

5. Let

$$A = \{x \in (0, \pi) : \log_{(2/\pi)} |\sin x| + \log_{(2/\pi)} |\cos x| = 2\}$$

and

$$B = \{x \in (0, \sqrt{x}(\sqrt{x} + 4)) : \sqrt{x} + 2 = 6\}$$

$n(A \cap B)$ is equal to:

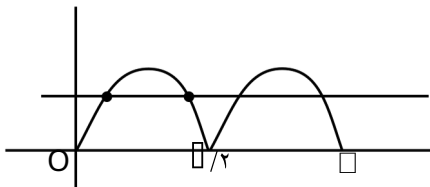
- (1) 8 (2) 2
(3) 1 (4) 6

Ans. (2)

Sol. $A : \log_{(2/\pi)} |\sin x| + \log_{(2/\pi)} |\cos x| = 2$

$$\log_{(2/\pi)} (|\sin x \cdot \cos x|) = 2$$

$$|\sin x \cos x| = \left(\frac{2}{\pi}\right)^2$$



Number of solution ξ

$B : \text{let } \sqrt{x} = t > 0$

then $t^2 - \xi t + \xi t - \xi t + \xi t = 4$

$$t^2 - \xi t + \xi t - \xi t + \xi t = 4$$

$$x = 0, x = 1$$

again let $\sqrt{x} = t$

$$t^2 - \xi t + \xi t - \xi t + \xi t = 4$$

$$t = 2, \xi$$

$$x = 4, 16$$

Total number of solutions

$$n(A \cap B) = \xi + \xi = 8$$

6.

Let the position vectors of three vertices of a

triangle be $4p + 3q + 2r$ and

$2p + 2q + 2r$. If the position vectors of the

orthocenter and the circumcenter of the triangle are

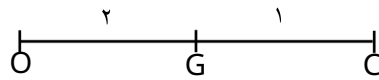
$\frac{p + q + r}{4}$ and $p + q + r$ respectively, then

$\frac{p + q + r}{4}$ is equal to:

- (1) 3 (2) 1
(3) 6 (4) 8

Ans. (1)

Sol. We know that



O (orthocentre) $\frac{p + q + r}{3}$

C (circum centre) $\frac{p + q + r}{3}$

C (centroid) = $\frac{p + q + r}{3}$

by relation

$$2 \cdot OG = GC \Rightarrow 2 \cdot \frac{p + q + r}{3} = \frac{p + q + r}{3}$$

$$8 \cdot \frac{p + q + r}{3} = \frac{p + q + r}{3}$$

$$8p = r, 8q = r, 8r = r$$

$$p = \frac{r}{8}, q = \frac{r}{8}, r = \frac{r}{8}$$

$$8p + 8q + 8r = r$$

$$8 \cdot \frac{r}{8} + 8 \cdot \frac{r}{8} + 8 \cdot \frac{r}{8} = r$$

7. Let $[x]$ denote the greatest integer function, and let m and n respectively be the numbers of the points, where the function $f(x) = [x] + |x - 2|$, $-2 < x < 2$, is not continuous and not differentiable.

Then $m + n$ is equal to:

- (1) 6 (2) 8

(3) 9

Ans. (2)

Sol. $f(x) = [x] + |x - 2|$

$$f(x) = [x], \quad -2 < x < 0$$

$$f(x) = [x] + 1, \quad 0 < x < 1$$

$$f(x) = [x] + 2, \quad 1 < x < 2$$

$$f(x) = [x] + 3, \quad 2 < x < 2$$

$$f(x) = [x]$$

$$f(x) = [x]$$

So $f(x)$ is not continuous at ξ points and not differentiable at ξ point

$$\text{So } m + n = \xi + \xi = 8$$

JEE-MAIN EXAMINATION – JANUARY 2025

(HELD ON FRIDAY 24th JANUARY 2025)

TIME : 3:00PM TO 6:00 PM

MATHEMATICS

SECTION-A

1. The equation of the chord of the ellipse

$$\frac{x^2}{25} + \frac{y^2}{16} = 1, \text{ whose mid-point is } (r, 1) \text{ is :}$$

- (1) $4x + 20y = 169$ (2) $4x + 12y = 134$
 (3) $20x + 10y = 176$ (4) $2x + 16y = 31$

Ans. (1)

Sol. Equation of chord with given middle point

$$T = S$$

$$\frac{rx}{25} + \frac{y}{16} = 1 \Rightarrow \frac{rx}{25} + \frac{y}{16} = 1$$

$$4rx + 20y = 164 + 20 \cdot 4rx$$

$$+ 20y = 169 \text{ Ans.}$$

2. The function $f : (-\infty, \infty) \rightarrow (-\infty, \infty)$ defined by

$$f(x) = \frac{2x - 2x}{2x - 2x}$$
 is :

- (1) One-one but not onto
 (2) Onto but not one-one
 (3) Both one-one and onto
 (4) Neither one-one nor onto

Ans. (1)

Sol. $f(x) = \frac{2^{2x} - 1}{2^{2x} - 1}$

$$= \frac{2^{2x} - 1}{2^{2x} - 1}$$

$$f(x) = \frac{2^{2x} - 1}{2^{2x} - 1} \cdot \frac{2^{2x} - 1}{2^{2x} - 1} \text{ i.e always +ve}$$

so $f(x)$ is $\frac{1}{2}$ function

$$f(-1) = -1$$

$$f(1) = 1$$

$f(x) \in (-1, 1)$ co-domain

so function is one-one but not onto

TEST PAPER WITH SOLUTION

3. If $\cot^{-1} \frac{1}{x} + \cot^{-1} \frac{1}{y} = \cot^{-1} \frac{1}{z}$ then the expression

$$\cot^{-1} \frac{1}{x} + \cot^{-1} \frac{1}{y} = \cot^{-1} \frac{1}{z}$$

$$\cot^{-1} \frac{1}{x} + \cot^{-1} \frac{1}{y} = \cot^{-1} \frac{1}{z}$$
 is equal to:

- (1) $\frac{1}{2} (\cot^{-1} \frac{1}{x} + \cot^{-1} \frac{1}{y})$ (2) $\frac{1}{2} \cot^{-1} \frac{1}{z}$
 (3) $\frac{1}{2} \cot^{-1} \frac{1}{z}$ (4) $\frac{1}{2} \cot^{-1} \frac{1}{z}$

Ans. (4)

Sol. $\cot^{-1} \frac{1}{x} + \cot^{-1} \frac{1}{y} = \cot^{-1} \frac{1}{z}$

$$\tan^{-1} \frac{1}{\cot^{-1} \frac{1}{x} + \cot^{-1} \frac{1}{y}} = \tan^{-1} \frac{1}{\cot^{-1} \frac{1}{z}}$$

4. Let $f : (0, \infty) \rightarrow \mathbb{R}$ be a function which is differentiable at all points of its domain and satisfies the condition $xf'(x) = 2xf(x) + 3$, with $f(1) = 1$. Then $2f(2)$ is equal to:

- (1) 29 (2) 19
 (3) 39 (4) 23

Ans. (3)

Sol. $xf'(x) - 2xf(x) = 3$

$$\frac{d}{dx} \left(\frac{f(x)}{x^2} \right) = \frac{3}{x^3}$$

$$\frac{d}{dx} \left(\frac{f(x)}{x^2} \right) = \frac{3}{x^3}$$

Integrating both sides

$$\frac{f(x)}{x^2} = \frac{1}{x^2} + C$$

$$f(x) = \frac{1}{x} + Cx^2$$

put $x = 1$
 $1 = -1 + C \Rightarrow C = 2$

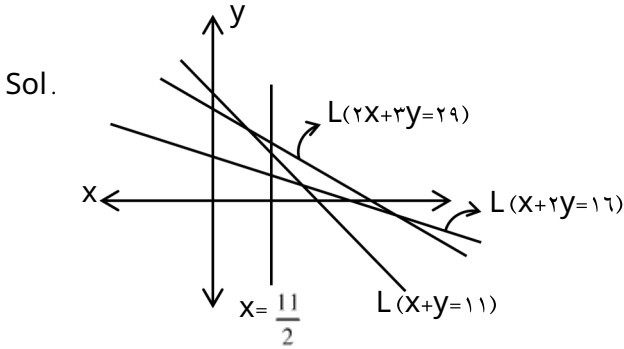
$$f(x) = \frac{1}{x} + 2x^2$$

Now $2 \times f(2) = 2 \left(\frac{1}{2} + 2 \times 2^2 \right) = 2 \left(\frac{1}{2} + 8 \right) = 2 \times 8.5 = 17$

8. Let the points $(1, 2), (2, 3)$ lie on or inside the triangle with sides $x + y = 11$, $x + 2y = 16$ and $2x + 3y = 29$. Then the product of the smallest and the largest values of λ is equal to :

- (1) 22
- (2) 44
- (3) 33
- (4) 55

Ans. (3)



Sol.

Point of intersection of $x = \frac{11}{2}$ with L & L gives,

$$\lambda_{\min} = \frac{11}{2}$$

$$\text{and } \lambda_{\max} = 6$$

$$\lambda_{\min} \cdot \lambda_{\max} = \frac{11}{2} \cdot 6 = 33$$

9. In an arithmetic progression, if $S_5 = 103$ and $S_{10} = 57$, then $S_5 - S_{10}$ is equal to :

- (1) 510
- (2) 515
- (3) 520
- (4) 505

Ans. (2)

Sol. Let a & d are first term and common diff of an AP

$$S_5 = \frac{5}{2} [2a + 4d] = 103 \quad \dots (1)$$

$$S_{10} = \frac{10}{2} [2a + 9d] = 57 \quad \dots (2)$$

by (1) & (2)

$$a = \frac{57}{5} \quad d = \frac{2}{5}$$

$$S_5 - S_{10} = \frac{5}{2} [2a + 4d] - \frac{10}{2} [2a + 9d]$$

$$= 5a - 10d$$

$$= 515$$

10. If $V = 5 + \frac{1}{7}(5 + \lambda) + \frac{1}{7^2}(5 + 2\lambda) + \frac{1}{7^3}(5 + 3\lambda) + \dots$, then the value of λ is :

- (1) 1
- (2) $\frac{6}{7}$
- (3) 6
- (4) $\frac{1}{7}$

Ans. (3)

Sol. Let $S = 5 + \frac{1}{7}(5 + \lambda) + \frac{1}{7^2}(5 + 2\lambda) + \dots$

$$\frac{1}{7}S = \frac{1}{7} [5 + \lambda] + \frac{1}{7^2} [5 + 2\lambda] + \dots$$

$$\frac{1}{7}S = 5 + \lambda + \frac{1}{7} [5 + 2\lambda] + \frac{1}{7^2} [5 + 3\lambda] + \dots$$

$$6 = 5 + \lambda \Rightarrow \lambda = 1$$

11. If the system of equations $x + 2y - 3z = 2$, $2x + \lambda y + 5z = 0$, $\lambda x + 3y + \mu z = 0$ has infinitely many solutions, then

- (1) $\lambda + \mu$ is equal to 13
- (2) $\lambda + \mu$ is equal to 11
- (3) $\lambda + \mu$ is equal to 10
- (4) $\lambda + \mu$ is equal to 12

Ans.

$$\text{Sol. } D = \begin{vmatrix} 1 & 2 & -3 \\ 2 & \lambda & 5 \\ \lambda & 3 & \mu \end{vmatrix} = 0 \Rightarrow \lambda\mu + 5\lambda - 3\mu + 10 = 0$$

$$D_1 = 2\mu + 99 - 10\mu + 200$$

$$D_2 = 12 - \mu$$

$$D_3 = 0 + 0$$

$$D_4 = 0 \Rightarrow \mu = 13 \text{ \& } D_2 = 0 \Rightarrow \lambda = -1$$

check & verify for these values D & $D_1 = 0$

12. Let (b, c) be the largest open interval in which the function $f(x) = 2 \log_e(x - 2) - x^2 + ax + 1$ is strictly increasing and (a, b) be the largest open interval in which the function $g(x) = (x - 1)(x + 2 - a)$ is strictly decreasing. Then $(a + b - c)$ is equal to :

- (1) 280
- (2) 360
- (3) 420
- (4) 160

Ans. (2)

Sol. $f(x) = \frac{y}{x+y} - yx + a = 0$

$$f(x) = \frac{x^2}{x^2} - 2x^2 = 0$$

$$f(x) = x$$

$$f(y) = y$$

$$y - y + a = 0$$

$$a = 0$$

$$a_{\min} = 0$$

$$g(x) = (x-1)(x+y-a)$$

$$g(x) = (x-1)(x-2)$$

$$g'(x) = (x-1)(x-2) + (x-2)(x+1)$$

$$= (x-1)(x-2)(2x-2+x+1)$$

$$= (x-1)(x-2)(3x-1) > 0$$

$$x = \frac{1}{3}, 2$$

$$1 \cdot (a+b-c) = 1 \cdot (4 - \frac{1}{3} - 2) = \frac{2}{3}$$

13. Suppose A and B are the coefficients of x^m and x^r terms respectively in the binomial expansion of $(1+x)^{m-1}$. If $rA = 6B$, then n is equal to:

(1) 22

(2) 21

(3) 20

(4) 19

Ans. (2)

Sol. $A = {}^{m-1}C_r$, $B = {}^{m-1}C_m$

$$r {}^{m-1}C_r = 6 {}^{m-1}C_m$$

$$\frac{r!}{(r-1)!1!} = 6 \frac{(m-1)!}{(m-1-r)!r!}$$

$$\frac{r}{1} = 6 \frac{(m-1)!}{(m-1-r)!r!}$$

$$\frac{1}{3} \cdot \frac{1}{2} \dots \frac{1}{r} = 6 \frac{(m-1)!}{(m-1-r)!r!}$$

$$rn - 12 = 30$$

$$n = 21$$

14. Let $a = 3i + 4j + 2k$, $b = a + (i + 2k)$ and $e = b + k$.

Then the projection of $2j$ on a is:

(1) $3\sqrt{7}$

(2) $\sqrt{14}$

(3) $2\sqrt{14}$

(4) $2\sqrt{7}$

Ans. (3)

Sol. $b = a + (i + 2k)$

$$\begin{vmatrix} i & j & k \\ r & 1 & 2 \\ 1 & 0 & 2 \end{vmatrix} = r(i \wedge j) + k(i \wedge j) + j(k \wedge i)$$

$$e = b + k = a + (i + 2k) + k$$

$$e = a + (i + 3k)$$

Projection of $2j$ on a

$$\frac{(2j \cdot a) \langle a, a \rangle}{\langle a, a \rangle \langle a, a \rangle} = \frac{2 \cdot 0}{\sqrt{14} \cdot \sqrt{14}}$$

$$= \frac{2 \cdot 0}{14} = 0$$

15. For some a, b, let

$$f(x) = \begin{vmatrix} a - \frac{\sin x}{x} & 1 & b \\ a & 1 - \frac{x}{x} & b \\ a & 1 & b - \frac{\sin x}{x} \end{vmatrix}, x \in \mathbb{R}$$

$\lim_{x \rightarrow 0} f(x) = a - b$. Then (a, b) is equal to:

to:

(1) 20

(2) 9

(3) 36

(4) 16

Ans. (3)

Sol. $\lim_{x \rightarrow 0} f(x) = \begin{vmatrix} a - 1 & 1 & b \\ a & 0 & b \\ a & 1 & b - 1 \end{vmatrix}$

$$= (a-1)(b-1) - b + 1(ab - a(b-1)) + ba$$

$$= (a-1)(b-1) - a + ab$$

$$= b + a - 1 = 0 \Rightarrow a + b = 1$$

$$a = 2, b = -1 \Rightarrow (a, b) = (2, -1)$$

16. Group A consists of r boys and s girls, while group B consists of t boys and u girls. The number of ways, x boys and y girls can be invited for a picnic if u of them must be from group A and the remaining r from group B, is equal to:

(1) ${}^r C_0 {}^s C_0$

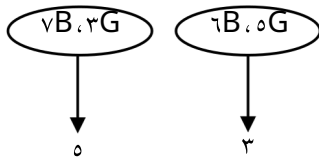
(2) ${}^r C_1 {}^s C_0$

(3) ${}^r C_2 {}^s C_0$

(4) ${}^r C_0 {}^s C_0$

Ans. (3)

Sol.



- C-I (rG & rB) & (oG & rB) (rG &
- C-II rB) & (rG & oB) (oG & rB) &
- C-III (rG & oB)

Total = C-I + C-II + C-III
 = $C_1^r C_1^o C_1^r + C_1^r C_1^o C_1^o + C_1^r C_1^o C_1^r$
 = 1920

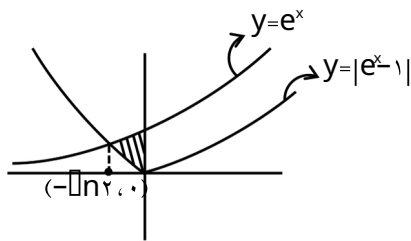
17. The area of the region enclosed by the curves

$y = e^x$, $y = |e^{-x} - 1|$ and y-axis is:

- (1) $1 + \log_e 2$
- (2) $\log_e 2$
- (3) $2 \log_e 2 - 1$
- (4) $1 - \log_e 2$

Ans. (4)

Sol.



For Area $\int_{-\ln 2}^0 e^x dx + \int_0^{\ln 2} (e^{-x} - 1) dx$

$= [e^x]_{-\ln 2}^0 + [-e^{-x} - x]_0^{\ln 2}$

$= (1 - (1 + \ln 2))$

$= 1 - \ln 2$

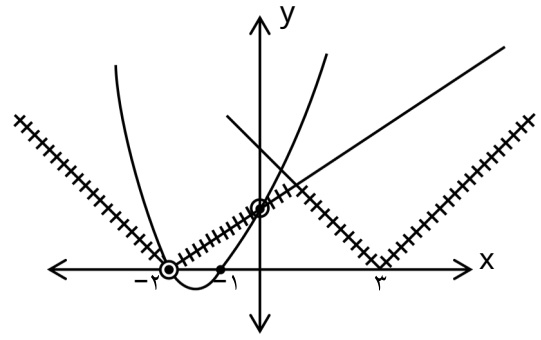
18. The number of real solution(s) of the equation

$x^2 + 2x + 2 = \min(|x - 2|, |x + 2|)$ is :

- (1) 2
- (2) 0
- (3) 3
- (4) 1

Ans. (1)

Sol.



Only 2 solutions.

19. Let $A = [a_{ij}]$ be a square matrix of order n with entries either 0 or 1. Let E be the event that A is an invertible matrix. Then the probability P(E) is :

- (1) $\frac{5}{8}$
- (2) $\frac{3}{16}$
- (3) $\frac{1}{8}$
- (4) $\frac{3}{8}$

Ans. (4)

Sol. C-I $\begin{vmatrix} 1 & \\ & 1 \end{vmatrix}$ 4 ways

C-II $\begin{vmatrix} 1 & \\ & 1 \end{vmatrix}$ & $\begin{vmatrix} 1 & \\ & 1 \end{vmatrix}$ 2 ways

$P = \frac{\text{favourable}}{\text{total}} = \frac{6}{8}$

20. If the equation of the parabola with vertex

$(\frac{3}{2}, 3)$ and the directrix $x + 2y = 0$ is

$ax^2 + by - cxy - 2x - 6y + 20 = 0$, then

$\frac{a}{b}$ is equal to:

- (1) 6
- (2) 8
- (3) 7
- (4) 9

Ans. (4)

Sol. Equation of axis $y - 2 = \sqrt{2}x - \frac{3}{2}$

$y - 2x = \frac{3}{2}$

foot of directrix

$y - 2x = 0$

& $(0, 0)$

$2y + x = 0$

Focus = (3, 1)

PS = PM

$$(x-3)^2 + (y-1)^2 = \sqrt{(x-3)^2 + (y-1)^2}$$

$$x^2 + y^2 - 6x - 2y + 10 = \sqrt{(x-3)^2 + (y-1)^2}$$

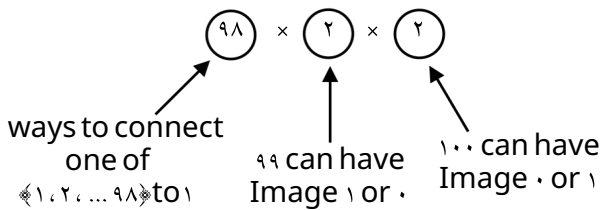
$$x^2 + y^2 - 6x - 2y + 10 = \sqrt{x^2 + y^2 - 6x - 2y + 10}$$

SECTION-B

21. Number of functions $f : \{1, 2, \dots, 10\} \rightarrow \{1, 2\}$ that assign 1 to exactly one of the positive integers less than or equal to 10, is equal to _____

Ans. (99)

Sol.

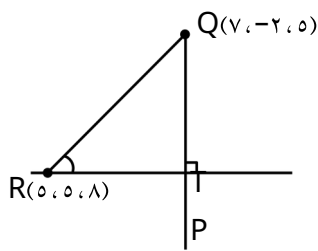


21 Ans.

22. Let P be the image of the point Q(1, -2, 0) in the line L: $\frac{x-1}{2} = \frac{y-1}{3} = \frac{z}{4}$ and R(0, p, q) be a point on L. Then the square of the area of ΔPQR is _____

Ans. (90)

Sol.



Let $R(1 + \lambda, -2 + \lambda, \lambda)$

$$\lambda + 1 = 0$$

$$\lambda = -1$$

$R(0, -1, -1)$

Let $T(1 + \lambda, -2 + \lambda, \lambda)$

$$QT = \sqrt{(1-\lambda)^2 + (-2+\lambda-2)^2 + \lambda^2}$$

$$b = \sqrt{1 + \lambda^2}$$

$$QT \cdot b$$

$$\lambda^2 - 12 + 9\lambda + 3 + 16\lambda - 20 = 0$$

$$\lambda = 1$$

$T(2, -1, 1)$

$$QT = \sqrt{3} \quad RT = \sqrt{19}$$

$$\text{Area of } \Delta PQR = \frac{1}{2} \sqrt{3 \cdot 19 - 3 \cdot 19}$$

$$= 90$$

23. Let $y = y(x)$ be the solution of the differential equation $y \cos x \frac{dy}{dx} = \sin x - y \sin x$, $x \in (0, \frac{\pi}{2})$.

If $y(\frac{\pi}{3}) = 2$, then $y(\frac{\pi}{4})$ is equal to _____

Ans. (1)

Sol. $\frac{dy}{dx} = \frac{\sin x - y \sin x}{y \cos x}$

$$I.F. = e^{\int \tan x dx} = \sec x$$

$$y \sec x = \int \frac{\sin x}{\cos^2 x} dx$$

$$\int \tan x \sec x dx$$

$$= \sec x + C$$

$$C = -2$$

$$y = \cos x - 2 \cos x$$

$$y = \frac{1}{\sqrt{2}}$$

$$y = -\sin x + 2 \cos x$$

$$y = \frac{1}{4} \sqrt{2}$$

$$y = \frac{1}{4} \sqrt{2}$$

24. Let $H_1: \frac{x^2}{a^2} - \frac{y^2}{b^2} = 1$ and $H_2: \frac{x^2}{A^2} - \frac{y^2}{B^2} = 1$ be two

hyperbolas having length of latus rectum $5a$ and $10\sqrt{5}$ respectively. Let their eccentricities be

$e_1 = \frac{\sqrt{5}}{2}$ and e_2 respectively. If the product of the

lengths of their transverse axes is $10\sqrt{10}$, then

$25e_2$ is equal to _____

Ans. (00)

Sol. $\frac{bx}{a} = \frac{1}{\sqrt{a}}$

$$1 = \frac{bx}{a} = \frac{1}{\sqrt{a}}$$

$$a = \sqrt{bx}$$

$$\frac{bx}{a} = \frac{1}{\sqrt{a}}$$

$$bx = \frac{1}{\sqrt{a}}$$

$$bx = \frac{1}{\sqrt{a}}$$

$$B = \frac{1}{\sqrt{a}}$$

$$A = \frac{1}{\sqrt{a}}$$

$$e^x = \frac{Ax}{B}$$

$$1 = \frac{1}{\sqrt{a}}$$

$$e^x = \frac{1}{\sqrt{a}}$$

$$x = e^x$$

20. If $\int \frac{2x^2 + 5x + 9}{\sqrt{x^2 + 1}} dx = \sqrt{x^2 + 1} + \log_e \left| x + \frac{1}{2} \sqrt{x^2 + 1} \right| + C$, where C is the

constant of integration, then $\frac{A}{B}$ is equal to _____

constant of integration, then $\frac{A}{B}$ is equal to _____

Ans. (16)

Sol. $2x^2 + 5x + 9 = A(x + \sqrt{x^2 + 1}) + B(x + 1) + C$

$$A = 2 \quad B = \frac{3}{2} \quad C = \frac{11}{2}$$

$$\int \sqrt{x^2 + 1} dx = \frac{1}{2} \int \frac{2x^2 + 1}{\sqrt{x^2 + 1}} dx = \frac{1}{2} \int \frac{dx}{\sqrt{x^2 + 1}}$$

$$\int \frac{2x^2 + 1}{\sqrt{x^2 + 1}} dx = \int \frac{2x^2 + 1}{\sqrt{x^2 + 1}} dx = \int \frac{2x^2 + 1}{\sqrt{x^2 + 1}} dx$$

$$\int \frac{2x^2 + 1}{\sqrt{x^2 + 1}} dx = \int \frac{2x^2 + 1}{\sqrt{x^2 + 1}} dx = \int \frac{2x^2 + 1}{\sqrt{x^2 + 1}} dx$$

$$\int \frac{1}{\sqrt{x^2 + 1}} dx = \log_e \left| x + \sqrt{x^2 + 1} \right| + C$$

$$\frac{A}{B} = \frac{2}{\frac{3}{2}} = \frac{4}{3}$$

$$\frac{A}{B} = \frac{4}{3}$$

JEE-MAIN EXAMINATION – JANUARY 2025

(HELD ON FRIDAY 24th JANUARY 2025)

TIME : 3 : 00PM TO 6 : 00 PM

PHYSICS

TEST PAPER WITH SOLUTION

SECTION-A

28.

26. Young's double slit interference apparatus is immersed in a liquid of refractive index 1.44 . It has slit separation of 1.0 mm. The slits are illuminated by a parallel beam of light whose wavelength in air is 690 nm. The fringe-width on a screen placed behind the plane of slits at a distance of 0.72 m, will be :

- (1) 0.22 mm (2) 0.23 mm
(3) 0.13 mm (4) 0.16 mm

Ans. (1)

Sol.
$$\frac{D}{d} \lambda = \frac{0.72}{1.0} \times \frac{690 \times 10^{-9}}{1.44} = 0.22 \text{ mm}$$

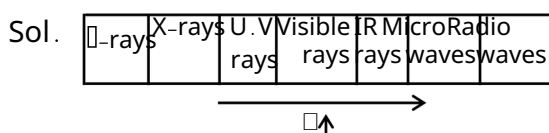
27. Arrange the following in the ascending order of wavelength (λ) :

- (A) Microwaves (λ)
(B) Ultraviolet rays (λ)
(C) Infrared rays (λ)
(D) X-rays (λ)

Choose the most appropriate answer from the options given below :-

- (1) $4 \rightarrow 3 \rightarrow 2 \rightarrow 1$
(2) $4 \rightarrow 2 \rightarrow 3 \rightarrow 1$
(3) $1 \rightarrow 2 \rightarrow 3 \rightarrow 4$
(4) $1 \rightarrow 3 \rightarrow 2 \rightarrow 4$

Ans. (3)



Given below are two statements. One is labelled as Assertion (A) and the other is labelled as Reason (R).

Assertion (A) : A electron in a certain region of uniform magnetic field is moving with constant velocity in a straight line path.

Reason (A) : The magnetic field in that region is along the direction of velocity of the electron.

In the light of the above statements, choose the correct answer from the options given below :

- (1) (A) is false but (R) is true
(2) Both (A) and (R) are true and (R) is the correct explanation of (A)
(3) Both (A) and (R) are true but (R) is NOT the correct explanation of (A)
(4) (A) is true but (R) is false

Ans. (2)

Sol.
$$\vec{F} = q \vec{v} \times \vec{B}$$

$$\vec{F} \perp \vec{v}$$

$$\vec{v} \parallel \vec{B}$$

$\lambda = \frac{h}{mv}$

29. A solid sphere is rolling without slipping on a horizontal plane. The ratio of the linear kinetic energy of the centre of mass of the sphere and rotational kinetic energy is :

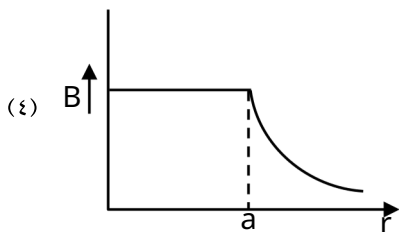
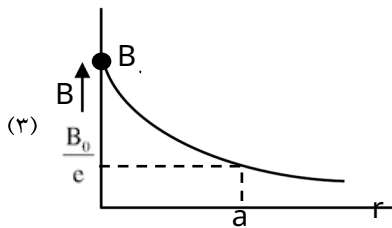
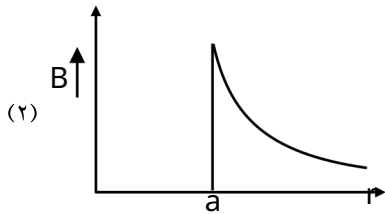
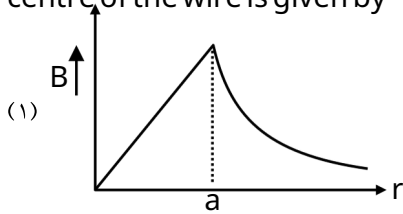
- (1) $\frac{2}{5}$ (2) $\frac{5}{7}$
(3) $\frac{3}{2}$ (4) $\frac{2}{3}$

Ans. (2)

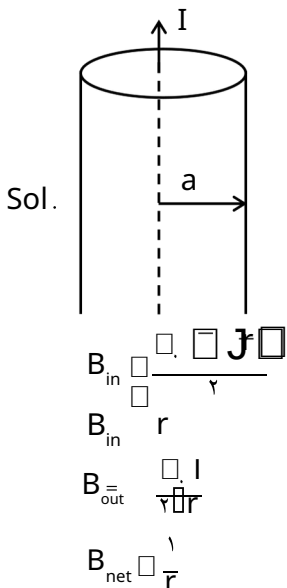
Sol.
$$\frac{\text{Linear KE}}{\text{Rotational K.E}} = \frac{\frac{1}{2} m v_{cm}^2}{\frac{1}{2} I \omega^2}$$

$$\frac{m v_{cm}^2}{\frac{5}{2} m R^2 \omega^2} = \frac{5}{7} \quad (V = R\omega)$$

30. A long straight wire of a circular cross-section with radius 'a' carries a steady current I. The current I is uniformly distributed across this cross-section. The plot of magnitude of magnetic field B with distance r from the centre of the wire is given by



Ans. (1)



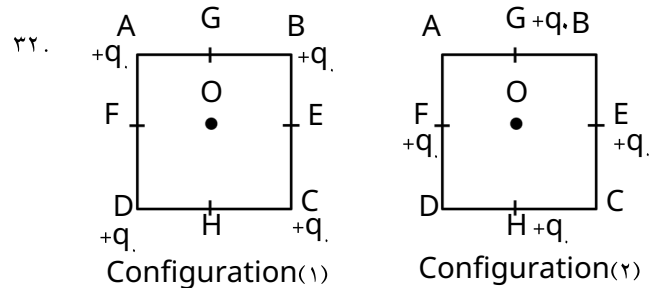
31. Given below are two statements. One is labelled as Assertion (A) and the other is labelled as Reason (R). Assertion (A) : In an insulated container, a gas is adiabatically shrunk to half of its initial volume. The temperature of the gas decreases. Reason (R) : Free expansion of an ideal gas is an irreversible and an adiabatic process. In the light of the above statement, choose the correct answer from the options given below : (1) Both (A) and (R) are true and (R) is the correct explanation of (A) (2) (A) is true but (R) is false (3) (A) is false but (R) is true (4) Both (A) and (R) are true but (R) is NOT the correct explanation of (A)

Ans. (3)

Sol. (A) $TV^\gamma = \text{const}$, $TV^\gamma = \text{const}$

Temp increases

(R) Free expansion is assumed fast, so Adiabatic



In the first configuration (1) as shown in the figure, four identical charges (q) are kept at the corners A, B, C and D of square of side length 'a'. In the second configuration (2), the same charges are shifted to mid points G, E, H and F, of the square. If

$K \frac{1}{4\pi\epsilon_0}$, the difference between the potential

energies of configuration (2) and (1) is given by :

(1) $\frac{Kq^2}{a} \sqrt{2}$ (2) $\frac{Kq^2}{a} \sqrt{3}$

(3) $\frac{Kq^2}{a} \sqrt{2}$ (4) $\frac{Kq^2}{a} \sqrt{3}$

Ans. (4)

Sol. $U_x = \frac{\epsilon Kq\gamma}{a} = \frac{\gamma Kq\gamma}{\sqrt{\gamma a}} = \frac{Kq\gamma}{a} = 4\gamma$

$U_y = \frac{Kq\gamma}{a} = \frac{\epsilon Kq\gamma}{\sqrt{\gamma a}} = \frac{Kq\gamma}{a} = \sqrt{\gamma}$

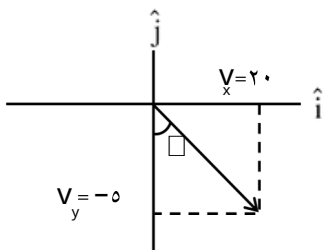
$U = \sqrt{U_x^2 + U_y^2} = \sqrt{16\gamma + \gamma} = \sqrt{17\gamma}$

33. The position vector of a moving body at any instant of time is given as $r = \epsilon t^2 \hat{i} + \gamma t^2 \hat{j}$ m. The magnitude and direction of velocity at $t = \gamma$ s is.

- (1) $\sqrt{2}\gamma$ m/s, making an angle of $\tan^{-1} \epsilon$ with -ve Y axis
- (2) $\sqrt{2}\gamma$ m/s, making an angle of $\tan^{-1} \epsilon$ with +ve X axis
- (3) $\sqrt{2}\gamma$ m/s, making an angle of $\tan^{-1} \epsilon$ with -ve Y axis
- (4) $\sqrt{2}\gamma$ m/s, making an angle of $\tan^{-1} \epsilon$ with +ve X axis

Ans. (3)

Sol. $r = \epsilon t^2 \hat{i} + \gamma t^2 \hat{j}$
 $v = 2\epsilon t \hat{i} + 2\gamma t \hat{j}$
 at $t = \gamma$ sec



$\tan \theta = \frac{V_x}{V_y} = \frac{2\epsilon\gamma}{2\gamma^2} = \frac{\epsilon}{\gamma}$

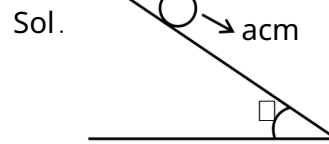
$\theta = \tan^{-1} \frac{\epsilon}{\gamma}$

From -ve Y-axis

34. A solid sphere and a hollow sphere of the same mass and of same radius are rolled on an inclined plane. Let the time taken to reach the bottom by the solid sphere and the hollow sphere be t_1 and t_2 respectively.

- (1) $t_1 > t_2$
- (2) $t_1 = t_2$
- (3) $t_1 = \gamma t_2$
- (4) $t_1 < t_2$

Ans. (4)



$t = \sqrt{\frac{2\ell}{a_{cm}}}$

$a_{cm} = \frac{g \sin \theta}{1 + \frac{I_{cm}}{MR^2}}$

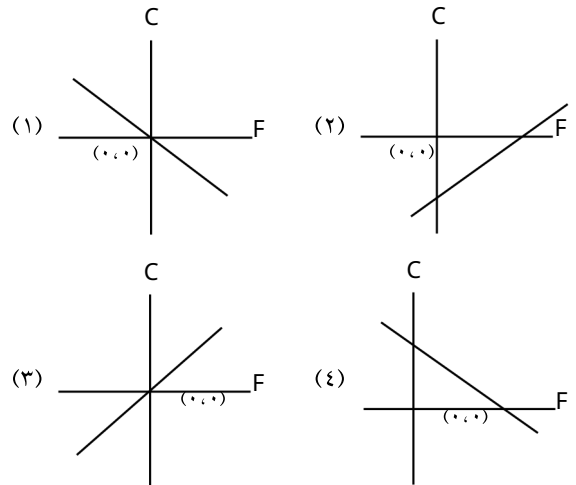
$a_1 = a_{cm} = \frac{g \sin \theta}{2}$ Solid

$a_2 = a_{cm} = \frac{2g \sin \theta}{3}$ Hollow

$a_1 < a_2$

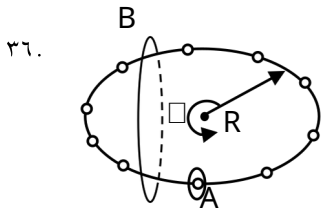
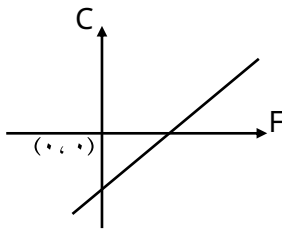
$t_1 > t_2$

35. Which of the following figure represents the relation between Celsius and Fahrenheit temperatures?



Ans. (2)

Sol. $C = \frac{F}{q}$ $C = \frac{eF}{q}$ $\frac{1}{q}$

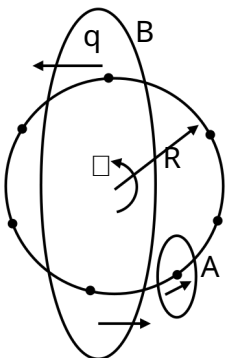


N equally spaced charges each of value q , are placed on a circle of radius R . The circle rotates about its axis with an angular velocity ω as shown in the figure. A bigger Amperian loop B encloses the whole circle where as a smaller Amperian loop A encloses a small segment. The difference between enclosed currents, $I_A - I_B$, for the given

Amperian loops is

- (1) $\frac{N\omega}{\gamma} q$ (2) $\frac{\gamma}{N} q$
 (3) $\frac{N}{\gamma} q$ (4) $\frac{N}{\omega} q$

Ans. (3)
Sol.



$$I_A = \frac{Nq}{\gamma}$$

$$I_B = \frac{Nq}{\gamma} \cdot I_B = \dots$$

$$I_A - I_B = \frac{Nq}{\gamma}$$

37. In photoelectric effect, the stopping potential (V_s) vs frequency (ν) curve is plotted.

(h is the Planck's constant and ϕ is work function of metal)

(A) V_s vs ν is linear

(B) The slope of V_s vs ν curve = $\frac{h}{e}$

(C) h constant is related to the slope of V_s vs ν line

(D) The value of electric charge of electron is not required to determine h using the V_s vs ν curve.

(E) The work function can be estimated without knowing the value of h .

Choose the correct answer from the options given below :

(1) (A), (B) and (C) only

(2) (C) and (D) only

(3) (A), (C) and (E) only

(4) (D) and (E) only

Ans. (3)

Sol. $h\nu = \phi + KE_{max}$

$$KE_{max} = eV_s$$

$$V_s = \frac{h\nu}{e} - \frac{\phi}{e}$$

(A) V_s vs ν is linear correct

(B) Slope

$$V_s = \frac{h}{e} \nu - \frac{\phi}{e} \text{ Wrong}$$

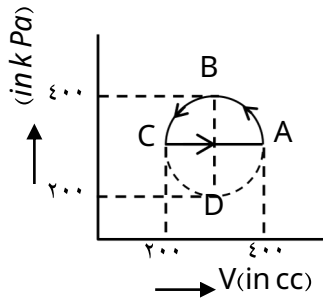
$$\text{Slope} = \frac{h}{e}$$

(C) Correct

(D) Incorrect

(E) Correct

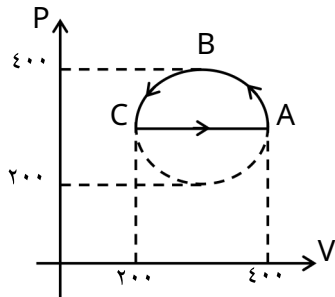
38. The magnitude of heat exchanged by a system for the given cyclic process ABCA (as shown in figure) is (in SI unit)



- (1) 10^5 (2) 0
 (3) zero (4) 4×10^5

Ans. (2)

Sol.



$W = \frac{1}{2} R^2$

$\frac{1}{2} R^2 = \frac{1}{2} \times 200 \times 200 = 20000 \text{ J}$

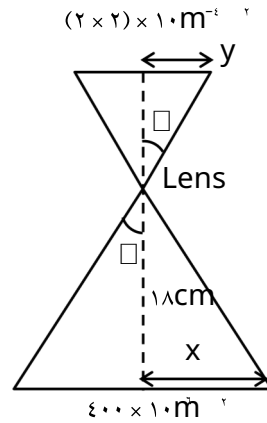
$\frac{1}{2} R^2 = 5000 \text{ J}$

39. A photograph of a landscape is captured by a drone camera at a height of 1.8 km. The size of the camera film is 2 cm × 2 cm and the area of the landscape photographed is $4 \times 10^6 \text{ km}^2$. The focal length of the lens in the drone camera is :

- (1) 1.8 cm (2) 2.8 cm
 (3) 2.0 cm (4) 0.9 cm

Ans. (1)

Sol.



$H = 1.8 \text{ km}$

Size of camera film = $2 \text{ cm} \times 2 \text{ cm}$

$A_{\text{image}} = 4 \times 10^{-4} \text{ km}^2$

$x = 2 \times 10^{-2} \text{ m} = 2 \times 10^{-5} \text{ km}$

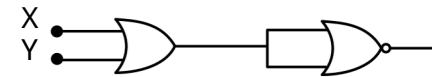
$y = 2 \times 10^3 \text{ m}$

$\frac{x}{y} = \frac{1.8 \text{ km}}{f}$

$f = 1.8 \times 10^{-2} \text{ m} = 1.8 \text{ mm}$

$f = 1.8 \text{ cm}$

40. The output of the circuit is low (zero) for :



(A) $X = 0, Y = 0$ (B) $X = 0, Y = 1$

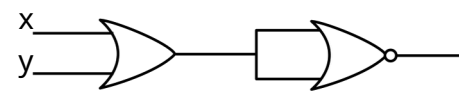
(C) $X = 1, Y = 0$ (D) $X = 1, Y = 1$

Choose the correct answer from the options given below : (1) (A), (C) and (D) only (2) (A), (B) and (C) only (3) (B), (C) and (D) only (4) (A), (B) and (D) only

(3)

Ans.

Sol.



X	Y	Output
0	0	1
0	1	1
1	0	1
1	1	0

ε1. The temperature of a body in air falls from $\xi^\circ\text{C}$ to $\gamma^\circ\text{C}$ in ξ minutes. The temperature of the air is $\gamma^\circ\text{C}$. The temperature of the body in the next ξ minutes will be :

- (1) $\frac{\gamma\xi}{\gamma}$ °C (2) $\frac{\gamma\lambda}{\xi}$ °C
 (3) $\frac{\gamma\xi}{\gamma}$ °C (4) $\frac{\gamma\lambda}{\xi}$ °C

Ans. (3)

Sol. $\frac{T_1 - T}{t} = K(T - T_{\text{avg}})$

$T_1 = \xi^\circ\text{C}; T = \gamma^\circ\text{C}; t = \xi; T = \gamma^\circ\text{C}$

$\frac{\xi - \gamma}{\xi} = K \frac{\xi + \gamma}{2}$

$K = \frac{\xi - \gamma}{\xi} \cdot \frac{2}{\xi + \gamma}$

Now $\frac{\gamma - T}{\xi} = K \frac{\gamma + T}{2}$

$\gamma - T = \frac{T - \gamma}{\gamma} \cdot \frac{\xi + \gamma}{2}$

$\frac{\gamma - T}{\gamma} = \frac{\xi + \gamma}{2}$

$T = \frac{\gamma\xi}{\gamma}$ °C

ε2. The energy E and momentum p of a moving body of mass m are related by some equation. Given that c represents the speed of light, identify the correct equation.

- (1) $E = pc + mc^2$ (2) $E = pc + mc$
 (3) $E = pc + mc^2$ (4) $E = pc + mc$

Ans. (4)

Sol. $E = ML^2T^{-2}$

$pc = MLT^{-1} \cdot LT^{-1} = ML^2T^{-2}$

$mc^2 = ML^2T^{-2}$

$E = ML^2T^{-2}$

$E = pc + mc^2$

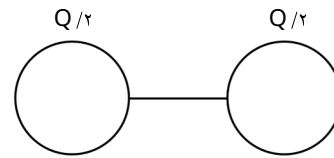
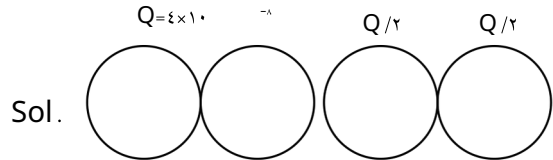
ε3. A small uncharged conducting sphere is placed in contact with an identical sphere but having $\xi \times 10^{-8}$ C charge and then removed to a distance such that the force of repulsion between them is 9×10^{-4} N.

The distance between them is (Take $\frac{1}{4\pi\epsilon_0}$ as

9×10^9 in SI units)

- (1) 2 cm (2) 3 cm
 (3) 4 cm (4) 1 cm

Ans. (1)



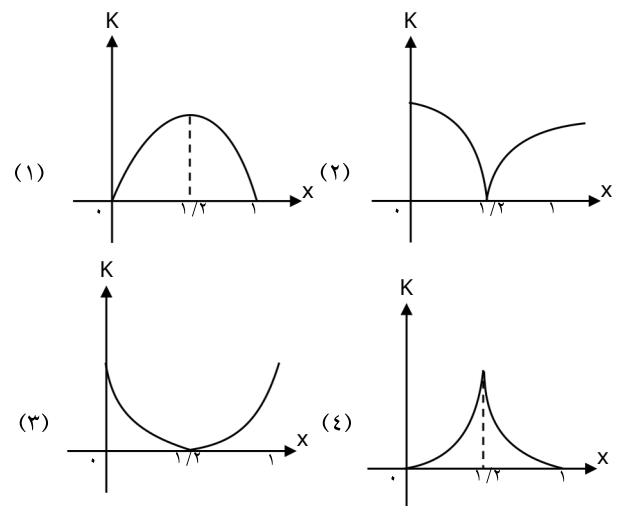
$F = \frac{k(Q/2)^2}{r^2} = 2 \times 10^{-4}$

$9 \times 10^9 \cdot \frac{(\xi \times 10^{-8})^2}{4r^2} = 2 \times 10^{-4}$

$r^2 = \frac{9 \times 10^9 \cdot \xi^2 \cdot 10^{-16}}{8 \times 10^{-4}}$

$r = 2 \times 10^{-2} \text{ m} = 2 \text{ cm}$

ε4. A particle oscillates along the x-axis according to the law, $x(t) = x \sin \frac{2\pi t}{2}$ where $x = 1$ m. The kinetic energy (K) of the particle as a function of x is correctly represented by the graph.



Ans. (1)

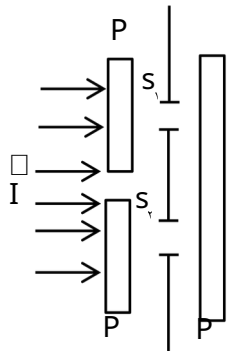
Sol. $x = x_0 \sin \omega t$ $y = x_0 \cos \omega t$

$x = \frac{x_0}{\sqrt{2}}$ $y = \frac{x_0}{\sqrt{2}}$

where $x_0 = \lambda$

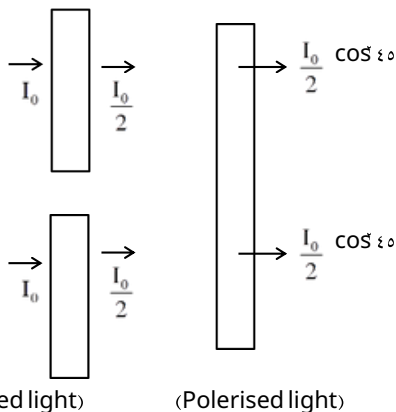
$x = \frac{\lambda}{\sqrt{2}}$ $y = \frac{\lambda}{\sqrt{2}}$

Particle is oscillating between $x = 0$ to $x = \lambda$ In a Young's double slit experiment, three polarizers are kept as shown in the figure. The transmission axes of P and P are orthogonal to each other. The polarizer P covers both the slits with its transmission axis at ϵ_0° to those of P and P. An unpolarized light of wavelength λ and intensity I is incident on P and P. The intensity at a point after P where the path difference between the light waves from S and S is



- (1) $\frac{I}{\sqrt{2}}$
- (2) $\frac{I}{\epsilon}$
- (3) I
- (4) $\frac{I}{\sqrt{2}}$

Ans. (3)



Sol.

after passing through third polariser, Intensity of both the waves must be $\frac{I}{\epsilon}$

now, at a point where path diff is $\frac{\lambda}{\sqrt{2}}$, phase difference

$$I_{res} = \sqrt{\left(\frac{I}{\epsilon}\right)^2 + \left(\frac{I}{\epsilon}\right)^2 + 2\left(\frac{I}{\epsilon}\right)\left(\frac{I}{\epsilon}\right)\cos\frac{2\pi}{\sqrt{2}}}$$

$$= \frac{I}{\epsilon}$$

Q6. A tightly wound long solenoid carries a current of 1.0 A. An electron is executing uniform circular motion inside the solenoid with a time period of 10 ns. The number of turns per metre in the solenoid is _____ kg, charge of electron $|q| = 1.6 \times 10^{-19}$ C.

$$n = \frac{I}{\epsilon_0 \mu_0} \times 10^{-7} \text{ N/A}^2, 10 \text{ ns} = 10^{-8} \text{ s}$$

Ans. (200)

Sol. Since time period of a revolving charge is $\frac{2\pi m}{qB}$

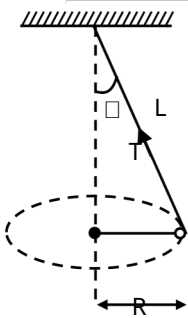
Where B = magnetic field due to a solenoid = $\mu_0 n I$

$$T = \frac{2\pi m}{q(\mu_0 n I)}$$

$$75 \times 10^{-9} = \frac{(2\pi)(9 \times 10^{-31})}{1.6 \times 10^{-19} \times 4\pi \times 10^{-7} \times n \times 1.5}$$

N = 200

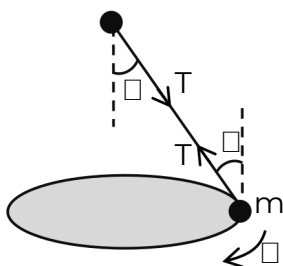
47.



A string of length L is fixed at one end and carries a mass of M at the other end. The mass makes 3 rotations per second about the vertical axis passing through end of the string as shown. The tension in the string is _____ ML .

Ans. (36)

Sol.



$$T \cos \theta = mg \quad \dots (1)$$

$$T \sin \theta = M \frac{v^2}{R} \quad \dots (2)$$

Using equation (2)

$$T \sin \theta = M \frac{v^2}{R} \sin \theta$$

$$T = M \frac{v^2}{R} = M \frac{3^2 \times 4 \pi^2 R}{2 \times 60^2} = 36 ML$$

48.

The ratio of the power of a light source S to that of the light source S is γ . S is emitting $\gamma \times 10^{10}$ photons per second at 600 nm . If the wavelength of the source S is 300 nm , then the number of photons per second emitted by S is _____ $\times 10^{10}$.

Ans. (5)

Sol. Since power emitting by a source is given as

$$P = \frac{\text{Total energy emitted}}{\text{time}} = \frac{(E \text{ photon}) \times \text{Number of photons (N)}}{t}$$

$$P = (E) \eta$$

$$\frac{P_1}{P_2} = \frac{(E_1) \eta_1}{(E_2) \eta_2} = \frac{h \nu_1 \eta_1}{h \nu_2 \eta_2}$$

$$\frac{P_1}{P_2} = \frac{\eta_1 \nu_1}{\eta_2 \nu_2}$$

Substituting the given values

$$2 = \frac{300 \times 2 \times 10^{15}}{600 \times \eta_2}$$

$$\eta_2 = \frac{1}{2} \times 10^{15} = 0.5 \times 10^{15} \text{ Photon/sec}$$

49. The increase in pressure required to decrease the volume of a water sample by 0.1% is $P \times 10^6 \text{ Nm}^{-2}$. Bulk modulus of water is $2.15 \times 10^9 \text{ Nm}^{-2}$. The value of P is _____.

Ans. (43)

Sol. Since bulk modulus is given as

$$B = \frac{P}{\frac{\Delta V}{V}}$$

$$2.15 \times 10^9 = \frac{P}{\frac{0.2}{100}}$$

$$P = 2.15 \times 10^9 \times \frac{0.2}{100} = 4.3 \times 10^6 = 43 \times 10^5 \text{ N/m}^2$$

50. Acceleration due to gravity on the surface of earth is 'g'. If the diameter of earth is reduced to one third of its original value and mass remains unchanged, then the acceleration due to gravity on the surface of the earth is _____ g.

Ans. (9)

Sol. \therefore acceleration due to gravity on surface is given by

$$g = \frac{GM}{R^2}$$

Now since diameter is reduced to $\frac{1}{3}$, radius also reduces to $\frac{1}{3}$, keeping mass constant
New value of acceleration due to gravity on Earth's surface is

$$g' = \frac{GM}{\left(\frac{R}{3}\right)^2} = 9 \frac{GM}{R^2} = 9g$$

JEE-MAIN EXAMINATION – JANUARY 2025

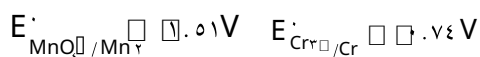
(HELD ON FRIDAY 24th JANUARY 2025)

TIME : 3 : 00PM TO 6 : 00 PM

CHEMISTRY

SECTION-A

Q1. Based on the data given below :



the strongest reducing agent is :

- (1) Mn^{2+} (2) Cr
 (3) MnO_4^- (4) Cl^-

Ans. (2)

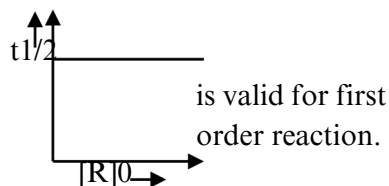
Sol. For strongest reducing agent

Reduction potential should be lowest
 Hence Cr is the strongest reducing agent.

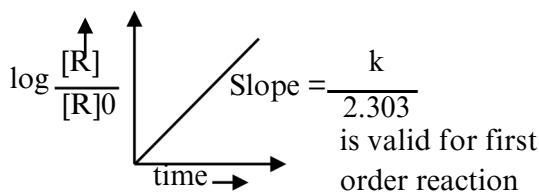
Options (2)

Q2. Given below are two statements :

Statement(I) :



Statement(II) :



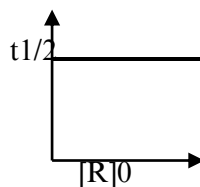
In the light of the above statements, choose the correct answer from the options given below :

- (1) Both Statement I and Statement II are false
 (2) Statement I is false but Statement II is true
 (3) Both Statement I and Statement II are true
 (4) Statement I is true but Statement II is false

Ans. (4)

TEST PAPER WITH SOLUTIONS

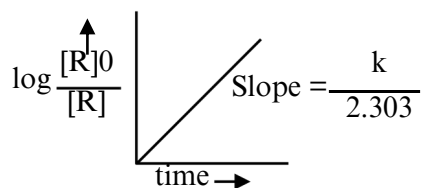
Sol. For first order reaction $t_{1/2} = \frac{0.693}{k}$



For first order reaction

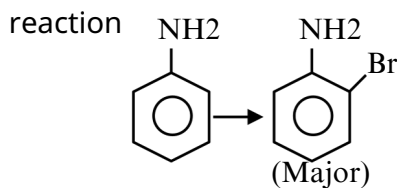
$$\log \frac{R_0}{R} = \frac{k}{2.303} t$$

$$\log \frac{R_0}{R} = \frac{k}{2.303} t$$



Options (4)

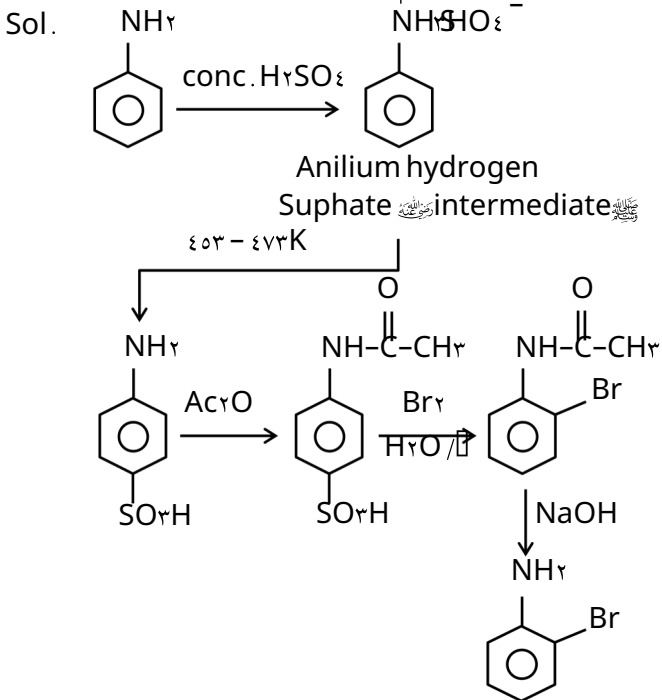
Q3. For



The correct order of set of reagents for the above conversion is :

- (1) $\text{Br}_2/\text{FeBr}_3, \text{HO}^+/\text{H}^+, \text{NaOH}$
 (2) $\text{HSO}_3^-, \text{Ac}_2\text{O}, \text{Br}_2, \text{HO}^+/\text{H}^+, \text{NaOH}$
 (3) $\text{Ac}_2\text{O}, \text{Br}_2, \text{HO}^+/\text{H}^+, \text{NaOH}$
 (4) $\text{Ac}_2\text{O}, \text{HSO}_3^-, \text{Br}_2, \text{NaOH}$

Ans. (2)



Q4. For hydrogen atom, the orbital/s with lowest energy is/are :

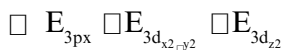
- (A) 1s (B) 2px
 (C) 2d_{x²-y²} (D) 2d_{z²}
 (E) 3p_z

Choose the correct answer from the options given below :

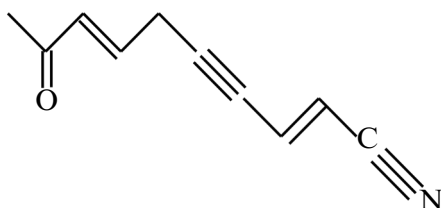
- (1) (A) and (E) only (2) (B) only
 (3) (A) only (4) (B), (C) and (D) only

Ans. (4)

Sol. In hydrogen atom the orbitals in a shell are degenerate means energy depends only on 'n'

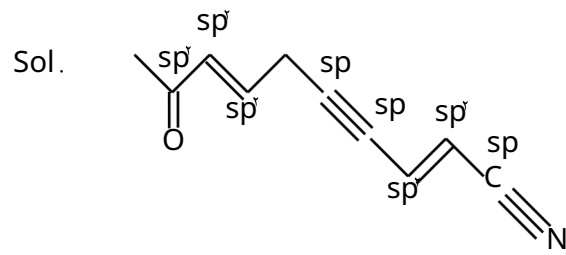


Q5. In the given structure, number of sp and sp² hybridized carbon atoms present respectively are :



- (1) 3 and 6 (2) 3 and 5
 (3) 4 and 6 (4) 4 and 5

Ans. (2)



Number of sp and sp² hybridized carbon atoms are 3 and 5.

- Q6. Which of the following mixing of 1M base and 1M acid leads to the largest increase in temperature:
 (1) 30 mL HCl and 30 mL NaOH
 (2) 30 mL CH₃COOH and 30 mL NaOH
 (3) 50 mL HCl and 20 mL NaOH
 (4) 50 mL CH₃COOH and 20 mL NaOH

Ans. (1)

Sol. Higher the number of milli moles of acid or base reacted higher will be temperature rise.

Option (4) acid or base reacted = 20 m mol
 Option (2) acid or base reacted = 30 m mol but less energy will be released by neutralisation reaction of weak acid hence option (2) can not be correct.

Option (3) = 20 m mol
 Option (1) = 30 m mol

Hence Correct Option (1)

Q7. Given below are two statements :

Statement (I) : Experimentally oxygen-oxygen bond lengths in the O₃ are found to be same and the bond length is greater than that of a O=O (double bond) but less than that of a single (O-O) bond.

Statement (II) : The strong lone pair-lone pair repulsion between oxygen atoms is responsible for the fact that the bond length in ozone is smaller than that of a double bond (O=O) but more than that of a single bond (O-O).

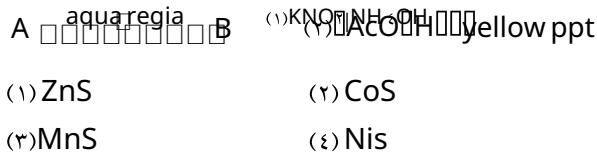
In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is true but Statement II is false
 (2) Both Statement I and Statement II are true
 (3) Both Statement I and Statement II are false
 (4) Statement I is false but Statement II is true

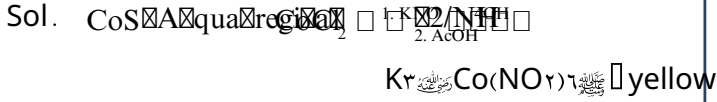
Ans. (1)

Sol. Due to resonance bond length is identical in ozone. Therefore statement I is true and statement II is false

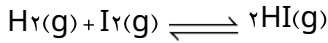
Q8. Find the compound 'A' from the following reaction sequences.



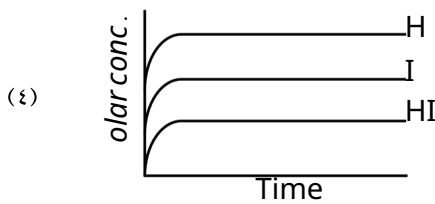
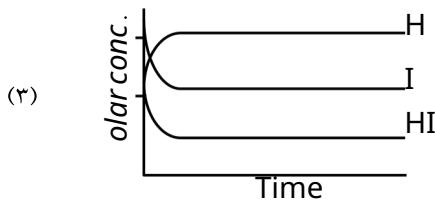
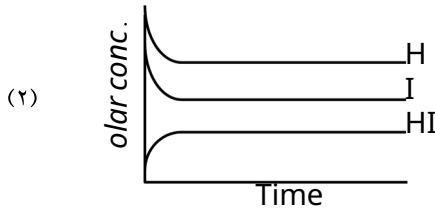
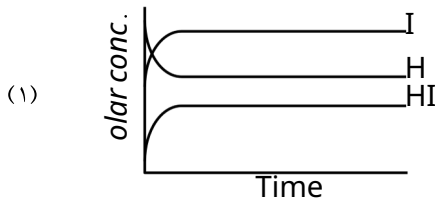
Ans. (2)



Q9. For the reaction,



Attainment of equilibrium is predicted correctly by:



Ans. (2)



Concentration of H_2 and I_2 decreases until equilibrium condition and concentration of HI increases till equilibrium condition and after equilibrium concentration of all the reactant and products remain constant.

Correct option (2)

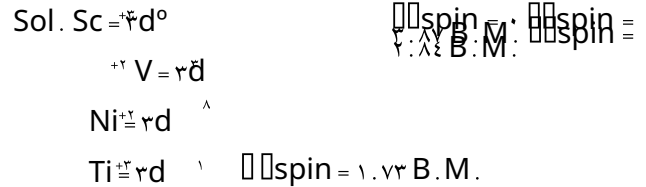
Match List-I with List-II.

List-I	List-II
(Transition metal ion)	(Spin only magnetic moment)
(A) Ti^{3+}	(B. M.)
(B) V (C) Ni (D) Sc	(I) 1.73 (II) 2.83

Choose the correct answer from the options given below :

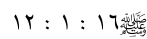
- (1) (A)-(III), (B)-(I), (C)-(II), (D)-(IV)
 (2) (A)-(III), (B)-(I), (C)-(IV), (D)-(II)
 (3) (A)-(IV), (B)-(II), (C)-(III), (D)-(I)
 (4) (A)-(II), (B)-(IV), (C)-(I), (D)-(III)

Ans. (2)



Q11. The elemental composition of a compound is 54.2% C, 9.2% H and 36.6% O. If the molar mass of the compound is 122 g mol⁻¹, the molecular formula of the compound is :

Given : The relative atomic mass of C : H : O =



- (1) $\text{C}_8\text{H}_8\text{O}_2$ (2) $\text{C}_7\text{H}_{12}\text{O}_2$
 (3) $\text{C}_7\text{H}_{12}\text{O}_2$ (4) $\text{C}_8\text{H}_8\text{O}_2$

Ans. (3)

Sol.	C	H	O
	54.2	9.2	36.6
	$\frac{54.2}{12}$	$\frac{9.2}{1}$	$\frac{36.6}{16}$
	4.516		2.287
	4.516	9.2	2.287
	$\frac{4.516}{4.516}$	$\frac{9.2}{4.516}$	$\frac{2.287}{4.516}$
	1.00	2.04	0.50

$\text{C}_2\text{H}_4\text{O}$ Empirical formula

E. F. mass = 28 + 16 = 44

16 = 44

and molar mass = 122 ($\text{C}_2\text{H}_4\text{O}$)₃

Hence molecular formula $\text{C}_6\text{H}_{12}\text{O}_3$

Correct Option (3)

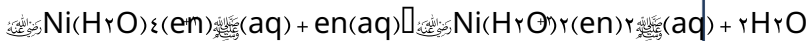
72. When Ethane-1,2-diamine is added progressively to an aqueous solution of Nickel (II) chloride, the sequence of colour change observed will be :
 (1) Pale Blue → Blue → Green → Violet
 (2) Pale Blue → Blue → Violet → Green
 (3) Green → Pale Blue → Blue → Violet
 (4) Violet → Blue → Pale Blue → Green

Ans. (3)

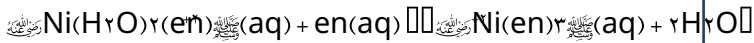
Sol.



→ Green



Pale Blue



→ Blue / purple

73. The conditions and consequence that favours the t_{2g}^3, e_g^3 configuration in a metal complex are :
 (1) weak field ligand, high spin complex
 (2) strong field ligand, high spin complex
 (3) strong field ligand, low spin complex
 (4) weak field ligand, low spin complex

Ans. (1)

Sol. For d^6

If ligand is SFL : t_{2g}^6, e_g^0 (Low spin)

If ligand is WFL : t_{2g}^4, e_g^2 (High spin)

74. Identify correct statement/s :

- (A) $-\text{OCH}_3$ and $-\text{NHCOCH}_3$ are activating group
- (B) $-\text{CN}$ and $-\text{OH}$ are meta directing group
- (C) $-\text{CN}$ and $-\text{SO}_3\text{H}$ are meta directing group
- (D) Activating groups act as ortho - and para directing groups
- (E) Halides are activating groups

Choose the correct answer from the options given below :

- (1) (A), (C) and (D) only
- (2) (A), (B) and (E) only
- (3) (A) only
- (4) (A) and (C) only

Ans. (1)

Sol. (B) $-\text{CN}$ is meta directing But $-\text{OH}$ is ortho / para directing.
 (E) Halides are deactivating groups.

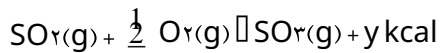
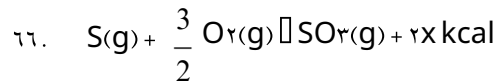
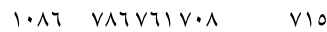
Given below are two statements :
 Statement (I) : The first ionization energy of Pb is greater than that of Sn

Statement (II) : The first ionization energy of Ge is greater than that of Si

In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is true but Statement II is false
- (2) Both Statement I and Statement II are false
- (3) Statement I is false but Statement II is true
- (4) Both Statement I and Statement II are true

Ans. (1)
 Sol. Order of I. E. (in KJ/mol) :



The heat of formation of $\text{SO}_2(\text{g})$ is given by :

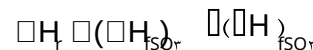
(1) $\frac{2x}{y} \text{ kcal}$

(2) $y - 2x \text{ kcal}$

(3) $2x + y \text{ kcal}$

(4) $x + y \text{ kcal}$

Ans. (2)



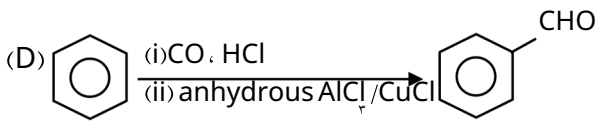
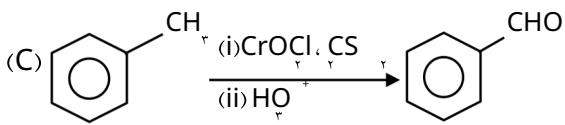
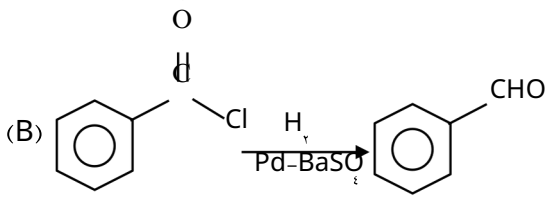
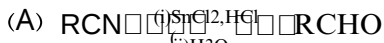
$$-y = 2x + \Delta H_f(\text{SO}_2)$$

$$\Delta H_f(\text{SO}_2) = -y - 2x$$

Option (2)

77. Match List-I with List-II

List-I



List-II

(I) Etard reaction

(II) Gatterman-Koch reaction

(III) Rosenmund reduction

(IV) Stephen reaction

Choose the correct answer from the options given below :

(1) (A)-(IV), (B)-(III), (C)-(I), (D)-(II)

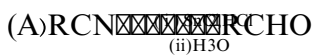
(2) (A)-(III), (B)-(IV), (C)-(II), (D)-(I)

(3) (A)-(I), (B)-(III), (C)-(II), (D)-(IV)

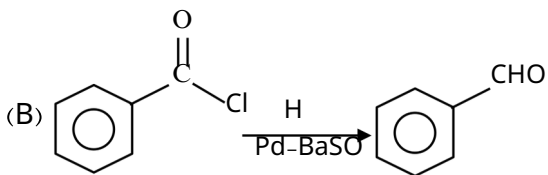
(4) (A)-(III), (B)-(IV), (C)-(I), (D)-(II)

Ans. (1)

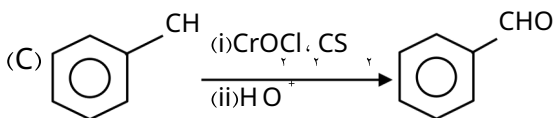
Sol. List-I



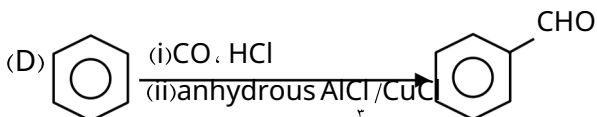
Stephen reaction



Rosenmund reduction

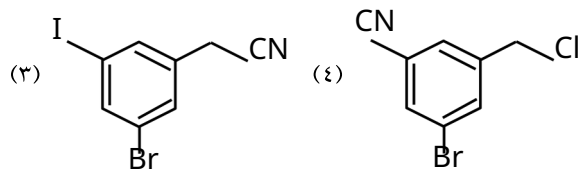
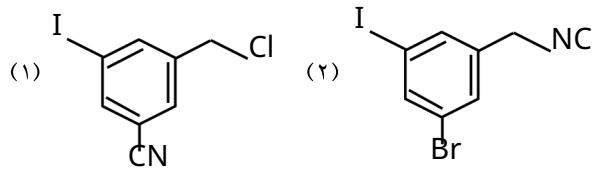
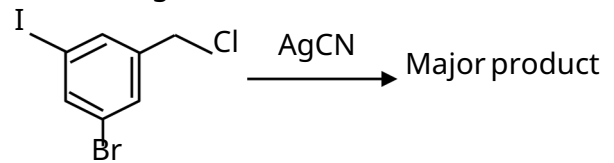


Etard reaction

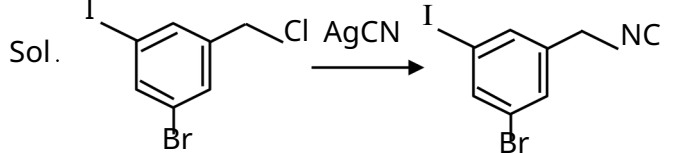


Gatterman-Koch reaction

78. The structure of the major product formed in the following reaction is :



Ans. (2)

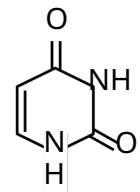


79. Match List-I with List-II.

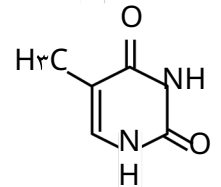
List-I

List-II

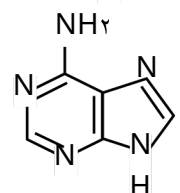
(A) Adenine (I)



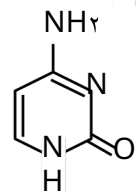
(B) Cytosine (II)



(C) Thymine (III)



(D) Uracil (IV)



Choose the correct answer from the options given below :

(1) (A)-(III), (B)-(IV), (C)-(II), (D)-(I)

(2) (A)-(III), (B)-(I), (C)-(IV), (D)-(II)

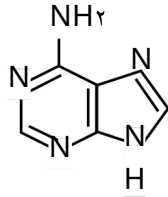
(3) (A)-(IV), (B)-(III), (C)-(II), (D)-(I)

(4) (A)-(III), (B)-(IV), (C)-(I), (D)-(II)

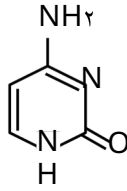
Ans. (1)

Sol.

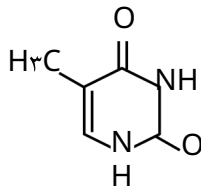
(A) Adenine (III)



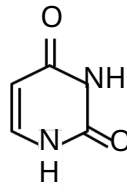
(B) Cytosine (IV)



(C) Thymine (II)



(D) Uracil (I)



60. The successive ionisation energies of an element are 800, 2327, 3608, 5024 and 32825 kJ/mol, respectively. By using the above values predict the group in which the above element is present : (1) Group 2 (2) Group 13 (3) Group 8 (4) Group 18

Ans. (2)

Sol. The IE₅ is suddenly very high therefore element must have 4 valence e⁻(s) and it belong to group 13

SECTION-B

61. The observed and normal masses of compound MX₂ are 70.6 and 178 respectively. The percent degree of ionisation of MX₂ is _____%. (Nearest integer)

Ans. (70)

Sol. $MX_2 \rightleftharpoons M^{+} + 2X^{-}$

$$i = \frac{\text{normal molar mass}}{\text{observed molar mass}}$$

$$i = \frac{178}{70.6}$$

$$1 - (2i) = \frac{178}{70.6}$$

$$2i = \frac{98.4}{70.6}$$

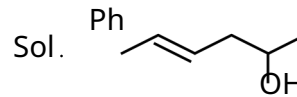
$$i = 0.70$$

percent dissociation = 70%

Ans. 70

62. The possible number of stereoisomers for o-phenylpent-2-en-3-ol is _____.

Ans. (8)



n (stereogenic unit) = 2, ∴ 8 stereoisomers are possible.

63. Consider a complex reaction taking place in three steps with rate constants k₁, k₂ and k₃ respectively. The overall rate constant k is given by the

expression $k = \sqrt{\frac{k_1 k_2}{k_3}}$. If the activation energies

of the three steps are 60, 30 and 10 kJ mol⁻¹ respectively, then the overall energy of activation in kJ mol⁻¹ is _____. (Nearest integer)

Ans. (20)

Sol. $K = \sqrt{\frac{k_1 k_2}{k_3}}$

$$A_1 e^{-E_{a1}/RT} = \sqrt{\frac{A_1 e^{-E_{a1}/RT} \cdot A_2 e^{-E_{a2}/RT}}{A_3 e^{-E_{a3}/RT}}}$$

By comparing exponential term

$$\frac{E_{a1}}{RT} = \frac{1}{2} \left(\frac{E_{a1}}{RT} + \frac{E_{a2}}{RT} \right) - \frac{E_{a3}}{RT}$$

$$E_{a1} = \frac{(E_{a1} + E_{a2})}{2} - E_{a3}$$

$$E_a = (60 + 10 - 30)/2 = 20 \text{ kJ mol}^{-1}$$

Ans. 20

Q. The hydrocarbon (X) with molar mass 80 g mol⁻¹ and 90% carbon has _____ degree of unsaturation.

Ans. (3)

Sol. Mass of carbon = $\frac{80 \times 90}{100} = 72 \text{ gm}$

Number of C-atoms = $\frac{72}{12} = 6$

Mass of hydrogen = $\frac{80 \times 10}{100} = 8 \text{ gm}$

Number of H-atoms = $\frac{8}{1} = 8$

So molecular formula C₆H₈

D. U. = $\frac{6 + 1 - 8}{2} = 3 - 0 = 3$

In Carius method of estimation of halogen, 0.20 g of an organic compound gave 0.16 g of silver bromide (AgBr). The percentage of bromine in the compound is _____ % (Nearest integer).

(Given : Molar mass of Ag is 108 and Br is 80 g mol⁻¹)

Ans. (200)

Sol. % Bromine = $\frac{\text{Molar Mass of Bromine}}{\text{Molar Mass of Silver bromide}} \times \frac{\text{Weight of AgBr}}{\text{Weight of sample}} \times 100$

$= \frac{80}{188} \times \frac{0.16}{0.20} \times 100$

$= \frac{4800}{188} = 25.53 = 200 \times 10^{-1}$