mg of product B. in complete (Consider the reactions result conversion.)

Given molar mass of C, H, O, N and Cl as 12, 1, 16, 14 and 35.5 g mol<sup>-1</sup> respectively

Ans. 93



$x \times 10^{-1} = 93 \times 0.1$

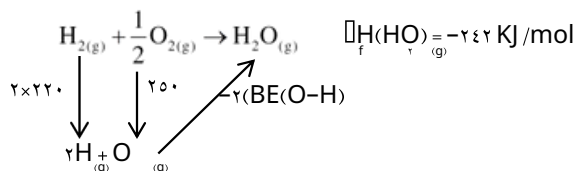
$x = 93$  mg

25. The formation enthalpies,  $\Delta_f H^\circ$  for H<sub>2</sub>O(g) and O<sub>2</sub>(g) are 241.8 and 0 kJ mol<sup>-1</sup> respectively, at 298.15 K, and  $\Delta_f H^\circ$  for H<sub>2</sub>O(l) is -285.8 kJ mol<sup>-1</sup> at the same temperature. The average bond enthalpy of the O-H bond in water at 298.15 K is \_\_\_\_\_ kJ mol<sup>-1</sup> (nearest integer).

Ans. 466

Sol.  $\frac{1}{2} H_{2(g)} \rightarrow H(g) \quad \Delta_f H^\circ(H) = 218 \text{ kJ/mol}$

$\frac{1}{2} O_{2(g)} \rightarrow O(g) \quad \Delta_f H^\circ(O) = 249 \text{ kJ/mol}$



$\Delta_f H^\circ(H_2O(l)) = -285.8 = 2(218) + 2(249) - 2(B.E.(O-H))$

$B.E.(O-H) = 466 \text{ kJ/mol}$