

JEE-MAIN EXAMINATION – JANUARY 2025

(HELD ON THURSDAY 23rd JANUARY 2025)

TIME : 3:00PM TO 6:00 PM

MATHEMATICS

SECTION-A

1. If in the expansion of $(1+x)(1-x)^n$, the coefficients of x and x^2 are 1 and -2 , respectively, then $p+q$ is equal to :

- (1) 1
- (2) 11
- (3) 13
- (4) 20

Ans. (3)

Sol. $(1+x)(1-x)^n = (1+x)(1-x)^p(1-x)^q$

coeff of x is $1 - n + nC_1 = 1$

$p - q = 1$

coeff of x^2 is $nC_2 - nC_1 + nC_2 = -2$

$\frac{n(n-1)}{2} - n + n = -2$

$n^2 - n - 2n + 2n = -4$

$(n-1)^2 - (n+1) = -4$

$n + 1 = 0$

$n = -1$

$p = 1$

so $p + q = 1 + 1 = 2$

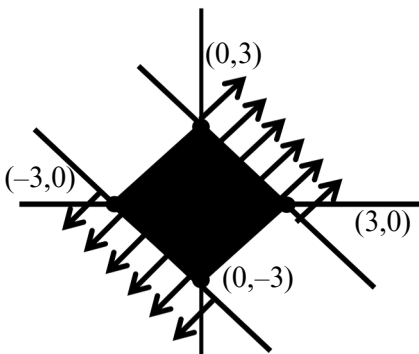
2. Let $A = \{(x, y) \in \mathbb{R} \times \mathbb{R} : |x+y| \leq 2\}$ and $B = \{(x, y) \in \mathbb{R} \times \mathbb{R} : |x| + |y| \leq 2\}$. If $C = \{(x, y) \in A \cap B : x = 0 \text{ or } y = 0\}$, then

$\int_C |x-y| dx dy$ is :

- (1) 10
- (2) 11
- (3) 24
- (4) 12

Ans. (4)

Sol.



$C = \{(x, 0) : -2 \leq x \leq 2\} \cup \{(0, y) : -2 \leq y \leq 2\}$

$\int_C |x-y| dx dy = 12$

TEST PAPER WITH SOLUTION

3. The system of equations $x+y+z=6$, $x+2y+3z=9$, $x+ay+2z=\mu$ has no solution if

- (1) $\mu = 17, \mu \neq 11$
- (2) $\mu \neq 17, \mu \neq 11$
- (3) $\mu = 10, \mu \neq 17$
- (4) $\mu = 17, \mu \neq 11$

Ans. (1)

Sol. $D = \begin{vmatrix} 1 & 1 & 1 \\ 1 & 2 & 3 \\ 1 & a & 2 \end{vmatrix} = 0$

$\mu = 17$

$D_z = \begin{vmatrix} 1 & 1 & 6 \\ 1 & 2 & 9 \\ 1 & a & \mu \end{vmatrix} = 0$

$\mu \neq 11$

4. Let $\int_0^{\pi} x \sin x dx = C$, where C is the constant of integration.

If $\int_0^{\pi} x^2 \sin x dx = 3 - 2C$, then $\int_0^{\pi} x^3 \sin x dx$ equals:

- (1) 0
- (2) 4π
- (3) $4\pi^2$
- (4) 6π

Ans. (1)

Sol. $\int_0^{\pi} x \sin x dx = C$

$\int_0^{\pi} x^2 \sin x dx = 3 - 2C$

$\int_0^{\pi} x^3 \sin x dx = 6 - 3C$

So $\int_0^{\pi} x^3 \sin x dx = 6 - 3C$

$6 - 3C = 3 - 2(3 - 2C)$

$6 - 3C = 3 - 6 + 4C$

$6 - 3C = -3 + 4C$

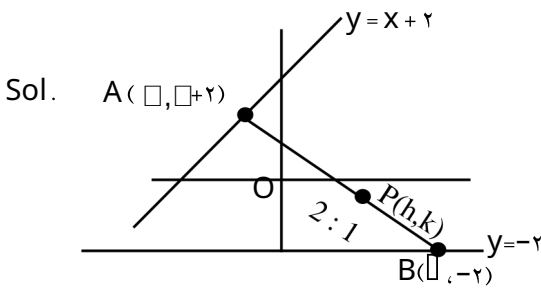
$9 = 7C \Rightarrow C = \frac{9}{7}$

So $\int_0^{\pi} x^3 \sin x dx = 0$

5. A rod of length eight units moves such that its ends A and B always lie on the lines $x - y + z = 0$ and $y + z = 0$, respectively. If the locus of the point P, that divides the rod AB internally in the ratio $2:1$ is $9(x^2 + y^2 + z^2) - 4xy + 2(x + z)y = 0$, then $4 - 2 - 2$ is equal to :

- (1) 24
- (2) 23
- (3) 21
- (4) 22

Ans. (2)



Sol. $A(x, x+z)$
 $B(z, -z)$
 $P(h, k)$
 $2:1$

$$h = \frac{2z + x}{3}$$

$$k = \frac{2(-z) + (x+z)}{3} = \frac{x - z}{3}$$

$x = 3h - 2z$
 $z = x - 3k = 3h - 2z - 3k$
 $5z = 3h - 3k$
 $z = \frac{3h - 3k}{5}$
 $x = 3h - 2(\frac{3h - 3k}{5}) = \frac{15h - 6h + 6k}{5} = \frac{9h + 6k}{5}$

$9(x^2 + y^2 + z^2) - 4xy + 2(x + z)y = 0$
 $9(\frac{81h^2 + 36hk + 36k^2}{25} + \frac{9h^2 - 6hk + 9k^2}{25} + \frac{9(3h - 3k)^2}{25}) - 4(\frac{9h + 6k}{5})(\frac{x - z}{3}) + 2(\frac{9h + 6k}{5} + \frac{3h - 3k}{5})(\frac{x - z}{3}) = 0$

$9(81h^2 + 36hk + 36k^2 + 9h^2 - 6hk + 9k^2 + 9(3h - 3k)^2) - 4(9h + 6k)(x - z) + 2(9h + 6k + 3h - 3k)(x - z) = 0$
 $9(90h^2 + 30hk + 36k^2 + 9(3h - 3k)^2) - 4(9h + 6k)(x - z) + 2(12h + 3k)(x - z) = 0$

6. The distance of the line $\frac{x}{2} + \frac{y}{3} + \frac{z}{4} = 1$ from the point $(1, 1, 1)$ along the line $\frac{x}{1} + \frac{y}{2} + \frac{z}{3} = 1$ is :

- (1) $\sqrt{17}$
- (2) $\sqrt{14}$
- (3) $\sqrt{15}$
- (4) $\sqrt{13}$

Ans. (2)

Sol. Let the parallel lines

$$\frac{x}{1} + \frac{y}{2} + \frac{z}{3} = \lambda$$

so their point of intersection is

$$(\lambda + 1, 2\lambda + \lambda, 3\lambda) = (\lambda + 1, 3\lambda, 3\lambda)$$

$$\lambda = 2\lambda + 1$$

$$2\lambda + 1 = 3\lambda \Rightarrow \lambda = 1$$

so POI is $(2, 3, 3)$

so distance $= \sqrt{(2-1)^2 + (3-1)^2 + (3-1)^2} = \sqrt{1 + 4 + 4} = \sqrt{9} = 3$

7. Let the point A divide the line segment joining the points $P(-1, -1, 2)$ and $Q(0, 0, 1)$ internally in the ratio $1:(r < 0)$. If O is the origin and

$2(OQ \cdot OA)^2 = |OP|^2 + |OQ|^2$, then the value of r

is :

- (1) 1/4
- (2) 3
- (3) $\sqrt{7}$
- (4) 7

Ans. (4)

Sol. $A = \frac{1}{1+r}P + \frac{r}{1+r}Q$

$2(OQ \cdot OA)^2 = |OP|^2 + |OQ|^2$... (1)

$OQ \cdot OA = \frac{1}{1+r} |OP| + \frac{r}{1+r} |OQ|^2$

$|OP|^2 = |OA|^2 + |AP|^2$

so by equation (1)

$$\frac{1}{1+r} |OP| + \frac{r}{1+r} |OQ|^2 = \sqrt{|OP|^2 + |AP|^2} + \frac{r}{1+r} |OQ|^2$$

$$r^2 - 14r = 0$$

$$r = 14, r = 0$$

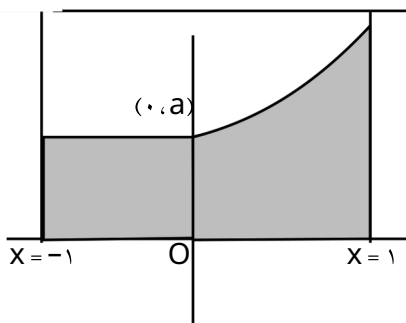
8. If the area of the region $\{(x, y) : x^2 + y^2 \leq a, e^{-|x|} - e^{-|y|} \geq 0\}$ is

$\frac{e^2 - 8e}{e}$, then the value of a is :

- (1) 7
- (2) 6
- (3) 8
- (4) 0

Ans. (4)

Sol.



required area is $\int_{-1}^1 a e^x dx$

$$a \int_{-1}^1 e^x dx$$

$$a(e^1 - e^{-1}) = a(e - \frac{1}{e})$$

9. A spherical chocolate ball has a layer of ice-cream of uniform thickness around it. When the thickness of the ice-cream layer is 1 cm, the ice-cream melts

at the rate of $\frac{1}{4}$ cm/min and the thickness of the ice-cream layer decreases at the rate

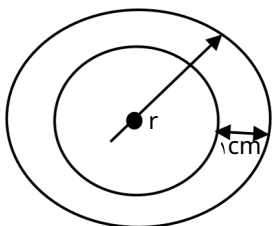
$\frac{1}{4}$ cm/min. The surface area (in cm²) of the

chocolate ball (without the ice-cream layer) is :

- (1) 220
- (2) 128
- (3) 196
- (4) 206

Ans. (4)

Sol



$$V = \frac{4}{3}\pi r^3$$

$$\frac{dV}{dt} = 4\pi r^2 \frac{dr}{dt}$$

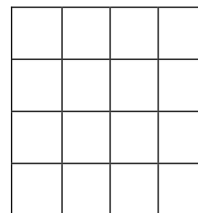
$$\frac{1}{4} = 4\pi r^2 \frac{dr}{dt}$$

$$r = 1$$

$$r = 1$$

$$\text{surface area of chocolate} = 4\pi(r-1)^2 = 206$$

10. A board has 16 squares as shown in the figure :



Out of these 16 squares, two squares are chosen at random. The probability that they have no side in common is :

- (1) $\frac{4}{5}$
- (2) $\frac{7}{10}$
- (3) $\frac{3}{5}$
- (4) $\frac{23}{30}$

Ans. (1)
Sol. Total ways for selecting any two squares = $C^2_{16} = 120$

Total ways for selecting common side squares

$$4 \times 2 \times 2 + 4 \times 2 \times 2$$

Horizontal side vertical side

$$= 24$$

so required probability

$$= \frac{120 - 24}{120}$$

$$= \frac{4}{5}$$

11. Let $x = x(y)$ be the solution of the differential equation

$$y \frac{dx}{dy} + x \sin \frac{x}{y} = y^2, \quad y < 0 \text{ and } x(1) = \frac{\pi}{2}$$

Then $\cos(x(2))$ is equal to :

- (1) $1 - 2(\log 2)^{-1}$
- (2) $2(\log 2)^{-1}$
- (3) $2(\log 2) - 1$
- (4) $1 - 2(\log 2)$

Ans. (2)

Sol. $y dy = (x dy - y dx) \sin \frac{x}{y}$

$$\frac{dy}{y} = \frac{x dy - y dx}{y^2} \sin \frac{x}{y}$$

$$\frac{dy}{y} = \sin \frac{x}{y} d \left(\frac{x}{y} \right)$$

$$\int \frac{dy}{y} = \int \sin \frac{x}{y} d \left(\frac{x}{y} \right)$$

$$\ln y = -\cos \frac{x}{y} + C$$

10. The length of the chord of the ellipse $\frac{x^2}{4} + \frac{y^2}{2} = 1$,

whose mid-point is $(\frac{1}{2}, \frac{1}{2})$, is:

- (1) $\frac{2}{3}\sqrt{15}$ (2) $\frac{5}{3}\sqrt{15}$

- Ans. (1) $\frac{1}{3}\sqrt{15}$ (3) $\sqrt{15}$

Sol. T=S

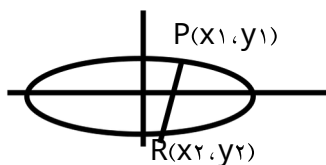
$$\frac{x \cdot 1}{\xi} + \frac{y \cdot 1}{\eta} = \frac{1}{\xi} + \frac{1}{\eta}$$

$$x + y = \eta$$

solve with ellipse

$$PR = \sqrt{(x-x_1)^2 + (y-y_1)^2}$$

$$= \sqrt{\eta(x-x_1)}$$



$$y_r = \eta - x_r$$

$$y_1 = \eta - x_1$$

$$y_r - y_1 = x_1 - x_r$$

$$x_1^2 + \eta y_1 = \xi$$

$$x_1^2 + \eta(x_1 - x_r) = \xi$$

$$1x_1^2 - 1x_r + 1 = 0$$

$$x_1 + x_r = \eta$$

$$x_1 x_r = 1/\eta$$

$$|x_r - x_1| = \sqrt{(x_1 + x_r)^2 - 4x_1 x_r}$$

$$= \sqrt{\eta^2 - 4/\eta}$$

$$PR = \sqrt{\eta} \cdot \frac{\sqrt{\eta^3 - 4}}{\sqrt{\eta}} = \sqrt{\eta^3 - 4}$$

16. Let $A = \begin{bmatrix} a & a & a \\ a & a & a \\ a & a & a \end{bmatrix}$ be a 3×3 matrix such that

$$A \begin{bmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \\ 0 & 0 & 0 \end{bmatrix} = \begin{bmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \\ 0 & 0 & 0 \end{bmatrix} \text{ and } A \begin{bmatrix} 1 & 1 & 1 \\ 1 & 1 & 1 \\ 1 & 1 & 1 \end{bmatrix} = \begin{bmatrix} 2 & 2 & 2 \\ 2 & 2 & 2 \\ 2 & 2 & 2 \end{bmatrix}$$

a_r equals:

- (1) -1 (2) 0
- (3) 2 (4) 1

Ans. (1)

Sol. Let $A = \begin{bmatrix} a & a & a \\ a & a & a \\ a & a & a \end{bmatrix}$

$$A \begin{bmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \\ 0 & 0 & 0 \end{bmatrix} = \begin{bmatrix} 0 & 0 & 0 \\ 0 & 0 & 0 \\ 0 & 0 & 0 \end{bmatrix}$$

$$A \begin{bmatrix} 1 & 1 & 1 \\ 1 & 1 & 1 \\ 1 & 1 & 1 \end{bmatrix} = \begin{bmatrix} 2 & 2 & 2 \\ 2 & 2 & 2 \\ 2 & 2 & 2 \end{bmatrix}$$

$$-a + a + a = 0 \Rightarrow a = 0$$

17. The number of complex numbers z , satisfying $|z| = 1$

and $\left| \frac{z}{\bar{z}} - \frac{\bar{z}}{z} \right| = 1$, is:

- (1) 6 (2) 8
- (3) 10 (4) 12

Ans. (4)

Sol. $z = e^{i\theta}$

$$\frac{z}{\bar{z}} = e^{i2\theta}$$

$$\left| \frac{z}{\bar{z}} - \frac{\bar{z}}{z} \right| = |e^{i2\theta} - e^{-i2\theta}| = 2|\sin 2\theta| = 1$$

^ solution in $(0, 2\pi)$

18. If the square of the shortest distance between the

lines $\frac{x-2}{1} = \frac{y-1}{2} = \frac{z-3}{3}$ and $\frac{x-1}{2} = \frac{y-3}{4} = \frac{z-5}{5}$

is $\frac{m}{n}$, where m, n are coprime numbers, then $m+n$

is equal to:

- (1) 6 (2) 9
- (3) 11 (4) 13

Ans. (2)

Sol. a $\begin{vmatrix} \alpha & \beta & \gamma \\ \beta & \gamma & \alpha \\ \gamma & \alpha & \beta \end{vmatrix}$
 b $\begin{vmatrix} \alpha & \beta & \gamma \\ \beta & \gamma & \alpha \\ \gamma & \alpha & \beta \end{vmatrix}$

$$P = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ \alpha & \beta & \gamma \\ \beta & \gamma & \alpha \end{vmatrix}$$

$$P \hat{i} + Q \hat{j}$$

 b $\alpha \hat{i} + \beta \hat{j} + \gamma \hat{k}$

$$S_d = \frac{|\alpha \beta \gamma|}{|\alpha \beta \gamma|}$$

$$\frac{1}{\sqrt{\alpha}}$$

$$S_d = \frac{\xi}{\alpha}$$

19. If $I = \int_0^{\frac{\pi}{2}} \frac{\sin^2 x}{\sin^2 x + \cos^2 x} dx$,
 then $\int_0^{\frac{\pi}{2}} \frac{x \sin x \cos x}{\sin^4 x + \cos^4 x} dx$ equals:
 (1) $\frac{1}{16}$ (2) $\frac{1}{4}$
 (3) $\frac{1}{8}$ (4) $\frac{1}{12}$
 Ans. (1)

Sol. For I
 Apply king (P-α) and add

$$I = \int_0^{\frac{\pi}{2}} \frac{x \sin x \cos x}{\sin^4 x + \cos^4 x} dx$$

 Apply king and add

$$I_2 = \int_0^{\frac{\pi}{2}} \frac{\tan x \sec x dx}{\tan^2 x + 1}$$

 put $\tan x = t$

$$\frac{1}{\lambda} \frac{dt}{1+t^2}$$

$$\frac{1}{\lambda} \cdot \frac{1}{2} \cdot \frac{1}{1+t^2}$$

20. $\lim_{x \rightarrow \infty} \frac{2x^2 + 3x + 5}{3x^2 + 5x + 4} \sqrt{\frac{x}{3x^2 + 2}}$ is equal to:
 (1) $\frac{2}{\sqrt{3e}}$ (2) $\frac{2e}{\sqrt{3}}$
 (3) $\frac{2e}{3}$ (4) $\frac{2}{3\sqrt{e}}$

Ans. (4)
 Sol. $\lim_{x \rightarrow \infty} \frac{2x^2 + 3x + 5}{3x^2 + 5x + 4} \sqrt{\frac{x}{3x^2 + 2}}$

$$\lim_{x \rightarrow \infty} \frac{2 + \frac{3}{x} + \frac{5}{x^2}}{3 + \frac{5}{x} + \frac{4}{x^2}} \sqrt{\frac{1}{3x + \frac{2}{x}}}$$

$$\lim_{x \rightarrow \infty} \frac{2}{3} \cdot \frac{1}{\sqrt{3x}}$$

$$\frac{2}{3} \cdot \frac{1}{\sqrt{3e}}$$

SECTION-B

21. The number of ways, α boys and β girls can sit in a row so that either all the boys sit together or no two boys sit together, is _____.

Ans. (1728)
 Sol. A : number of ways that all boys sit together = α! × β!
 B : number of ways if no 2 boys sit together = β! × α!
 A + B = _____
 Required no. of ways = α! × β! + β! × α! = 1728.
 Let α, β be the roots of the equation x² - ax - b = 0.

22. with Im(α) > Im(β). Let P = α - β. If P³ = 5√7i, P⁴ = 3√7i, P⁵ = 11√7i and P⁶ = 45√7i, then P⁴ is equal to _____.

Ans. (31)
 Sol. α + β = a, αβ = -b
 P = α - β

$$\alpha^2 - \beta^2 = a(\alpha + \beta) - b(\alpha - \beta)$$

$$\alpha^2 - \beta^2 = a^2 - b$$

30. The energy of a system is given as $E(t) = \frac{1}{2} e^{-\lambda t}$ where t is the time and $\lambda = 0.3 \text{ s}^{-1}$. The errors in the measurement of λ and t are 1.2% and 1.6% respectively. At $t = 0.5 \text{ s}$, maximum percentage error in the energy is:

- (1) 1.4%
- (2) 11.6%
- (3) 1%
- (4) 1.8%

Ans. (3)

Sol. $E = \frac{1}{2} e^{-\lambda t}$

$\ln E = \frac{1}{2} \ln \frac{1}{2} - \lambda t$

$$\frac{dE}{E} = \frac{d\lambda}{\lambda} + \frac{dt}{t}$$

$\frac{dE}{E}_{\max} = \frac{1.2}{100} + (0.3 \times 1.6 \times 0.5) \times 100$
 $= 1.2\%$

31. In photoelectric effect an em-wave is incident on a metal surface and electrons are ejected from the surface. If the work function of the metal is 2.1 eV and stopping potential is 1 V , what is the wavelength of the em-wave?

(Given $hc = 1242 \text{ eVnm}$ where h is the Planck's constant and c is the speed of light in vacuum.)

- (1) 400 nm
- (2) 600 nm
- (3) 200 nm
- (4) 300 nm

Ans. (4)

Sol. $eV = E - \phi$

$1 \text{ eV} = E - 2.1 \text{ eV}$

$E = 3.1 \text{ eV}$

$E = \frac{hc}{\lambda}$

$\lambda = \frac{1242}{3.1} = 400 \text{ nm}$

32. A circular disk of radius R meter and mass M kg is rotating around the axis perpendicular to the disk. An external torque is applied to the disk such that $\theta(t) = \alpha t + \lambda t^2$, where $\theta(t)$ is the angular position of the rotating disc as a function of time t .

How much power is delivered by the applied torque, when $t = 2 \text{ s}$?

- (1) $6 \alpha MR^2$
- (2) $7 \alpha MR^2$
- (3) $1.8 \alpha MR^2$
- (4) $8 \alpha MR^2$

Ans. (1)

Sol. $\theta = \alpha t + \lambda t^2$

$\omega = \frac{d\theta}{dt} = \alpha + 2\lambda t$

$\alpha = \frac{d\omega}{dt} = 2\lambda$

$\tau = I\alpha$

$= \frac{1}{2} MR^2 \alpha$

$\tau = \frac{1}{2} MR^2 (2\lambda)$

$\tau = MR^2 \lambda (1 + 2t)$

Put $t = 2$

$\tau = 6 \alpha MR^2$

Water flows in a horizontal pipe whose one end is closed with a valve. The reading of the pressure gauge attached to the pipe is P . The reading of the pressure gauge falls to P' when the valve is opened. The speed of water flowing in the pipe is proportional to

- (1) $\sqrt{P - P'}$
- (2) $(P - P')$
- (3) $(P - P')^2$
- (4) $P - P'$

Ans. (1)

Sol. By Bernoulli equation

$P + \frac{1}{2} \rho v^2 = P' + \frac{1}{2} \rho V^2$

$v = \sqrt{2(P - P')}$

33. Match List-I with List-II.

List-I	List-II
(A) Permeability of free space	(I) $M^2 L T^{-2} A^{-2}$
(B) Magnetic field	(II) $M T A^{-1}$
(C) Magnetic moment	(III) $M L^2 A^{-1}$
(D) Torsional constant	(IV) $L A$

Choose the correct answer from the options given below : (1) (A)-(I), (B)-(IV), (C)-(II), (D)-(III) (2) (A)-(II), (B)-(I), (C)-(III), (D)-(IV) (3) (A)-(IV), (B)-(III), (C)-(I), (D)-(II) (4) (A)-(III), (B)-(II), (C)-(IV), (D)-(I)

Ans. (4)

Sol. $B = \frac{\mu_0 I}{2\pi r}$

$\frac{MLT^{-2}A^{-1}}{A^{-1}} = \frac{MLT^{-2}}{A}$

magnetic field $F = qvB$

$B = \frac{MLT^{-2}}{ATL/T} = \frac{MTA}{A}$

$M = \frac{MLT^{-2}}{A} = \frac{MLT^{-2}}{A}$

$\frac{MLT^{-2}}{A} = \frac{MLT^{-2}}{A}$

30. If a satellite orbiting the Earth is 4 times closer to the Earth than the Moon, what is the time period of rotation of the satellite? Given rotational time period of Moon = 27 days and gravitational attraction between the satellite and the moon is neglected.

(1) 1 day (2) 11 days
(3) 27 days (4) 3 days

Ans. (1)

Sol. $T \propto R^2$

$\frac{T_m}{T_s} = \left(\frac{R_m}{R_s}\right)^2$

$T = \frac{27}{4} = 6.75$ days

31. Two point charges $-4 \mu C$ and $4 \mu C$ constituting an electric dipole, are placed at $(-4, 0, 0)$ cm and $(4, 0, 0)$ cm in a uniform electric field of strength 10^6 NC. The work done on the dipole in rotating it from the equilibrium through 180° is:

- (1) 14.4 mJ (2) 18.4 mJ
(3) 12.4 mJ (4) 16.4 mJ

Ans. (1)

Sol. $U = -PE \cos \theta$

$W_{ext} = U - U_i = -PE \cos 180^\circ + PE \cos 0^\circ$

$W_{ext} = 2PE$

$= 2 \times (4 \times 10^{-6}) \times (10^6) \times 10^{-2}$

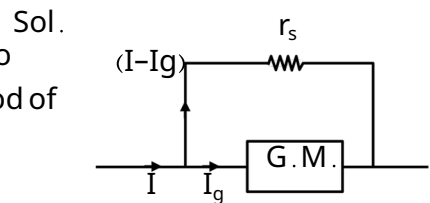
$= 14.4 \times 10^{-2}$

$= 14.4 \text{ mJ}$

37. A galvanometer having a coil of resistance 20Ω need 20 mA of current for full-scale deflection. If a maximum current of 1 A is to be measured using this galvanometer, the resistance of the shunt to be added to the galvanometer should be $\frac{r}{X}$, where X is

- (1) 447 (2) 298
(3) 149 (4) 596

Ans.



$I_g R_g = (I - I_g) r_s$

$20 \times 10^{-2} \times 20 = (1 - 0.02) \times r_s$

$r_s = \frac{0.6 \times 20}{0.98} = 12.24$

$X = \frac{20}{12.24} = 1.63$

38. The width of one of the two slits in Young's double slit experiment is d while that of the other slit is xd . If the ratio of the maximum to the minimum intensity in the interference pattern on the screen is $9 : 1$ then what is the value of x ? (Assume that the field strength varies according to the slit width.)

- (1) 2 (2) 3
(3) 0 (4) 4

Ans.

Sol. $I \propto (\text{width})^2$

$\frac{I_1}{I_2} = \left(\frac{d}{xd}\right)^2 = \frac{1}{x^2}$

$\frac{\sqrt{I_1}}{\sqrt{I_2}} = \frac{1}{x}$

$\frac{\sqrt{9}}{\sqrt{1}} = \frac{1}{x}$

$3 = \frac{1}{x}$

$x = \frac{1}{3}$

39. Given below are two statements. One is labelled as Assertion (A) and the other is labelled as Reason (R). Assertion (A) : The binding energy per nucleon is found to be practically independent of the atomic number A, for nuclei with mass numbers between 20 and 100. Reason (R) : Nuclear force is long range. In the light of the above statements, choose the correct answer from the options given below : (1) (A) is false but (R) is true (2) (A) is true but (R) is false (3) Both (A) and (R) are true and (R) is the correct

explanation of (A)

(4) Both (A) and (R) are true but (R) is NOT the correct explanation of (A)

Ans. (2)

Sol. Conceptual Water of mass m gram is slowly heated to increase the temperature from T to T'. The change in entropy of the water, given specific heat of water is 1 J/kgK, is :

- (1) zero (2) $m(T'-T)$
 (3) $m \ln \frac{T'}{T}$ (4) $m \ln \frac{T}{T'}$

Ans. (4)

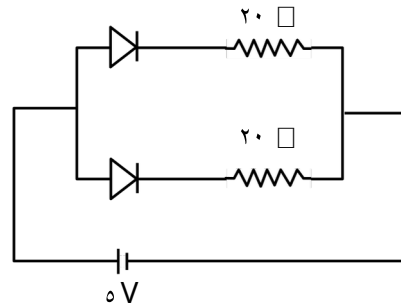
Sol. $dQ = msdT$

$$dS = \frac{dQ}{T} = \frac{msdT}{T}$$

$$\Delta S = \int_{T_i}^{T_f} \frac{msdT}{T} = ms \ln \frac{T_f}{T_i}$$

$$\Delta S = m \ln \frac{T_f}{T_i}$$

41. What is the current through the battery in the circuit shown below:



- (1) 1.0 A (2) 1.5 A
 (3) 0.5 A (4) 0.25 A

Ans. (3)

Sol. Both are forward biased

hence $R = 10 \Omega$

$$i = \frac{V}{R} = \frac{6}{10} = 0.6 \text{ A}$$

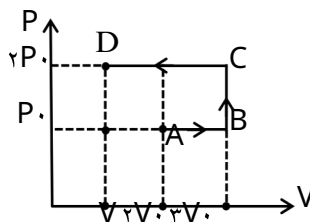
42. A plane electromagnetic wave of frequency 2.0 MHz travels in free space along the +x direction. At a particular point in space and time, the electric field vector of the wave is $E = 9.3 \text{ V/m}$. Then, the magnetic field vector of the wave at that point is-

- (1) $B_z = 9.3 \times 10^{-8} \text{ T}$ (2) $B_z = 1.00 \times 10^{-8} \text{ T}$
 (3) $B_z = 6.2 \times 10^{-8} \text{ T}$ (4) $B_z = 3.1 \times 10^{-8} \text{ T}$

Ans. (4)

Sol. $E = BC$
 $9.3 = B \times 3 \times 10^6$
 $B = \frac{9.3}{3 \times 10^6} = 3.1 \times 10^{-8} \text{ T}$

43.



Using the given P-V diagram, the work done by an ideal gas along the path ABCD is- (2) 3 PV

- (1) 4 PV (2) 3 PV
 (3) -4 PV (4) -3 PV

Ans. (2)

Sol. $w_{ABCD} = W_{AB} + W_{BC} + W_{CD}$
 $= PV + (-rP \times rV)$
 $= PV - \epsilon PV$
 $= -rPV$

εε. A concave mirror of focal length f in air is dipped in a liquid of refractive index μ . Its focal length in the liquid will be :

(1) $\frac{f}{\mu}$ (2) $\frac{f}{(\mu-1)}$
 (3) μf (4) f

Ans. (4)

Sol. Focal length of mirror will not change because focal length of mirror doesn't depend on medium.

εο. A massless spring gets elongated by amount x under a tension of $οN$. Its elongation is x under the tension of νN . For the elongation of $(οx - \nu x)$, the tension in the spring will be.

(1) $οN$ (2) νN
 (3) $οN$ (4) νN

Ans. (3)

Sol. $kx = οN$
 $kx = \nu N$
 $k(οx - \nu x) = οkx - \nu kx$
 $= ο \times ο - \nu \times \nu = 11 N$

SECTION-B

εε. An air bubble of radius λ mm is observed at a depth of r cm below the free surface of a liquid having surface tension $0.090 J/m$ and density $10 kg/m^3$. The difference between pressure inside the bubble and atmospheric pressure _____ N/m.

(Take $g = 10 m/s^2$)

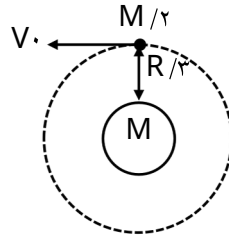
Ans. (2190)

Sol. $P_{in} = P_{atm} + \rho gh + \frac{2T}{R}$
 $P_{out} = P_{atm}$
 $\Delta P = P_{in} - P_{out} = \rho gh + \frac{2T}{R}$
 $= 10 \times 10 \times 2 + \frac{2 \times 0.090}{0.001}$
 $= 200 + 180$
 $= 380$

εν. A satellite of mass $\frac{M}{\gamma}$ is revolving around earth in a circular orbit at a height of $\frac{R}{\epsilon}$ from earth surface. The angular momentum of the satellite is $M \sqrt{\frac{GMR}{x}}$. The value of x is _____, where M and R are the mass and radius of earth, respectively (G is the gravitational constant)

Ans. (3)

Sol. (i) If earth is assumed to be stationary



orbital velocity $v = \sqrt{\frac{GM}{\epsilon R / \gamma}} = \sqrt{\frac{\gamma GM}{\epsilon R}}$

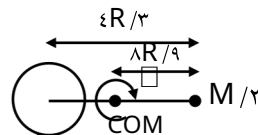
Angular momentum of satellite $L = v \cdot \frac{M}{\gamma} \cdot \frac{\epsilon R}{\gamma}$

$= \frac{M}{\gamma} \cdot \sqrt{\frac{\gamma GM}{\epsilon R}} \cdot \frac{\epsilon R}{\gamma}$

$= M \sqrt{\frac{GMR}{\epsilon}}$

$x = \epsilon$

(ii) Since mass of satellite is comparable to the mass of earth.



$\frac{GM}{\gamma} = \frac{M}{\gamma} \cdot \frac{\lambda R}{9}$

$\lambda = 4R$

$\lambda = \frac{GM}{\gamma}$

$\lambda = \frac{GM}{\gamma}$

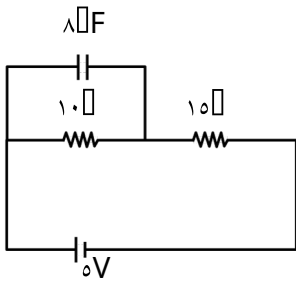
Angular momentum of satellite about common centre of mass.

$L = \frac{M}{\gamma} \cdot \frac{\lambda R}{9} \cdot v$

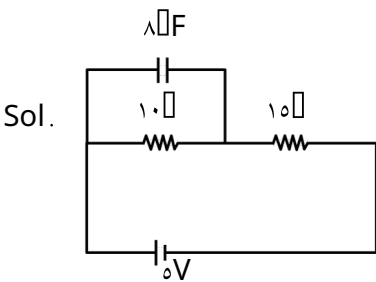
$L = \frac{M}{\gamma} \sqrt{\frac{8GM}{81}}$

$x = \frac{\lambda}{\epsilon} = 1$

At steady state the charge on the capacitor, as shown in the circuit below, is _____ μC .



Ans. (16)



Sol.

$$i = \frac{E}{R}$$

$$Q = CV$$

$$Q = (10 \times 10) = \frac{100}{10} = 10$$

$$Q = \frac{100 \times 10^{-6}}{10} = 10 \mu\text{C}$$

A time varying potential difference is applied between the plates of a parallel plate capacitor of capacitance $2.0 \mu\text{F}$. The dielectric constant of the medium between the capacitor plates is 1. It produces an instantaneous displacement current of 0.20 mA in the intervening space between the capacitor plates. the magnitude of the rate of change of the potential difference will be _____

V/s

Ans. (100)

Sol. $\frac{CdV}{dt} = I_d$

$$\frac{dV}{dt} = \frac{I_d}{C}$$

$$= \frac{0.20 \times 10^{-3}}{2.0 \times 10^{-6}}$$

$$= 100$$

In a series LCR circuit, a resistor of 300Ω , a capacitor of 20 nF and an inductor of 100 mH are used. For maximum current in the circuit, the angular frequency of the ac source is _____ $\times 10^4$ radians/s.

Ans. (2)

Sol. $\omega = \frac{1}{\sqrt{LC}}$

$$= \frac{1}{\sqrt{20 \times 10^{-9} \times 100 \times 10^{-3}}}$$

$$= \frac{10^6}{0.01} = 2 \times 10^4$$

JEE-MAIN EXAMINATION – JANUARY 2025

(HELD ON THURSDAY 23rd JANUARY 2025)

TIME : 3 : 00PM TO 6 : 00 PM

CHEMISTRY

TEST PAPER WITH SOLUTION

SECTION-A The effect of temperature reactions on spontaneity of are represented as:

	ΔH	ΔS	Temperature	Spontaneity
(A)	+	-	any T	Non spontaneous
(B)	+	+	low T	spontaneous
(C)	-	-	low T	Non spontaneous
(D)	-	+	any T	spontaneous

- (1) (B) and (D) only
 (2) (A) and (D) only
 (3) (B) and (C) only
 (4) (A) and (C) only

Ans. (3)

Sol. $\therefore \Delta G = \Delta H - T\Delta S$

For spontaneity of reaction : $\Delta G = -ve$

Standard electrode potentials for a few half cells are mentioned below:

$E^\circ_{Cu^{2+}/Cu} = 0.34V, E^\circ_{Zn^{2+}/Zn} = 0.76V$

$E^\circ_{Ag^+/Ag} = 0.80V, E^\circ_{Mg^{2+}/Mg} = 2.37V$

Which one of the following cells gives the most negative value of ΔG° :

- (1) $Zn|Zn^{2+}(M)||Ag^+(M)|Ag$
 (2) $Zn|Zn^{2+}(M)||Mg^{2+}(M)|Mg$
 (3) $Ag|Ag^+(M)||Mg^{2+}(M)|Mg$
 (4) $Cu|Cu^{2+}(M)||Ag^+(M)|Ag$

Ans. (1)

Sol. $\therefore \Delta G^\circ = -nFE^\circ$

Option (1) $E^\circ = 0.80 + 0.76$
 $= 1.56V$

$\Delta G^\circ = -2 \times F \times 1.56$
 $= -3.12V$

Option (2) $E^\circ = -2.37 + 0.76$
 $= -1.61V$

$\Delta G^\circ = -2 \times F \times (-1.61)$
 $= +3.22V$

Option (3) $E^\circ = -2.37 - 0.80$
 $= -3.17V$

$\Delta G^\circ = -2 \times F \times (-3.17)$
 $= +6.34V$

Option (4) $E^\circ = 0.80 - 0.34$
 $= 0.46V$

$\Delta G^\circ = -2 \times F \times 0.46$
 $= -0.92V$

Q3. The α -Helix and β -Pleated sheet structures of protein are associated with its:

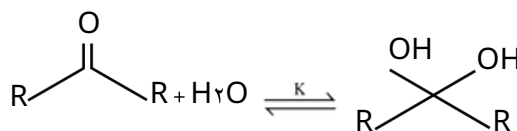
- (1) quaternary structure
 (2) primary structure
 (3) secondary structure
 (4) tertiary structure

Ans. (3)

Sol. β -helix and β -pleated sheet belongs to secondary structure of protein, which have hydrogen bonds.

Q4. Given below are two statements:

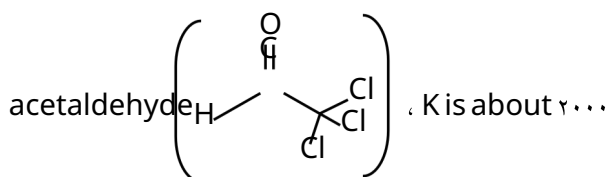
Consider the following reaction



Statement (I) : In the case of formaldehyde

$\text{H}-\overset{\text{O}}{\parallel}{C}-\text{H}$, K is about 2×10^4 , due to small substituents, hydration is faster.

Statement (II) : In the case of trichloro



due to -I effect of -Cl.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I true but Statement II is false
 (2) Both Statement I and Statement II are true
 (3) Statement I is false but Statement II is true
 (4) Both Statement I and Statement II are false

Ans. (2)

Sol. $k_{eq} = 2 \times 10^4$ is for HCHO

$k_{eq} = 2 \dots$ is for chloral

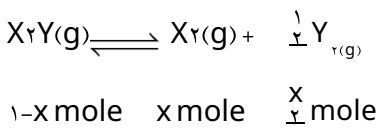
Both data is given in clayden and warren book. $k_{eq} < 1$ because HCHO and chloral are more electrophilic.

Consider the reaction
 $X_r Y(g) \rightleftharpoons X_r(g) + \frac{1}{r} Y_r(g)$
 The equation representing correct relationship

between the degree of dissociation (x) of $X_r Y(g)$ with its equilibrium constant K_p is
 Assume x to be very very small
 (r) $x \propto \sqrt{\frac{K_p}{P}}$ (s) $x \propto \sqrt{\frac{K_p}{P}}$

Ans. (r)

Sol. 1 mole



$$K_{X_r Y} = \frac{1 \cdot x \cdot P}{1 \cdot \frac{x}{r} \cdot P}$$

$$K_{X_r} = \frac{x \cdot P}{1 \cdot \frac{x}{r} \cdot P}$$

$$K_{Y_r} = \frac{x/r \cdot P}{1 \cdot \frac{x}{r} \cdot P}$$

$$K_p = \frac{x \cdot P \cdot x \cdot P}{1 \cdot \frac{x}{r} \cdot P \cdot \frac{x}{r} \cdot P} = \frac{x^2 P^2}{1 \cdot \frac{x}{r} \cdot P \cdot \frac{x}{r} \cdot P}$$

$$K_p = \frac{x^2 P^2}{1 \cdot \frac{x}{r} \cdot P \cdot \frac{x}{r} \cdot P} = \frac{x^2 P^2}{\frac{x^2}{r^2} P^2} = r^2$$

∴ x to be very very small

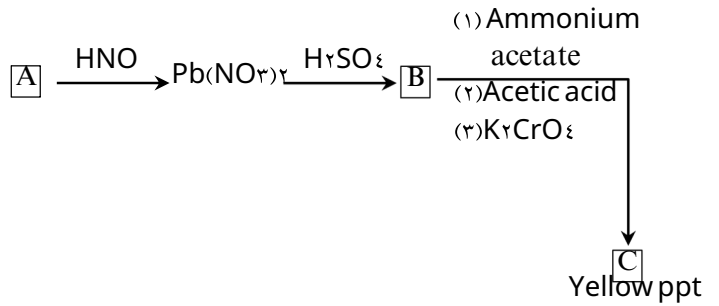
$$K_p = \frac{x^{r/r} \cdot P^r}{(r)^r}$$

$$x^r = \frac{K_p \cdot r^r}{P^r}$$

$$x^r = \frac{K_p \cdot r^r}{P^r}$$

$$x = \sqrt[r]{\frac{K_p \cdot r^r}{P^r}}$$

Identify A, B and C in the given below reaction sequence



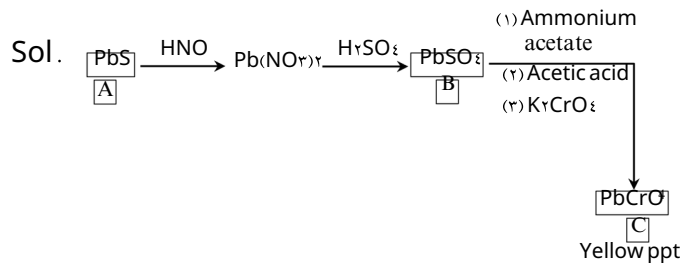
(i) $PbCl_2$, $PbSO_4$, $PbCrO_4$

(ii) PbS , $PbSO_4$, $PbCrO_4$

(iii) PbS , $PbSO_4$, $Pb(CH_3COO)_2$

(iv) $PbCl_2$, $Pb(SO_4)_2$, $PbCrO_4$

Ans. (ii)



Given below are two statements: Statement (I): The boiling points of alcohols and phenols increase with increase in the number of C-atoms. Statement (II): The boiling points of alcohols and phenols are higher in comparison to other class of compounds such as ether, haloalkanes

choose the correct answer from the options given below:

(i) Both Statement I and Statement II are false

(ii) Statement I is false but Statement II is true

(iii) Statement I is true but Statement II is false

(iv) Both Statement I and Statement II are true

Ans. (iv)

Sol. B.P. ∝ M.W.

B.P. ∝ Inter molecular hydrogen bonding

Alcohol & Phenol have intermolecular H-bonding

58. When a non-volatile solute is added to the solvent, the vapour pressure of the solvent decreases by 10 mm of Hg. The mole fraction of the solute in the solution is 0.2. What would be the mole fraction of the solvent if decrease in vapour pressure is 20 mm of Hg?

- (1) 0.2 (2) 0.4
(3) 0.6 (4) 0.8

Ans. (1)

Sol. $\therefore P^0 - P = X_{\text{solute}}$

and $\therefore 10 = 0.2$

$\therefore 20 = X$

$X_{\text{solvent}} = 1 - X_{\text{solute}}$

$$= 1 - 0.2$$

$$= 0.8$$

59. Given below are two statements:

Statement (I): For a given shell, the total number of allowed orbitals is given by n^2 .

Statement (II): For any subshell, the spatial orientation of the orbitals is given by $-l$ to $+l$ values including zero.

In the light of the above statements, choose the correct answer from the options given below:

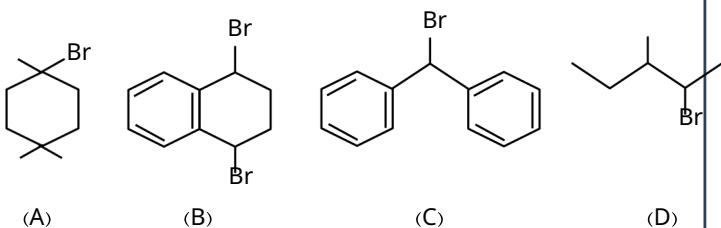
- (1) Statement I is true but Statement II is false
(2) Statement I is false but Statement II is true
(3) Both Statement I and Statement II are true
(4) Both Statement I and Statement II are false

Ans. (3)

Sol. For a shell total number of orbitals $= n^2$

Magnetic quantum number have values $(-l$ to $+l)$ including 0.

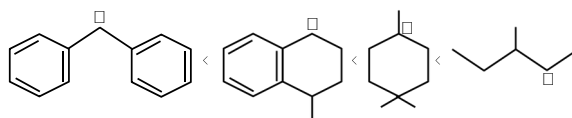
60. The ascending order of relative rate of solvolysis of following compounds is



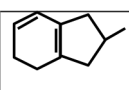
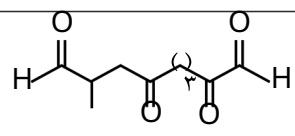
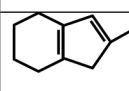
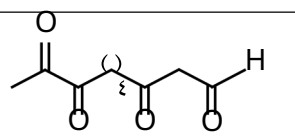
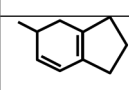
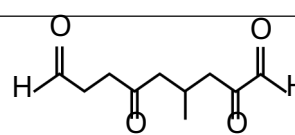
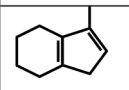
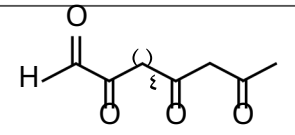
- (1) (D) > (A) > (B) > (C) (2) (C) > (B) > (A) > (D)
(3) (D) > (B) > (A) > (C) (4) (C) > (D) > (B) > (A)

Ans. (1)

Sol. Solvolysis or S_N1 stability of carbocation
Stability order



61. Match List - I with List - II.

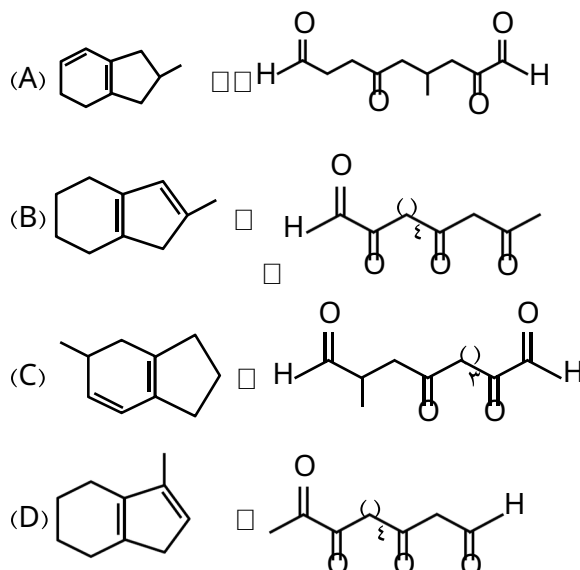
List - I (Isomers of $C_{10}H_{18}$)	List - II (Ozonolysis product)
(A) 	(I) 
(B) 	(II) 
(C) 	(III) 
(D) 	(IV) 

Choose the correct answer from the options given below:

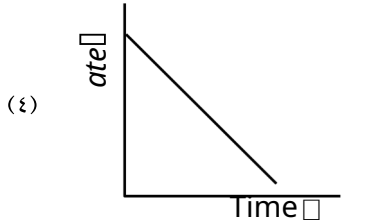
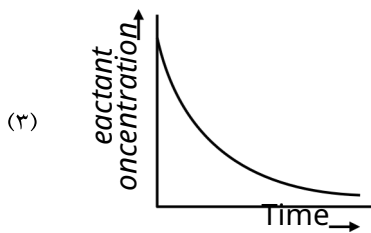
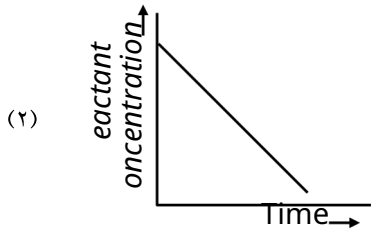
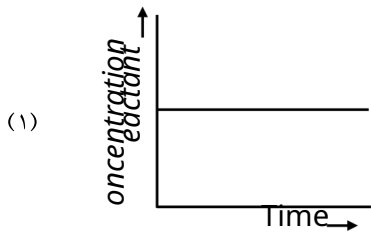
- (1) (A)-(II), (B)-(III), (C)-(I), (D)-(IV)
(2) (A)-(III), (B)-(IV), (C)-(I), (D)-(II)
(3) (A)-(III), (B)-(II), (C)-(I), (D)-(IV)
(4) (A)-(I), (B)-(IV), (C)-(III), (D)-(II)

Ans. (2)

Sol. Ozonolysis product



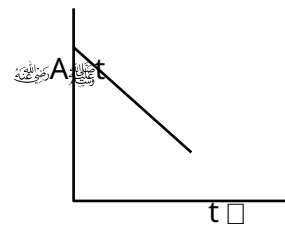
72. Which of the following graphs most appropriately represents a zero order reaction :



Ans. (2)

Sol. For zero order reaction : $\frac{d[A]}{dt} = -k$

$$A = a - kt$$



73. Match List - I with List - II.

List - I		List - II	
(A)	Bronze	(I)	Cu, Ni, Fe
(B)	Brass	(II)	Cr, Ni, C
(C)	UK silver coin	(III)	Cu, Zn, Cu
(D)	Stainless Steel	(IV)	Sn

Choose the correct answer from the options given below : (1) (A)-(IV), (B)-(II), (C)-(III), (D)-(I) (2) (A)-(IV), (B)-(III), (C)-(I), (D)-(II) (3) (A)-(III), (B)-(I), (C)-(IV), (D)-(II) (4) (A)-(III), (B)-(IV), (C)-(II), (D)-(I)

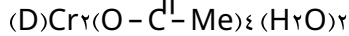
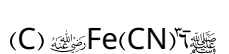
Sol. Bronze □ Cu, Sn

Brass □ Cu, Zn

UK silver coin □ Cu, Ni

Stainless steel □ Fe, Cr, Ni, C

74. Identify the coordination complexes in which the central metal ion has d configuration.



Choose the correct answer from the options given below :

(1) (C) and (E) only

(2) (B), (C) and (D) only

(3) (B) and (D) only

(4) (A), (B) and (E) only

Ans. (3)

Sol. $Fe^{+3} = [Ar] 3d^5$

$Mn^{+2} = [Ar] 3d^5$

$Fe^{+2} = [Ar] 3d^6$

$Cr^{+2} = [Ar] 3d^4$

$Ni^{+2} = [Ar] 3d^8$

75. Given below are the atomic numbers of some group 14 elements. The atomic number of the element with lowest melting point is :

(1) 14

(2) 16

(3) 82

(4) 50

Ans. (4)

Sol. Order of M.P. of group 14 : $C < Si < Ge < Pb < Sn$

element M.P. (°C)

$Z = 6 = C$ 3550

$Z = 14 = Si$ 1410

$Z = 32 = Ge$ 937

$Z = 50 = Sn$ 232

$Z = 82 = Pb$ 327

76. pH of water is 7 at 25°C. If water is heated to 80°C, its pH will :

(1) Decrease

(2) Remains the same

(3) H⁺ concentration increases, OH⁻ concentration decreases

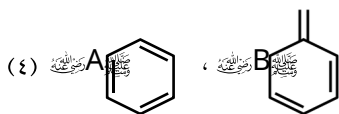
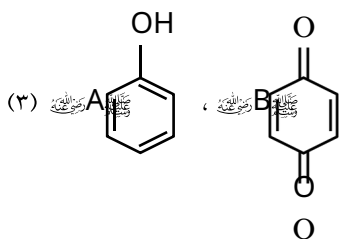
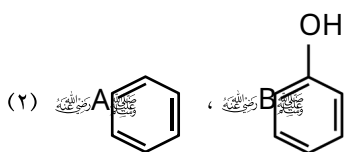
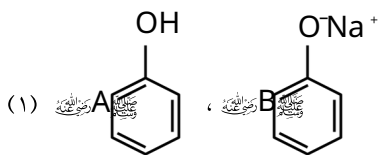
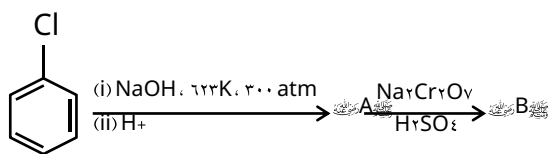
(4) Increase

Ans. (1)

Sol. With increase in temperature, K_w of water increases. So, degree of dissociation of water increase

□ pH as well as pOH of water decrease.

17. Identify the products A and B, respectively in the following reaction :



Ans. (3)

Sol. A is phenol and B is para benzoquinone.

18. Consider a binary solution of two volatile liquid components 1 and 2. x_1 and y_1 are the mole fractions of component 1 in liquid and vapour phase, respectively. The slope and intercept linear plot of $\frac{p_1}{p}$ vs $\frac{y_1}{p}$ are given respectively as :

of the $\frac{p_1}{p}$, $\frac{x_1 p_1}{p}$, $\frac{y_1 p_1}{p}$, $\frac{p_1}{p}$, $\frac{p_1}{p}$, $\frac{p_1}{p}$.



Ans. (1)

Sol. ∴ For liquid solution of two liquids '1' and '2'

$$p_1 = p y_1 = p x_1$$

$$\frac{p_1}{p} = \frac{p_1}{p} \frac{p_1}{p_1} = \frac{p_1}{p} \frac{x_1 p_1}{p_1} = \frac{p_1}{p} \frac{x_1 (p_1 + p_2)}{p_1}$$

$$\frac{p_1}{p} = \frac{p_1}{p} \frac{x_1 (p_1 + p_2)}{p_1} = \frac{p_1}{p} \frac{x_1 p_1 + x_1 p_2}{p_1} = \frac{p_1}{p} \frac{x_1 p_1 + x_1 p_2}{p_1}$$

$$\text{Slope} = \frac{p_1}{p}$$

$$\text{Intercept} = \frac{p_1}{p} \frac{p_2}{p_1}$$

19. Given below are two statements about X-ray spectra of elements :

Statement (I) : A plot of $\sqrt{\nu}$ (ν = frequency of X-rays emitted) vs atomic mass is a straight line.

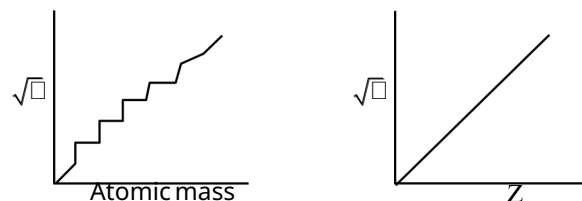
Statement (II) : A plot of ν (ν = frequency of X-rays emitted) vs atomic number is a straight line.

In the light of the above statements choose the correct answer from the options given below :

- (1) Statement I is true but Statement II is false
- (2) Both Statement I and Statement II are true
- (3) Both Statement I and Statement II are false
- (4) Statement I is false but Statement II is true

Ans. (2)

Sol.



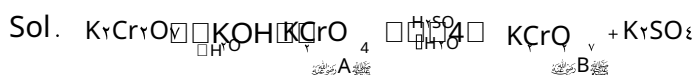
20. Consider the following reactions



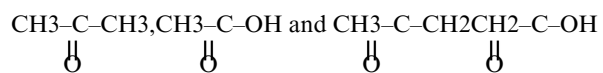
The products A and B, respectively are :

- (1) $\text{K}_2\text{Cr}(\text{OH})_6$ and Cr_2O_3
- (2) K_2CrO_4 and Cr_2O_3
- (3) K_2CrO_4 and $\text{K}_2\text{Cr}_2\text{O}_7$
- (4) K_2CrO_4 and CrO

Ans. (2)



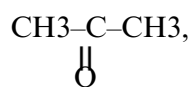
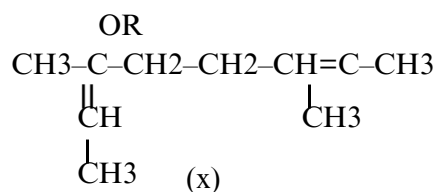
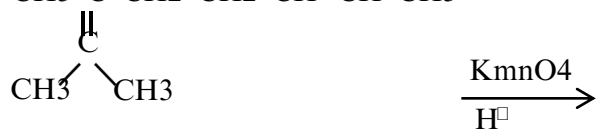
Q. A compound 'X' absorbs 7 moles of hydrogen and 'X' upon oxidation with KMnO_4/H^+ gives



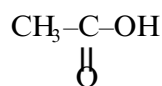
The total number of π bonds present in the compound 'X' is _____.

Ans. (27)

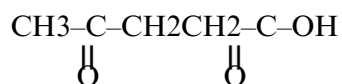
Sol. $\text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_2-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_3$



+



+



SECTION-B

71. 0.1 mole of an organic compound (X) containing 10% hydrogen, on complete combustion produces 0.1 mole of H_2O . The molar mass of (X) is _____ g mol⁻¹.

Ans. (100)

Sol. Organic compound $C_xH_yO_z$
 Molar mass = 100 gm

Mole of $H_2O = \frac{0.1}{18} = 0.0056$ mole

Mole of H in $H_2O = 0.0056 \times 2 = 0.0112$ mole
 = mole of H in 0.1 mole Organic compound

Wt of H atom in 0.1 mole compound = $0.0112 \times 1 = 0.0112$ gm

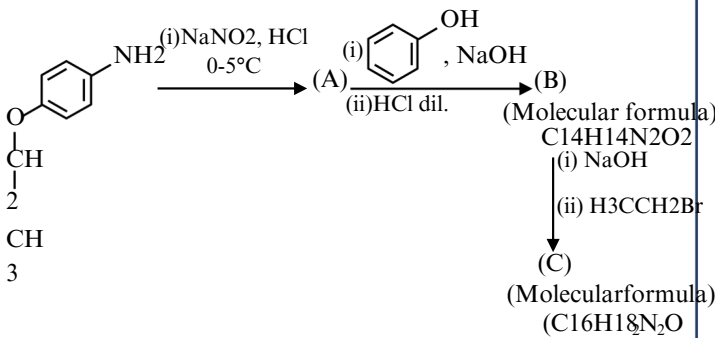
Wt of H atom in one mole compound = $\frac{0.0112}{0.1} = 0.112$ gm

∴ wt. % of H = $\frac{\text{wt. of H in one mole compound}}{\text{Molar mass of compound}} \times 100$

$10 = \frac{0.112}{M} \times 100$

$M = 112$

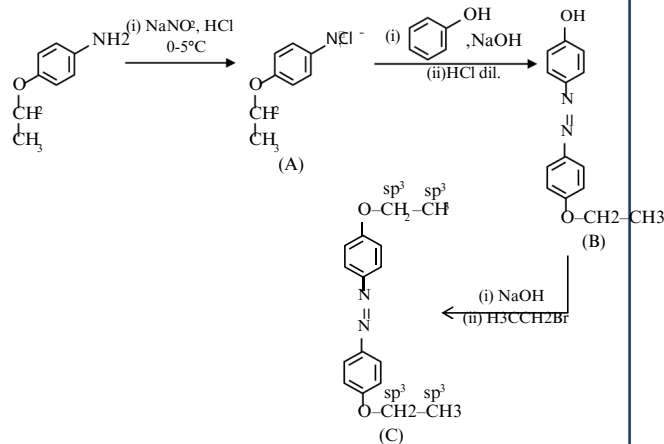
72. Consider the following sequence of reactions.



Total number of sp hybridised carbon atoms in the major product C formed is _____.

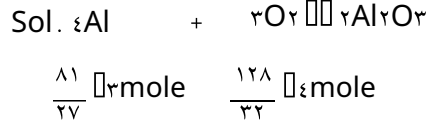
Ans. (2)

Sol.



73. When 18 g of aluminium is allowed to react with 128 g of oxygen gas, the mass of aluminium oxide produced in grams is _____. (Nearest integer) Given : Molar mass of Al is 27 g mol⁻¹

Ans. (102)



Limiting reagent

1 mole of Al_2O_3 formed $\times 3$ mole

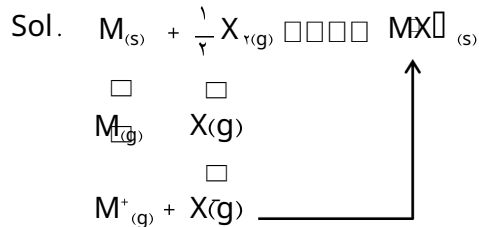
Wt. of Al_2O_3 formed = $\frac{2}{3} \times 102 = 68$ gm

74. The bond dissociation enthalpy of X_2 is H_{bond} calculated from the given data is _____ kJ mol⁻¹ (Nearest integer)

$MX(s) \rightarrow M(g) + X(g) \quad \Delta H^\circ_{lattice} = 100 \text{ kJ mol}^{-1}$
 $M(s) \rightarrow M(g) \quad \Delta H^\circ_{sub} = 100 \text{ kJ mol}^{-1}$
 $M(g) \rightarrow M(g) + e(g) \quad \Delta H^\circ_i = 500 \text{ kJ mol}^{-1}$
 $X(g) + e(g) \rightarrow X(g) \quad \Delta H^\circ_{eg} = -300 \text{ kJ mol}^{-1}$
 $M(s) \rightarrow \frac{1}{2} X_2(g) \rightarrow MX(s) \quad \Delta H^\circ_f = -400 \text{ kJ mol}^{-1}$

Given : MX is a pure ionic compound and X forms a diatomic molecule X_2 is gaseous state.

Ans. (200)



$0 = \Delta H_f(MX) + \Delta H_{sub}(M) + \Delta H_i(M) + \frac{1}{2} B.E.(X-X) + EG(X) + L.E.(MX)$

$-400 = (100) + (500) + \frac{1}{2} (B.E.) + (-300) + (-100)$

$B.E. = 200 \text{ kJ mole}^{-1}$